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## Effect of Pr Additions to Li-doped Bi2212 Bulk Superconductors Sintered at Low Temperature

Akihiro Miura, Dai Oikawa, Hiroya Andoh, Toko Sugiura, Takehiko Tsukamoto\*

*National Institute of Technology, Toyota College, 2-1, Eisei-cho, Toyota 471-8525, Japan*

### Abstract

The physical properties study of the  $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_y$  (Bi2212) ceramics samples sintered at low temperature has been carried out by measurement of electrical resistance and powder X-ray diffraction (XRD). The effects of Pr doping on the formation process of the Bi2212 phase, and its structural and superconducting features were studied. The samples were prepared by a solid-state reaction method from oxides and carbonate powders. The samples with nominal composition  $\text{Bi}_{2.12}\text{Sr}_{1.90}\text{Ca}_{1.02}\text{Cu}_{1.96}\text{Li}_{0.15}\text{Cl}_{0.15}\text{O}_y$  added with praseodymium oxide ( $\text{Pr}_2\text{O}_3$ ) were sintered in air at  $710^\circ\text{C}$ . The  $\text{Pr}_2\text{O}_3$  addition ranging between 0.1mass% and 1.2mass% promotes the formation of the Bi2212 phase in low temperature synthesis. The full-width at half maximum (FWHM) value of the (200) and (0010) XRD peaks is slightly decreasing in the samples for low level doping and reaches a minimum around 0.6mass%. The maximum zero resistance temperature ( $T_c$ ) is observed at 82.5K for the sample with 0.6mass%  $\text{Pr}_2\text{O}_3$  addition by sintering even at  $710^\circ\text{C}$ , which is about  $150^\circ\text{C}$  lower than that of the non-added Bi2212 phase.

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**Keywords:** Low temperature synthesis; Bi2212 bulk superconductor;  $\text{LiClO}_4$  and  $\text{Pr}_2\text{O}_3$  addition; Zero resistance

### 1. Introduction

The  $\text{Bi}_2\text{Sr}_2\text{CaCu}_2\text{O}_y$  (Bi2212) superconductor is known to be one of the most stable compounds among the copper oxide-based high- $T_c$  superconductors. In the last 25 years, chemical substitution and/or addition to the Bi2212 superconductors has been widely studied, because it is expected to improve its superconducting properties and to

\* Corresponding author. Tel.: +81-565-36-5853; fax: +81-565-36-5845.

E-mail address: [tsuka@toyota-ct.ac.jp](mailto:tsuka@toyota-ct.ac.jp) (Takehiko Tsukamoto).

influence its formation processes. It has been revealed that the addition of some elements, Li, Cl and Pr to Bi2212 system enhances Bi2212 phase formation and superconducting properties [1-6]. In general, lower temperature synthesis is desirable for commercial production of this important material. Noro et al. [2] reported that the X-ray diffraction (XRD) pattern of  $\text{Bi}_{2.12}\text{Sr}_{1.90}\text{Ca}_{1.02}\text{Cu}_{1.96}\text{Li}_{0.15}\text{Cl}_{0.15}\text{O}_y$  compounds sintered at  $720^\circ\text{C}$  shows the predominance of the Bi2212 phase. Then, the zero resistance of the sample is observed at 78K. Aloysius et al. [6] reported that the critical current density ( $J_c$ ) and the zero resistance temperature ( $T_c$ ) of the (Bi, Pb)-2212 system is highly enhanced upon the addition of Pr. However, no further studies have yet been made to clarify the effect of Pr addition on the physical properties of Bi2212 bulk samples sintered at low temperature. In this work, we study the effect of Pr additions on the  $T_c$  and the full-width at half maximum (FWHM) values of the (200) and (0010) XRD peaks for Bi2212 bulk samples sintered at low temperature, in comparison with that of non-added sample.

