



Aromatic interactions at the catalytic subsite of sucrose phosphorylase: Their roles in enzymatic glucosyl transfer probed with Phe⁵² → Ala and Phe⁵² → Asn mutants

Patricia Wildberger, Christiane Luley-Goedl, Bernd Nidetzky*

Institute of Biotechnology and Biochemical Engineering, Graz University of Technology, Petersgasse 12/1, A-8010 Graz, Austria

ARTICLE INFO

Article history:

Received 3 November 2010

Revised 23 December 2010

Accepted 29 December 2010

Available online 8 January 2011

Edited by Judit Ovádi

Keywords:

Sucrose phosphorylase

GH-13

Cation- π interaction

Oxocarbenium ion

Aromatic stacking

Glycoside hydrolase

Glycosyltransferase

ABSTRACT

Mutants of *Leuconostoc mesenteroides* sucrose phosphorylase having active-site Phe⁵² replaced by Ala (F52A) or Asn (F52N) were characterized by free energy profile analysis for catalytic glucosyl transfer from sucrose to phosphate. Despite large destabilization (≥ 3.5 kcal/mol) of the transition states for enzyme glucosylation and deglucosylation in both mutants as compared to wild-type, the relative stability of the glucosyl enzyme intermediate was weakly affected by substitution of Phe⁵². In reverse reaction where fructose becomes glucosylated, “error hydrolysis” was the preponderant path of breakdown of the covalent intermediate of F52A and F52N. It is proposed, therefore, that Phe⁵² facilitates reaction of the phosphorylase through (1) positioning of the transferred glucosyl moiety at the catalytic subsite and (2) strong cation- π stabilization of the oxocarbenium ion-like transition states flanking the covalent enzyme intermediate.

© 2011 Federation of European Biochemical Societies. Published by Elsevier B.V. All rights reserved.

1. Introduction

Recognition of carbohydrates by proteins often involves π -interactions from aromatic side chains of Phe, Tyr, and Trp. Aromatic rings undergo stacking with pyranoses and furanoses whereby their π -electron cloud interacts with aliphatic protons of the sugar ring [1–4]. The resulting interaction energies were reported to vary between -0.5 and -0.8 kcal/mol, depending on the nature of the aromatic ring and the carbohydrate [5]. Not surprisingly, therefore, carbohydrate- π interactions are widely utilized by glycoside hydrolases (e.g. [6,7]), glycosyltransferases (e.g. [8–10]) and carbohydrate-binding modules [11,12] to achieve recognition and specificity for their natural substrates. A striking commonality among the diverse families of glycoside hydrolases is the occurrence of a family-specific hydrophobic motif in their catalytic subsites. This motif usually consists of one or more aromatic residues and is found irrespective of whether the glycoside

hydrolase utilizes an α or β -configured pyranosyl substrate and hydrolyzes it with retention or inversion of anomeric configuration [13]. Its proposed role in catalysis is to aid in the conformational itinerary of the glycopyranoside upon moving from the ground state (e.g. 4C_1 chair) to the oxocarbenium ion-like transition state (e.g. 4H_3 half-chair). Non-polar contacts with the hydrophobic C₄–C₅-hydroxymethyl moiety of the glycopyranoside that become tightened in the transition state are believed to be the source of the catalytic facilitation provided. In addition to this conformational stabilization derived from carbohydrate- π interactions, there is the interesting question if aromatic side chains contributed to the catalytic subsite might also be important for electrostatic stabilization of the transition state of the glycoside hydrolase reaction.

Family GH-13 constitutes a large group of retaining hydrolases and transglycosidases that act on α -O-glycosidic substrates such as starch and sucrose [14]. Its members contain a special aromatic motif (Fig. 1; [15–21]) where a highly conserved Phe/Tyr is positioned at the B-face of the glucopyranosyl ring bound in the catalytic subsite. Sucrose phosphorylase (from *Leuconostoc mesenteroides*; LmSPase) was chosen for examining with site-directed mutagenesis the function of the relevant aromatic residue (Phe⁵²) in the catalytic mechanism. Note: the aromatic motif in Fig. 1 must be distinguished from the likewise present hydrophobic

Abbreviations: Glc1P, α -D-glucopyranosyl phosphate; Glc1F, α -D-glucopyranosyl fluoride; SPase, sucrose phosphorylase (EC 2.4.1.7); BaSPase, *Bifidobacterium adolescentis* SPase; LmSPase, *Leuconostoc mesenteroides* SPase

* Corresponding author. Fax: +43 316 873 8434.

