

# Measurements of binary diffusion coefficients and retention factors for dibenzo-24-crown-8 and 15-crown-5 in supercritical carbon dioxide by chromatographic impulse response technique

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## Abstract

Binary diffusion coefficients  $D_{12}$  and retention factors  $k$  for dibenzo-24-crown-8 and 15-crown-5 ethers at 308.18 and 313.20 K, and vitamin K<sub>1</sub> at 313.20 K were measured in supercritical carbon dioxide by the chromatographic impulse response technique, and the effects of molecular shapes on the  $D_{12}$  values were studied. At 313.2 K and 11.0 MPa the  $D_{12}$  value of 15-crown-5 was higher than that predicted from the  $D_{12}$  correlation with molecular weight MW over the range from 32 of methanol to 1138 of trinevonin reported in our previous studies, while the  $D_{12}$  value of dibenzo-24-crown-8, disk shape molecule with MW = 448.5, was almost in agreement with that of vitamin K<sub>1</sub>, long chain molecule with MW = 450.7, and with those predicted from the correlation.

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## 1. Introduction

Supercritical fluids have been employed as not only solvents in extraction and fractionation but also reaction media in synthetic, polymerizing and decomposition reactions. Binary diffusion coefficient  $D_{12}$  is one of the important physical properties for designing reactors, and a large number of  $D_{12}$  measurements for various compounds in supercritical fluids have been made [1,2]. Recently, we found that  $D_{12}$  values were primarily affected by solute molecular weight, and secondarily by the solute polarity [3]. Moreover, the  $D_{12}$  values at 313.2 K and 11.0 MPa were expressed with a straight line in logarithmic plots of  $D_{12}$  value versus molecular weight MW with a slope of  $-0.5$  [2]. However, most measured compounds having large molecular weights are long chain molecules, and the effect of the molecular shape on  $D_{12}$  value is not well understood. Although binary diffusion coefficients of disk shape molecules such as various crown ethers were measured in organic solvents

at an ambient pressure [4–8], those in supercritical fluids are not available in the literature. Thus, the objectives in this study are to measure  $D_{12}$  values for disk shape molecules, i.e. dibenzo-24-crown-8 and 15-crown-5, and to investigate the effect of molecular shapes on  $D_{12}$  values in supercritical carbon dioxide, as compared with vitamin K<sub>1</sub> having a long chain shape and the nearly same molecular weight as dibenzo-24-crown-8.

## 2. Theory

The theory for the chromatographic impulse response technique was described in detail in elsewhere [9,10]. When a tracer is pulse-injected into a fully developed laminar flow in a coated cylindrical column, the cross-sectional average concentration  $C_a(t)$  can be described as follows:

$$C_a(t) = \left( \frac{m}{\pi R^2} \right) \frac{1}{(1+k)\sqrt{4\pi at}} \exp \left\{ -\frac{(L - u_a t/(1+k))^2}{4at} \right\} \quad (1)$$

$$a = \frac{D_{12}}{1+k} + \frac{1+6k+11k^2}{(1+k)^3} \frac{R^2 u_a^2}{48 D_{12}} \quad (2)$$

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where  $m$  is the amount of the tracer species,  $R$  the radius,  $t$  the time,  $L$  the distance,  $u_a$  the average fluid velocity,  $k$  the retention factor, and  $D_{12}$  the binary diffusion coefficient of the tracer species. When  $u_a$  is measured,  $D_{12}$  and  $k$  can be simultaneously determined by minimizing the root-mean-square (rms) error  $\varepsilon$ , defined by Eq. (3), for the measured  $C_{\text{exp}}(t)$  and calculated  $C_a(t)$  curves.

$$\varepsilon = \left\{ \frac{\int_{t_1}^{t_2} (C_a(t) - C_{\text{exp}}(t))^2 dt}{\int_{t_1}^{t_2} (C_{\text{exp}}(t))^2 dt} \right\}^{1/2} \quad (3)$$

where  $t_1$  and  $t_2$  are chosen as the times at the front and latter 10% peak heights, respectively, of the measured response curve to exclude possible tailing portions of the response curves.

### 3. Experimental

#### 3.1. Chemicals

Dibenzo-24-crown-8 (2,3,14,15-dibenzo-1,4,7,10,13,16,19,22-octaoxacyclotetracos-2,14-diene, purity of 98%), 15-crown-5 (1,4,7,10,13-pentaoxacyclopentadecane, 98%), and acetone (99.5%) were purchased from Aldrich (USA), and vitamin K<sub>1</sub> (2-methyl-3-phytyl-1,4-naphthoquinone, no information on purity) and hexane (99%) from Sigma (USA). These reagents were used without further purification. Note that no other peaks appeared in the two dimensional chromatogram of wavelength against time when vitamin K<sub>1</sub> was loaded to the diffusion column. CO<sub>2</sub> (99.995%, contains water <40 ppm) was obtained from Showa Tansan Co., Japan.

