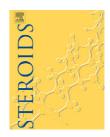


available at www.sciencedirect.com







Synthesis of exemestane labelled with ¹³C

Erminia Fontana^{a,*}, Alberto Pignatti^a, Danilo Giribone^a, Enrico Di Salle^b

- ^a Isotope Chemistry, Accelera Unit, Nerviano Medical Sciences, Viale Pasteur 10, 20014 Nerviano (MI), Italy
- ^b Global Medical Oncology, Pfizer Italia S.r.l, Via Lorenteggio 257, 20152 Milano, Italy

ARTICLE INFO

Article history:
Received 19 December 2007
Received in revised form
15 February 2008
Accepted 24 February 2008
Published on line 4 March 2008

Keywords: Exemestane Aromatase inhibitor Labelling ¹³C Internal standard LC-MS

ABSTRACT

The synthesis of exemestane (Aromasin®), an irreversible steroidal aromatase inhibitor, specifically labelled with 13 C is reported. The preparation of [13 C₃]exemestane was achieved according to an eight-step procedure starting from the commercially available testosterone. © 2008 Elsevier Inc. All rights reserved.

1. Introduction

Exemestane (Aromasin®, 6-methylenandrosta-1,4-diene-3,17-dione; molecular weight 296.41) (Fig. 1) is an orally active irreversible aromatase inhibitor which is clinically in use for the treatment of postmenopausal women with early or advanced breast cancer [1-3]. The initial analytical methods developed for the determination of exemestane in biological fluids involved the use of high-performance liquid chromatography (HPLC) combined with ultraviolet detection or with thermospray mass spectrometry or followed by specific radioimmunoassay [4–6]. However, a higher samples throughput as well as lower limits of quantitation were needed to assay exemestane in human plasma. Therefore, the preparation of a stable labelled version of the compound was required to be used as analytical internal standard (IS) to develop and validate a sensitive, specific and reliable method for the quantitation of exemestane in biological fluids using

liquid chromatography–mass spectrometry (LC–MS). In this paper, the approach followed to prepare a suitable stable labelled IS of exemestane is described.

2. Experimental

Chemicals and materials: All solvents and reagents were of analytical grade and were used without purification unless otherwise indicated. Testosterone (1) was purchased from Fluka, with a chemical purity >99%. [$^{13}C_2$]Acetyl chloride (99 atom % ^{13}C) and [^{13}C]formaldehyde (20% aqueous solution; 99 atom % ^{13}C) were obtained from Aldrich Chemical Co.

Instrumentation and equipment: Chemical purities were determined by HPLC using a series-200 pump (PerkinElmer) equipped with series-200 solvent degasser (PerkinElmer), series-200 autosampler (PerkinElmer) and a LC-235 UV diode array detector (PerkinElmer) connected with Totalchrom

^{*} Corresponding author. Tel.: +39 0331 581148; fax: +39 0331 581682. E-mail address: erminia.fontana@nervianoms.com (E. Fontana). 0039-128X/\$ – see front matter © 2008 Elsevier Inc. All rights reserved. doi:10.1016/j.steroids.2008.02.010

Fig. 1 - Structural formula of exemestane.

Client/Server (PENelson) as integrator *via* link 600 interface (PerkinElmer). Preparative-HPLC was performed at 25 °C using a PrepStar HPLC system (Varian). ¹H NMR data were recorded on Varian INOVA 400 spectrometer, equipped with a 5 mm 1H{15N-31P} z-axis pulse field gradient (PFG) indirect detection probe. ¹³C NMR data were obtained by direct detection on a varian INOVA 500 spectrometer, equipped with a 3 mm broadband 15N-31P{1H-19F} probe optimized for 13C sensitivity and by heteronuclear correlation spectra (H-C) recorded on Varian Unity INOVA 500 spectrometer, equipped with triple resonance 1H{13C, 15N} z-axis pulse field gradient (PFG) indirect detection Cold-Probe. Chemical shifts were referenced with respect to the residual solvent signals (DMSO-d6: 2.50 ppm for ¹H and 39.5 ppm for ¹³C).

Analytical method: HPLC: X-Terra Waters RP18 column (mm $4.6\times100,\,5~\mu m)$ eluting with $H_2\text{O:CH}_3\text{CN:HCOOH}$ 90:10:0.1 by volume (A) and $H_2\text{O:CH}_3\text{CN:HCOOH}$ 10:90:0.1 by volume (B): from 100% A to 0% A in 10 min; 5 min at 0% A; from 100% B to 100% A in 1 min; 4 min at 100% A. Flow rate: 1 ml/min. Column temperature: 25 °C. Analytical wavelength: 215 nm.

