

Available online at www.sciencedirect.com



Applied Radiation and Isotopes

Applied Radiation and Isotopes 66 (2008) 50-59

www.elsevier.com/locate/apradiso

DMSO increases radioiodination yield of radiopharmaceuticals

Ketai Wang, S. James Adelstein, Amin I. Kassis*

Department of Radiology, Armenise Building Room D2-137, 200 Longwood Avenue, Harvard Medical School, Boston, MA 02115, USA

Received 30 January 2007; received in revised form 28 June 2007; accepted 31 July 2007

Abstract

A high-yield radioiodination method for various types of molecules is described. The approach employs DMSO as precursor solvent, a reaction ratio of 2–5 precursor molecules per iodine atom, 5–10 µg oxidant, and a 10–25 µl reaction volume. The solution is vortexed at room temperature for 1–5 min and progress of the reaction is assessed by HPLC. Radioiodinated products are obtained in \geq 95% yield and meet the requirements for radiotracer imaging, biodistribution studies, and molecular and cellular biology research. © 2007 Elsevier Ltd. All rights reserved.

Keywords: DMSO; Radioiodination; No carrier added; High yield

1. Introduction

Radioactive isotopes of iodine possess excellent characteristics for non-invasive imaging (¹²³I, ¹²⁴I), radiotracer studies (¹²⁵I), tumor therapy (¹²⁵I, ¹³¹I), and molecular biology research. For many types of molecules, radioiodination presents problems of yield, purification, and stability of the radiolabeled product (Prusoff, 1959; Mannan et al., 1991; Tjuvajev et al., 1993; Van den Abbeele et al., 1996; Kumar et al., 2005). In a standard procedure, radioiodide is oxidized and added to the molecule to be labeled. The iodine cation then reacts with the active position (e.g., phenol, hydroxyl, tributyltin, benzamino or vinyl moiety) on the molecule.

During our studies in which radioiodination of waterinsoluble precursors in aqueous medium often produces low yields, we observed that dissolving these compounds in DMSO improves the efficiency of radioiodination. While the original purpose of the addition of DMSO was the solubilization of water-insoluble compounds, we have since realized that the presence of DMSO dramatically increases the radioiodination yields of various water-soluble as well as water-insoluble organic and non-organic molecules.

0969-8043/\$ - see front matter © 2007 Elsevier Ltd. All rights reserved. doi:10.1016/j.apradiso.2007.07.026

2. Materials and methods

2.1. General methods

Reagents were obtained from Sigma-Aldrich Chemical Company. HPLC separations were performed on a reversed-phase Zorbax SB C₁₈ column, 9.4 mm × 250 mm (Agilent Technology), at a flow rate of 3 mL/min, with UV absorption (Waters 486 detector) and γ -ray detection (gamma-ram, IN/US Systems) used to analyze the eluates. Na¹²⁵I was purchased from GE Healthcare Life Sciences (17.4 Ci/mg [643.8 GBq/mg], 453 mCi/mL [16.76 GBq/mL]) in 0.1 N sodium hydroxide, Na¹²³I from MDS Nordion as a no-carrier-added powder, and Na¹³¹I from PerkinElmer Life and Analytical Sciences (16.5 Ci/mg [610.5 GBq/mg], 10 mCi/mL [370 MBq/mL]) in 0.1 N sodium hydroxide.

2.2. Synthesis of radiolabeled $(^{123}I/^{125}I)$ 5-iodo-2'deoxyuridine $(^{123}IUdR/^{125}IUdR)$

A stock solution of tributylstannyl-2'-deoxyuridine (SnUdR, $20 \,\mu g/\mu L$ DMSO) was prepared. Phosphatebuffered saline (PBS, $10 \,\mu L$, 0.01 M, pH 7.4), SnUdR ($1 \,\mu L$, $1 \,\mu g/\mu L$ DMSO, methanol, or water), and Na¹²⁵I ($0.5-3 \,\mu L$) were placed in a reaction vial coated with 1,3,4,6-tetrachloro- 3α , 6α -diphenylglycouril (Iodogen, $10 \,\mu g$). When dichloromethane was used as precursor solvent, SnUdR ($1 \,\mu g/20 \,\mu L$ solvent) also adhered to the