#### 3.2. Equipment and procedure

The experimental apparatus and procedures are substantially the same as those in the previous studies [9,11], and briefly described below. A diffusion column used was a poly(ethylene glycol) coated capillary column (UACW-15W-1.0F, the Frontier Laboratories Ltd., Japan) with polymer thickness of 1  $\mu\text{m}$ , 0.515 mm I.D., 15.10 m long, and coil diameter of 0.27 m. CO<sub>2</sub> was fed by a syringe pump (100DX, ISCO, USA), and measurement was started after further passing 2–3 h from the moment that the flow system had become stabilized. A solute dissolved in an organic solvent (a 0.5  $\mu\text{L}$  acetone solution of dibenzo-24-crown-8 at a concentration of 3–8  $\text{kg m}^{-3}$ , a 0.2  $\mu\text{L}$  hexane solution of 15-crown-5 at 5  $\text{kg m}^{-3}$ , and a 0.5  $\mu\text{L}$  supercritical CO<sub>2</sub> solution of vitamin K<sub>1</sub>) through an HPLC injector (Model 7520, Rheodyne, USA) into the diffusion column. In the course of each measurement CO<sub>2</sub> flow rates were measured with a soap-bubble flow meter several times. Response curves, as time changes of the tracer concentration, in terms of absorbance spectra were monitored with a high-pressure multi-UV detector (MD-1510, JASCO, Japan) by scanning at wavelengths from 195 to 400 nm with increments of 1 nm and time interval of 1.6 s. By examining the wavelength dependences on the response curve intensity, the values of  $D_{12}$ ,  $k$  and  $\varepsilon$ , as described for various solutes in our previous studies [1,2,9–11], the response

curves monitored at 275, 200 and 325 nm were used to determine the  $D_{12}$  and  $k$  values for dibenzo-24-crown-8, 15-crown-5, and vitamin K<sub>1</sub>, respectively. Note that the solubility of dibenzo-24-crown-8 in supercritical carbon dioxide [12] is much lower than that of 15-crown-5.

All measurements were made at  $DeSc^{1/2} < 8$ , which is a criterion for the secondary flow effect due to a diffusion column coiling, and corresponds to less than 1% of the effect on  $D_{12}$  value in terms of the second moment [13], where  $De$  and  $Sc$  are the Dean number and Schmidt number, respectively. The CO<sub>2</sub> densities were obtained with the equations of Pitzer and Schreiber [14] and Span and Wagner [15], and the viscosities were estimated by the method of Fenghour et al. [16].

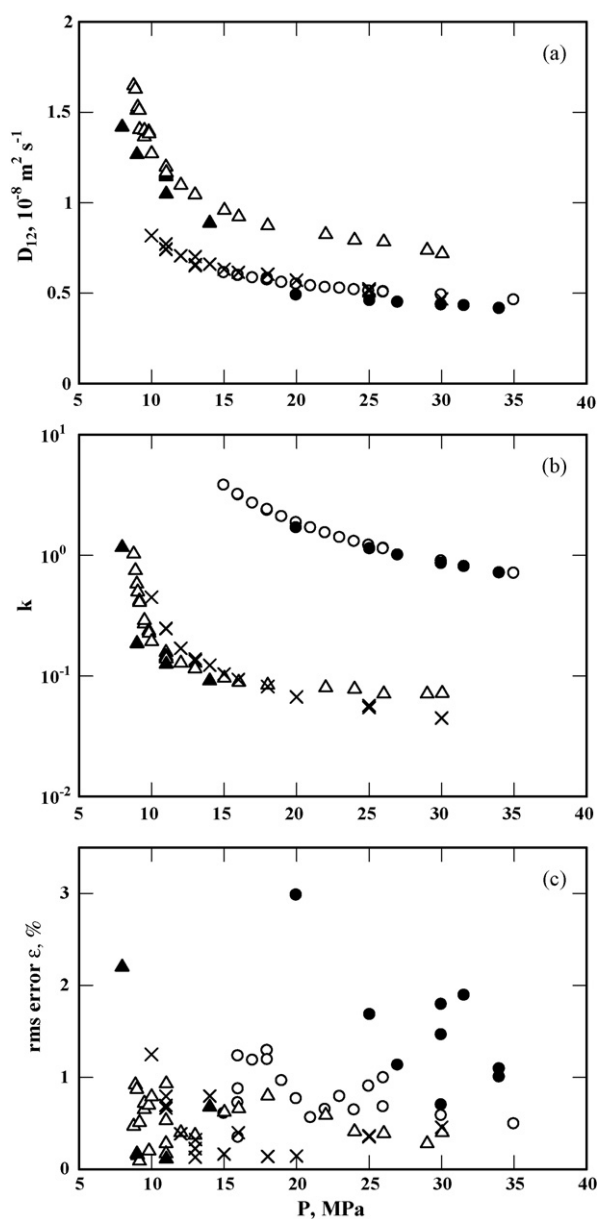


Fig. 1. Pressure dependences on (a)  $D_{12}$ , (b)  $k$ , and (c)  $\varepsilon$  for dibenzo-24-crown-8 (circle) and 15-crown-5 (triangle) at 308.18 K (solid key) and 313.20 K (blank key), and vitamin K<sub>1</sub> (cross) at 313.20 K.

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