2.1. 17β -Benzoyloxy-testosterone (2)

A solution of benzoyl chloride (BzCl; 49 ml, 42 mmol) in dry toluene (60 ml) was slowly dripped into a stirred and cooled (0 °C) solution of testosterone (1) (10.08 g, 34.9 mmol) in dry toluene (100 ml). After addition of dry pyridine (14 ml, 173 mmol), the reaction mixture was stirred at reflux for 20 h then cooled to room temperature. An aqueous solution of 0.5 N HCl (60 ml) was added and the mixture was stirred for 1 h. After separation, the organic layer was treated with 0.5 N HCl (4 ml \times 100 ml), 8% NaHCO₃ (3 ml \times 100 ml), brine (2 ml \times 100 ml) and dried over Na₂SO₄. The solvent was evaporated to dryness giving the intermediate (2) (14.03 g, 35.7 mmol), with a chemical purity of 91% (determined by HPLC; see analytical method; $R_t = 14.0$ min).

2.2. 17β -Benzoyloxy-5-oxo-3,5-seco-4-norandrostan-3-oic acid (3)

An aqueous solution of 2M K_2CO_3 (32.6 ml) was added to a suspension of (2) (13.00 g, 33.2 mmol) in tert-butanol (325 ml). The reaction mixture was heated at 40 °C under stirring and two separate solutions of 0.13 M KMnO₄ (42.25 ml) and 0.75 M KIO₄ (419 ml) were contemporaneously introduced into the reaction

flask. At the end of the addition (about 2h) the mixture was stirred for 2h at 40 °C then cooled to room temperature and evaporated to dryness. The obtained residue was dissolved with water (500 ml), transferred into a separating funnel and extracted with ethyl acetate (EtOAc 2 ml \times 200 ml). After separation, the combined organic layers were back extracted with 8% NaHCO3 (3 ml \times 100 ml). The basic aqueous phases were combined with the aqueous phase previously obtained, then 6 N HCl was added up to pH 2. After extraction with EtOAc (4 ml \times 250 ml) the obtained organic phases were combined, extracted with brine (2 ml \times 300 ml) and dried over Na₂SO₄. The solvent was removed by evaporation, and the obtained intermediate (3) (11.09 g, 28.9 mmol) was used without further purification in the next step.

2.3. $3-0xo-4-oxa-5-androsten-17\beta-yl$ benzoate (4)

The intermediate (3) (11.09 g, 28.9 mmol) was dissolved in dry EtOAc (480 ml) under nitrogen then a freshly prepared solution of acetic anhydride (94 ml) and 65% HClO₄ (0.24 ml) in dry EtOAc (480 ml) was introduced into the reaction flask. After stirring at room temperature for exactly 10 min, 8% NaHCO₃ (740 ml) was added, the mixture was stirred for 30 min then transferred into a separating funnel. After separation of the aqueous layer, the organic phase was extracted with 8% NaHCO₃ (3 ml × 300 ml), water $(2 \text{ ml} \times 300 \text{ ml})$, brine $(2 \text{ ml} \times 300 \text{ ml})$ and dried over Na₂SO₄. After solvent evaporation, the oily yellow residue was submitted to 3 crystallizations from mixtures of CH₃OH:H₂O 3:1 by volume. After drying under high vacuum at 40°C, the intermediate (4) was recovered as a white powder (5.79 g, 14.7 mmol) with a chemical purity of 90% (determined by HPLC; see analytical method; $R_t = 11.7 \text{ min}$). MS (ESI-MS): m/z395 ([MH]+).

2.4. $2-[1',2'-{}^{13}C_2]$ Acetyl-3-oxo-4-oxa-5-androsten-17 β -yl benzoate (5)

A solution of the intermediate (4) (1.51g, 3.83 mmol) in dry tetrahydrofuran (THF; 34.4 ml) was slowly added under nitrogen to a cooled (-78°C) and stirred 1M solution of lithium bis(trimethylsilyl)amide (7.66 ml; 7.66 mmol) in dry THF. After stirring under nitrogen at -78°C for 1h, a solution of [13C2]acetyl chloride (308 mg, 3.83 mmol) in dry THF (2.7 ml) was added. The mixture of reaction was stirred at -78 °C for 90 min, then 1 N HCl (4.8 ml) and water (4.8 ml) were added and the mixture was allowed to reach room temperature then diluted with EtOAc (50 ml). After transferring into a separating funnel and separation, the organic layer was washed with brine (4 ml × 200 ml), dried over Na₂SO₄ and evaporated to dryness. The obtained crude intermediate (5) was purified by flash chromatography on silica gel using a mixture n-hexane:EtOAc 2:1 by volume as the eluting solvent system. The collected fractions were combined as appropriate and after solvent evaporation to dryness, the intermediate (5) (446 mg; 1.02 mmol) was recovered as a white solid with a chemical purity >95% (determined by HPLC; see analytical method; $R_t = 13.8 \, \text{min}$) and an isotopic purity of 99 atom % 13C. MS (ESI-MS): m/z 439 $([MH]^+).$

Download English Version:

https://daneshyari.com/en/article/2028615

Download Persian Version:

https://daneshyari.com/article/2028615

<u>Daneshyari.com</u>