^{*}Corresponding author. Tel.: +16174327777; fax: +16174322419. *E-mail address:* amin_kassis@hms.harvard.edu (A.I. Kassis).

vial, and the rest of the conditions were the same. The mixture was vortex mixed at ambient temperature for 1 min. Progress of the reaction was checked by C_{18} HPLC with isocratic elution (3 mL/min) using 75% A (phosphate buffer, 0.05 M, pH 2.5) and 25% B (methanol) for 16 min.

For ¹²³IUdR, dry Na¹²³I (10 mCi) was dissolved in HCl (10 μ L, 0.1 M) and sodium thiosulfate (1 μ L, 1 μ g/ μ L water) was added. The mixture was shaken for 1 min and NaOH (9 μ L, 0.1 M) was then introduced, and the solution was vortex mixed for 10 s. Radioiodination then proceeded as above.

2.3. Synthesis of radiolabeled $({}^{123}I/{}^{125}I/{}^{131}I)$ ammonium 2-(2'-phosphoryloxyphenyl)-6-iodo-4-(3H)-quinazolinone $({}^{123}IQ_{2-P}/{}^{125}IQ_{2-P}/{}^{131}IQ_{2-P})$

A mixture of $\text{SnQ}_{2-P(I)}$ and SnQ_{2-P} (0.5 mg) was synthesized (Ho et al., 2002). ESI-HRMS $[M+H]^+$ calcd for $\text{SnQ}_{2-P(I)}$, 589.1296, found, 589.1296; calcd for SnQ_{2-P} , 609.1539, found, 609.1539; ³¹P NMR for $\text{SnQ}_{2-P(I)}$, -16.922 ppm, and for SnQ_{2-P} , -2.958 ppm. The mixture was dissolved in DMSO (100 µL), kept overnight until $\text{SnQ}_{2-P(I)}$ was converted to SnQ_{2-P} (at which point ³¹P NMR for solution -2.958 ppm), and then diluted to a stock solution of SnQ_{2-P} (20 µg/µL DMSO.

Phosphate-buffered saline $(10 \,\mu\text{L}, 0.01 \,\text{M}, \text{pH } 7.4)$ and SnQ_{2-P} (1 $\mu\text{L}, 1 \,\mu\text{g}/\mu\text{L}$ DMSO, methanol, or water) were placed in a reaction vial coated with Iodogen (10 μg), and $\text{Na}^{123}\text{I}/\text{Na}^{125}\text{I}/\text{Na}^{131}\text{I}$ (2.0 μL) was added. The mixture was vortex mixed at ambient temperature for 3 min and analyzed by C₁₈ HPLC using phosphate buffer (0.05 M, pH 2.5) as A and methanol as B, going from 10% A to 100% B in 7 min and remaining at 100% B for an additional 9 min (3 mL/min).

2.4. Synthesis of iodinated $({}^{125}I/{}^{127}I)$ rhodamine 123 $({}^{125}I-Rhod/{}^{127}I-Rhod)$

2.4.1. Synthesis of mono- and di-iodorhodamine 123

To a reaction vial containing rhodamine 123 (4.5 mg, 11.85 µmol) in ammonium acetate (0.5 mL, 0.2 M), peracetic acid (0.2 mL of 2%, 52.6 µmol) and NaI (2 mg/0.1 mL water, 13.3 µmol) were added. After vortex mixing for 2 h at room temperature, the reaction mixture was evaluated by HPLC, and iodorhodamine 123 (I-Rhod) and diiodorhodamine 123 (I_2-Rhod) were identified by mass spectroscopy. ESI-HRMS calcd for $C_{21}H_{15}N_2O_3I$ [M+H]⁺, 471.0212, found, 471.0206; calcd for $C_{21}H_{15}N_2O_3I_2$ [M+H]⁺, 596.9176, found, 596.9172.

