



Low-temperature photoluminescence in self-assembled diphenylalanine microtubes



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ABSTRACT

Bioinspired self-assembled structures are increasingly important for a variety of applications ranging from drug delivery to electronic and energy harvesting devices. An important class of these structures is diphenylalanine microtubes which are potentially important for optical applications including light emitting diodes and optical biomarkers. In this work we present the data on their photoluminescent properties at low temperatures (down to 12 K) and discuss the origin of the emission in the near ultraviolet (UV) range seen earlier in a number of reports. UV luminescence increases with decreasing temperature and exhibits several equidistant lines that are assigned to zero-phonon exciton emission line and its phonon replicas. We infer that the exciton is localized on the defect sites and significant luminescence decay is due to thermal quenching arising from the carrier excitation from these defects and non-radiative recombination.

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1. Introduction

Self-assembly of bioorganic nanomaterials has attracted significant interest due to various possibilities offered by their unique structure combined with design variability offered by non-covalent interactions in the structure [1–3]. Bioinspired amino acid- and peptide-based nanostructures are of special importance because they can be used in various applications ranging from biological scaffolds [4] and templates for nanowire fabrications [5] to light-emitting diodes (LED) [6] and gates for field-effect transistors [7]. The most studied self-organized peptide is diphenylalanine (FF) that combines fascinating physical properties with the ability to bind to several metals and other functional groups. It is derived from the core recognition motif of Alzheimer's disease β -amyloid polypeptide and thus is interesting from the biomedical point of view. It has been shown that FF-based materials are prone to self-assemble into tubular [8], spherical [9], rod-like [10], and fibrous structures [11] depending on the deposition method and solvent chemistry. Excellent mechanical [12], electrical [13], piezoelectric

[14], optical [15], and electrochemical [16] properties have shown great potential of using FF-structures in nano- and microdevices. Strong blue and near-UV luminescence of FF nano- and microstructures were observed earlier and assigned either to $n-\pi^*$ electronic transitions in the carbonyl group [6] or to the emission of exciton localized within the single nanotube (confined in about 9–10 Å) [17]. Furthermore, the photoluminescence (PL) was found to depend on the morphology of the tubes [10,18] and served as an indicator of the water presence [19]. Strong above-room temperature dependence of near-UV luminescence was observed by Gan et al. [20] rendering FF self-assembled structures as absolute temperature probes for bioapplications. However, the nature of the PL from FF structures is still elusive and current work is focused on studying their low-temperature emission. In this case, both exciton-phonon interactions and non-radiative recombination rates are reduced, which facilitates studying excitonic processes in bioinspired structures such as FF.

2. Materials and methods

The lyophilized form of the diphenylalanine (H-D-Phe-D-Phe-OH, FF) peptide was purchased from Bachem (Bubendorf, Switzerland). The solvent 1,1,1,3,3,3-hexafluoro-2-propanol (HFP) was purchased from Sigma-Aldrich. Fresh peptide stock solution was prepared by dissolving the peptide powder in HFP at a concentration of 100 mg/ml. To avoid any pre-aggregation, fresh stock solutions

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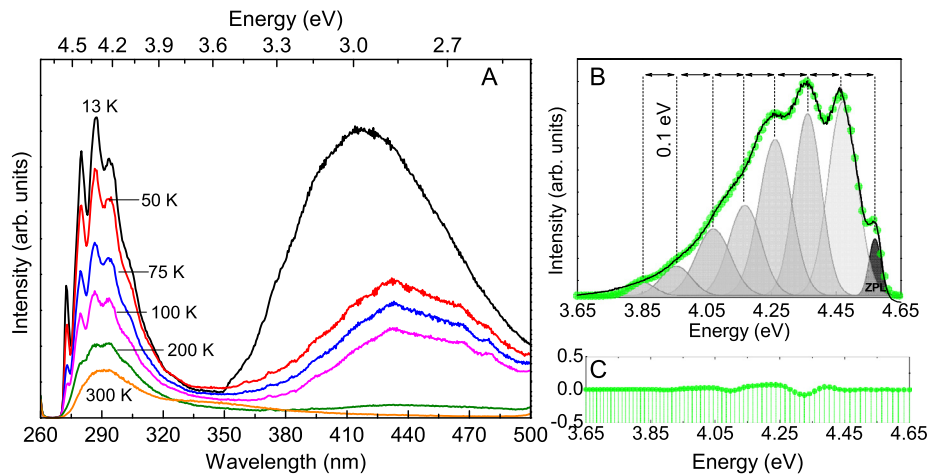


Fig. 1. (A) PL emission spectra of the FF microtubes excited at 255 nm for various temperatures. (B) PL emission spectrum (13 K) and the corresponding Gaussian components (shaded areas) of the cumulative curve (circles). (C) Fit residual plot (coefficient of determination $r > 0.99$).