## 2. Experimental

The Bi2212 bulk samples were prepared by a conventional solid-state reaction with a nominal composition of  $\text{Bi}_{2.12}\text{Sr}_{1.90}\text{Ca}_{1.02}\text{Cu}_{1.96}\text{Li}_{0.15}\text{Cl}_{0.15}\text{O}_y$  [7]. Raw materials of  $\text{SrCO}_3$  (99.9%),  $\text{CaCO}_3$  (99.9%) and  $\text{CuO}$  (99.9%) were mixed and calcined twice at  $950^\circ\text{C}$  for 20h in air. After grinding, the calcined powder was mixed with  $\text{Bi}_2\text{O}_3$  (99.9%),  $\text{LiClO}_4$  (98.0%) and various amounts of  $\text{Pr}_2\text{O}_3$  (99.9%). The mixtures were pelletized using a cylindrical die of 13mm diameter under a force of 30 kN. The Pr added samples were sintered at  $T_s=710^\circ\text{C}$  for 100h in air. The sintering temperatures of other samples were described later. Finally, the samples were annealed at  $700^\circ\text{C}$  for 20h and rapidly quenched in liquid nitrogen, to ensure optimum value of the oxygen concentration. All the products were examined at room temperature by powder XRD using  $\text{Cu-K}\alpha$  radiation (Rigaku Ultima IV). The measurements were carried out with the  $\theta$ - $2\theta$  scanning method in the range of  $2\theta = 20$ - $65^\circ$ . The chemical compositions of the sample were measured by wavelength-dispersive X-ray fluorescence spectrometry (WDXRF; Rigaku ZSX-Primus III). The electrical resistance  $R$  was measured by a standard DC four-probe method at temperatures between 50 and 290K.

## 3. Results and discussion

### 3.1. Effect of $\text{LiClO}_4$ and $\text{Pr}_2\text{O}_3$ addition

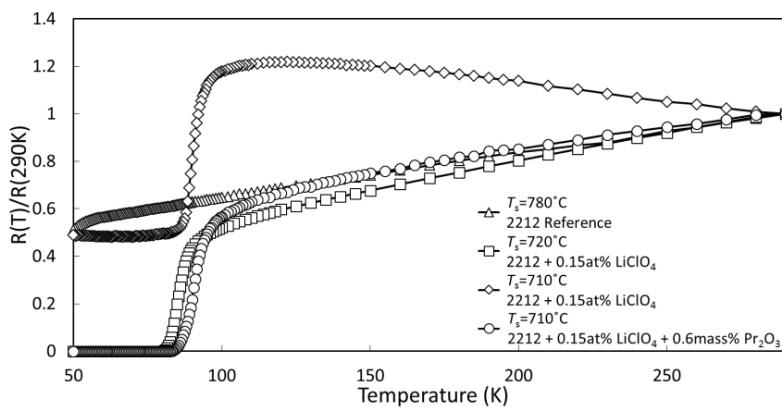


Fig. 1. Temperature dependence of the normalized resistance  $R(T)/R(290\text{K})$  for 0.15at%  $\text{LiClO}_4$  and 0.6mass%  $\text{Pr}_2\text{O}_3$  added sample sintered at  $710^\circ\text{C}$ , 0.15at%  $\text{LiClO}_4$  added samples sintered at 710 and  $720^\circ\text{C}$ , and reference (non-added) sample sintered at  $780^\circ\text{C}$ .

A comparison of the  $R$ - $T$  behavior of  $\text{LiClO}_4$  and  $\text{Pr}_2\text{O}_3$  added sample,  $\text{LiClO}_4$  added sample and non-added sample is shown in Fig. 1. The  $T_c$  ( $R=0$ ) value is 82.5K for  $\text{LiClO}_4$  and  $\text{Pr}_2\text{O}_3$  added sample sintered at  $T_s=710^\circ\text{C}$ . In addition, the  $T_c$  value is 78K for  $\text{LiClO}_4$  added sample sintered at  $T_s=720^\circ\text{C}$ . On the other hand, zero resistance is not observed above 50K and the resistance variation is found to be non-metallic from room temperature down to

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