2.4.2. Synthesis of radiolabeled (^{125}I) rhodamine 123 $(^{125}I-Rhod)$

A stock solution of Rhod $(3 \mu g/\mu L DMSO$, methanol, or water) was prepared. Copper chloride solution $(2 \mu L, 50 \mu g/\mu L$ water, pH 4.0), Rhod $(0.5 \mu L DMSO$, methanol, or water, $3 \mu g/\mu L$), and water $(20 \mu L)$ were placed in a

reaction vial. *N*-chloro-*p*-toluenesulfonamide sodium salt (chloramine-T [ChT], 1μ L, 10μ g/ μ L) and Na¹²⁵I (0.5–2 mCi) were added. The mixture was vortex mixed for 5 min and then analyzed by C₁₈ HPLC using phosphate buffer (0.05 M, pH 2.5) as A and methanol as B, going from 10% A to 100% B in 7 min, and remaining at 100% B for an additional 9 min (3 mL/min).

2.5. Synthesis of radiolabeled $(^{123}I/^{125}I/^{131}I)$ 2-iodo-8methyl-8H-quino[4,3,2-kl]acridine $(^{123}I-Acr/^{125}I-Acr/^{131}I-Acr)$

8,13-Dimethyl-2-(tributylstannyl)-8*H*-quinolino[2,3,4*nn*]acridin-13-ium iodide (SnAcr) was kindly provided by C.A. Laughton, University of Nottingham, UK. A stock solution of SnAcr ($20 \mu g/\mu L$ DMSO) was prepared. Phosphate-buffered saline (8 μL , 0.1 M, pH 5.0) and SnAcr (1 μL , 2 $\mu g/\mu L$ DMSO, methanol, or water) were placed in a reaction vial and Na¹²³I/Na¹²⁵I/Na¹³¹I (2.0 μL) and ChT (2 μL , 5 $\mu g/\mu L$ water) were added. After vortex mixing at ambient temperature for 15 min, the reaction mixture was analyzed by C₁₈ HPLC using phosphate buffer (0.05 M, pH 2.5) as A and methanol as B, going from 10% A to 100% B in 7 min, and remaining at 100% B for an additional 9 min (3 mL/min).

2.6. Synthesis of radiolabeled (^{125}I) Bolton–Hunter reagent $(^{125}I-BH)$

Crystalline *N*-succinimidyl 3-(4-hydroxyphenyl)propionate (Bolton–Hunter [BH] reagent) was dissolved in DMSO or methanol (1 μ g/ μ L). To a reaction vial coated with Iodogen (5 μ g), BH reagent in DMSO or methanol (1 μ L) and PBS (10 μ L, 0.01 M, pH 7.4) were added, followed by Na¹²⁵I (1 μ L, 1 mCi/2.5 μ L). When dichloromethane was used as precursor solvent, BH reagent (10 μ L, 0.1 μ g/ μ L solvent) also adhered to the vial, and the rest of the conditions were the same. The mixture was vortex mixed at ambient temperature for 1 min and was analyzed by C₁₈ HPLC using phosphate buffer (0.05 M, pH 2.5) as A and methanol as B, going from 10% A to 100% B in 7 min, and remaining at 100% B for an additional 9 min (3 mL/min).

2.7. Radioiodination (^{125}I) of immunoglobulin $G(^{125}I-IgG)$

Immunoglobulin G ($20 \mu g/2 \mu L$ DMSO or $20 \mu g/4 \mu L$ water) was placed in a vial coated with Iodogen ($20 \mu g$), and sufficient PBS (0.01 M, pH 7.4) to make the reaction volume $24 \mu L$ and Na¹²⁵I ($1 \mu L$, $1 \text{ mCi}/2.5 \mu L$) were added. The mixture was vortex mixed at ambient temperature for 1 min, then transferred to another vial lacking Iodogen. The product was identified by HPLC (Bio-Sil column, $300 \text{ mm} \times 7.8 \text{ mm}$) with isocratic elution using phosphate buffer (0.05 M, pH 6.8) at 1 mL/min.

Download English Version:

https://daneshyari.com/en/article/1877147

Download Persian Version:

https://daneshyari.com/article/1877147

Daneshyari.com