were prepared before each experiment. FF peptide stock solution was then diluted with distilled water (pH 6.67) to a final concentration of 2 mg/ml. Poly(tetrafluoroethylene) (PTFE) substrates were rinsed with ethanol and then with water in an ultrasonic bath for 10 min. In this work we have used the following method of sample preparation: first, 2 μL of FF stock solution was applied to the substrate and then 98 μL of water was added [1]. The resulting drops were dried at room temperature for one day. FF peptide tubes were then removed from the substrate and obtained powder of FF peptide tubes was stored at low temperature. Typical dimensions of the microtubes ranged from 1 to 5 μm in diameter and from 100 to 1000 μm in length. The microtubes consisted of multiple nanotubes as evidenced from the scanning electron microscopy (SEM) micrographs (see Fig. S1 of the Supplementary Information). SEM images also showed that a small fraction of the studied microtubes were in fact microrods (without a visible hole at the end). The structure of microtubes did not significantly change after repetitive cooling to liquid nitrogen temperature and below (Fig. S2).

The PL spectra were recorded in the temperature range from 320 to 12 K with a Horiba Scientific modular double grating excitation spectrofluorometer and a TRIAX 320 emission monochromator (Fluorolog-3) coupled to a R928 Hamamatsu photomultiplier by using the front-face acquisition mode. The excitation source was a 450 W Xe arc lamp. The temperature was controlled by a He closed-cycle cryostat and a Lakeshore 330 auto-tuning temperature controller with a resistance heater.

3. Results and discussion

Fig. 1 shows the typical emission spectra of peptide microtubes excited at 255 nm in near-UV and visible spectral ranges for various temperatures. At room temperature the emission spectra reveal a near-UV broad band centered at ~ 290 nm and a weaker shoulder at about 340 nm. Below 200 K, a series of narrower peaks in the near-UV band and a strong near-UV/blue/green emission (at 350–500 nm) appear. The near-UV band resembles that already reported by several authors [17,20], and will be the focus of this work as the visible luminescence at longer wavelengths is negligible at room temperature. We measured also the ratio Δ between the integrated intensity of these two emissions (at 290 and 350–500 nm, see Fig. S3A of the Supplementary Information) that can serve as a ratiometric thermometric parameter at low temperatures. We calculated the relative temperature sensitivity [21] defined as $S_r = \frac{\partial \Delta}{\partial T} / \Delta$ in the temperature range from 13 to 200 K (Fig. S3B of the Supplementary Information). Three

distinct temperature regions of the S_r vs. T behavior are clearly identified: i) 13–60 K, where S_r decreases from 3.44 to $0.03\% \cdot \text{K}^{-1}$; ii) 60–100 K, where S_r is maximum around 85 K ($0.31\% \cdot \text{K}^{-1}$), and iii) 100–200 K, where S_r gradually increases up to $1.97\% \cdot \text{K}^{-1}$.

Gan et al. reported a thermometric behavior in FF nanotubes at high temperature (278–338 K) through the temperature dependence of the intensity of the UV broad band and its lifetime [20]. We notice that the ratiometric (or self-referencing) readout based on the intensity ratio is indeed required for this purpose. Using the luminescence intensity ratio for temperature sensing is much more reliable because it is not compromised by the well-known experimental errors arising from the critical dependence of the PL intensity on the sensor concentration, material inhomogeneities, and optoelectronic drifts of the excitation source and detectors [21].

The low-temperature emission spectrum in the near-UV region was fitted by a sum of the Gaussian bands, revealing that they all are energetically equidistant (by ~ 0.1 eV), which could correspond to the energy of the phonons replicated the main narrow peak at 4.54 eV (zero phonon line, ZPL in Fig. 1B).

In order to estimate the fundamental absorption edge and the forbidden energy bandgap (E_g) of the studied FF microtubes we measured the diffuse reflectance (R) as a function of the energy of the incident UV radiation. For monodisperse and thick samples with size comparable to the wavelength of the incident light, E_g can be estimated through the equation $R \propto (h\nu - E_g)^{n/2}$, where n is equal to 1 or 3 for direct and indirect bandgap semiconductors, respectively [22]. E_g was calculated using the linearization of the previous equation, $R^2 \propto (h\nu - E_g)$ for direct band gap semiconductors [23]. The diffuse reflectance data was fitted in the linear region by a straight line. Extrapolation of this line to zero reflectance yielded the value of the forbidden bandgap $E_g = 4.69 \pm 0.03$ eV at room temperature (Fig. 2A). It is thus can be inferred that FF microtubes can be described as a direct bandgap semiconductor that emits in the UV spectral region via annihilation of the excitons after illumination by the near-bandgap radiation.

Fig. 2B represents the temperature dependence of the photoluminescence excitation (PLE) spectra of the peptide microtubes obtained by monitoring the emission maximum at 305 nm. The room-temperature excitation spectrum shows a broad band with an absorption edge around 269 nm (4.6 ± 0.2 eV), resembling that found in the emission spectrum. Similarly to the behavior of the PL emission, the intensity of the excitation spectra increases with decreasing temperature. For temperatures below 200 K, the PLE spectra consist of multiple narrower peaks (Fig. 2C). We notice that all the peaks are again energetically equidistant (~ 0.1 eV) as observed for the emission spectra. Amdursky et al. reported a similar

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