

Available online at www.sciencedirect.com



C electrochemistry communications

Electrochemistry Communications 9 (2007) 891-894

www.elsevier.com/locate/elecom

Operation of a planar free-breathing PEMFC in a dead-end mode

Olli Himanen^{a,*}, Tero Hottinen^b, Saara Tuurala^a

^a Helsinki University of Technology, Laboratory of Advanced Energy Systems, P.O. Box 2200, 02015 TKK, Finland ^b Wärtsilä Corporation, Fuel Cells, Tekniikantie 14, 02150 Espoo, Finland

Received 9 November 2006; received in revised form 1 December 2006; accepted 1 December 2006 Available online 8 January 2007

Abstract

Operation of a planar free-breathing PEMFC with pressurized anode was studied by constant current and polarization curve measurements. The anode outlet was equipped with a purge valve, whose duty cycle was varied. The results indicate that the cell is able to operate with 0.1 and 0.25 barg hydrogen pressures, with both dry and humid hydrogen. There were no significant differences in cell performance at different purge valve duty cycles. When the purge valve was completely closed signs of flooding were observed. Anode side flooding was reduced at higher hydrogen overpressure due to increased hydraulic permeation of water through the membrane. © 2006 Elsevier B.V. All rights reserved.

Keywords: PEMFC; Free-breathing; Dead-end; Purge valve; Pressurized; Flooding

1. Introduction

Small PEMFCs are potential power sources for portable electronic devices. In such applications, the energy density of the power source should be as high as possible. This requires minimization of the power consumption of auxiliary devices, such as air pumps. One possible way to achieve this is to use the so called free-breathing cells, i.e. cells which take the oxygen passively from the surrounding air. Also fuel utilization should be as high as possible. This can be accomplished by operating the cell in dead-end mode. In dead-end mode, the hydrogen outlet is blocked and thus the anode is pressurized. Complete blocking of the hydrogen outlet may cause serious flooding problems in long term operation due to water accumulation in the anode compartment, see e.g. [1]. It may also cause increase in nitrogen concentration in the anode, see for example [2]. Dead-end operation also affects the water balance of the cathode due to decreased total water removal from the cell. Therefore, the anode outlet should be equipped with a purge valve to remove excess water. In this work, the operation of a planar free-breathing PEMFC in dead-end mode is studied. The cell was tested both with dry and humid hydrogen and with different purge valve duty cycles.

2. Experimental

The cell used in this work was a planar free-breathing PEMFC introduced and used in [3–5] and its active area was 6 cm². Cathode side gas diffusion backing was open to ambient air and the cell took the oxygen needed in the cell reactions directly from the surroundings. The MEA used in this work was GORE^{TM} PRIMEA[®] Series 58, with the membrane thickness of 18 µm. Ambient temperature and relative humidity were approximately 22 °C and 50%, respectively. The cell was separated from the laboratory space by partition walls to minimize the effects of air and people movements in the laboratory. The cell was not heated externally and its temperature was not controlled.

The measurement system is illustrated in Fig. 1. It was fundamentally similar to that described in [3–5] with the exception that the hydrogen outlet was equipped with a Bürkert 6013 solenoid valve. The valve was controlled with an Agilent 34970A Data Acquisition/Switch Unit connected to a computer. Hydrogen pressure was controlled

^{*} Corresponding author. Tel.: +358 94513209; fax: +358 94513195. *E-mail address:* olli.himanen@tkk.fi (O. Himanen).

^{1388-2481/\$ -} see front matter \odot 2006 Elsevier B.V. All rights reserved. doi:10.1016/j.elecom.2006.12.002



Fig. 1. Measurement system.

with Brooks 5866 pressure controller, which was also connected to the Agilent 34970A. A needle valve was installed between the gas handling unit and pressure controller to limit the hydrogen flow and pressure peaks during the purge.

Two different hydrogen pressure levels, 0.1 and 0.25 barg, were used in these measurements. The selection was based on the unpublished test measurements done with the same cell. In those test measurements, it was noticed that 0.25 barg is the maximum pressure for this cell. With larger pressures the performance decreased due to larger mechanical stresses, and the hydrogen also leaked to the surroundings from the anode compartment.

Dry hydrogen and humidified hydrogen with dew-point temperature of 30 °C were used in these measurements. These were selected to simulate different possible hydrogen storage methods. Dry hydrogen corresponds to either metal hydride storages or pressurized gas vessels. Humidified hydrogen corresponds to hydrogen produced from NaBH₄ solution with room temperature micro-hydrogen-generator, as reported e.g. in [6]. Needle valve, pressure controller and gas pipes between gas unit and fuel cell were heated to 35 °C to avoid condensation of water in these components.

Duty cycle of the purge valve was varied from completely closed to 1/60 (seconds open/seconds closed), 1/ 120, 1/300, 1/600, and 1/1200. In dead-end measurements, the hydrogen flow rate was only measured but not controlled. Reference measurements were also conducted in flow-through mode. In these measurements the hydrogen flow rate was controlled, and the stoichiometric factor was kept at 1.5. Dew-point temperature of hydrogen in the reference measurements was 30 °C.

At each humidity and pressure level, and purge valve duty cycle, polarization and constant current measurements were performed. To minimize the effects of load history, the following measurement sequence was applied. First the purge valve and pressure controller were turned on and the current was increased to 0.6 A (100 mA cm⁻²) in 0.1 A/30 s steps. The cell was operated 15 min at 0.6 A current, after which the polarization curve was measured. During polarization measurements, the current was changed in 0.1 A steps with 30 s intervals. The scan was reversed when the cell voltage decreased under 0.4 V or 0.5 V, depending on the conditions (see Section 3). After the scan

the cell was operated in an open-circuit mode for 3 h. This procedure was repeated three times and the third scan was used in the results. Typically the second and third scans were already identical, so it could be concluded that with this procedure the results are not affected by the load history. The constant current measurement was started after the third scan. The cell current was kept at 1.5 A (250 mA cm⁻² for 8 h and the cell voltage was recorded during that period.

3. Results and discussion

Polarization curves measured with purge valve closed and with 1/60 and 1/1200 duty cycles are illustrated in Figs. 2 (0.1 barg) and 3 (0.25 barg). The polarization curves measured with other three duty cycles were between these extreme cases and they are left out for clarity. The performance of the cell was slightly better than in [4,5] where only dry hydrogen was used in a flow-through mode. Hence, the slightly better performance observed here was most probably caused by better conductivity of the membrane due to higher water content.

Polarization curves measured with 0.1 barg hydrogen pressure show that with periodically opened purge valve, the performance of the cell was similar to reference measurements and practically not dependent on the duty cycle. When the purge valve was completely closed, the performance of the cell slightly decreased. With completely closed purge valve, the polarization scan had to be reversed already at 0.5 V, because the voltage typically collapsed suddenly from 0.4 V to under 0.2 V, which was the shutdown limit of the measurement system.

The humidity of the hydrogen had no effect on the polarization curves when the anode side was purged periodically. When the purge valve was closed, the performance of the cell was only slightly decreased with humidified hydrogen, probably due to water accumulation



Fig. 2. Polarization curves measured with 0.1 barg hydrogen pressure.

Download English Version:

https://daneshyari.com/en/article/181860

Download Persian Version:

https://daneshyari.com/article/181860

Daneshyari.com