E-mail address: bernd.nidetzky@tugraz.at (B. Nidetzky).

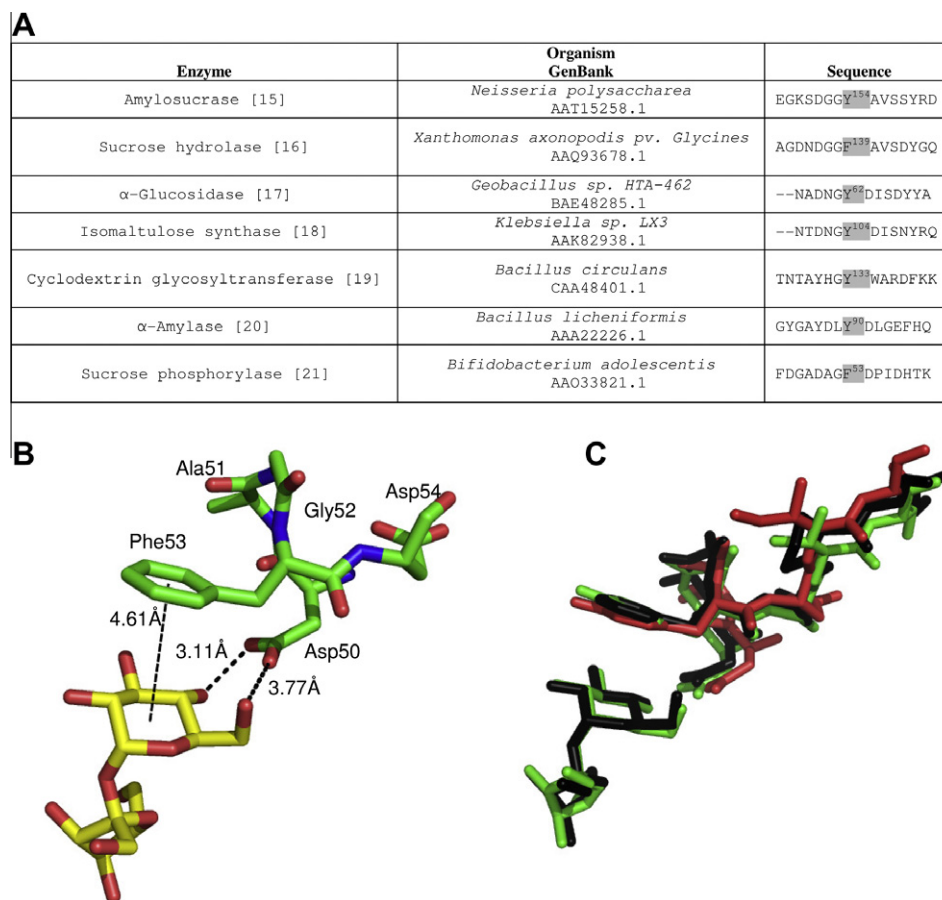


Fig. 1. The aromatic motif at the catalytic subsite of selected glucoside hydrolases and transglucosidases from family GH-13, revealed by Phe/Tyr residue conservation in both sequence (A) and three-dimensional structure (B and C). Panel B shows relevant interactions with the glucopyranosyl moiety of sucrose in the catalytic subsite of *B. adolescentis* sucrose phosphorylase (BaSPase). Note: Phe⁵² of LmSPase is homologous to the shown Phe⁵³ of BaSPase. Panel C is a structural overlay of the corresponding catalytic subsite parts of BaSPase (2GDU; black), *N. polysaccharea* amylosucrase (1JG1; green) and *Klebsiella* sp. isomaltulose synthase (1M53; red).

motif (Phe¹⁶⁰ in LmSPase) of conformational stabilization [13] that is not considered herein. LmSPase belongs to a particular group of transglucosidases within family GH-13 that promote reversible glucosyl transfer from sucrose to phosphate (P_i), yielding α -D-glucopyranosyl phosphate (Glc1P) and D-fructose as products [22]. Like in other GH-13 enzymes [23–26], catalytic reaction of LmSPase involves formation (*glucosylation*) and breakdown (*deglycosylation*) of a covalent β -glucosyl enzyme intermediate [27]. The proposed role of Phe⁵² involves stabilization, by strong cation- π interactions, of the oxocarbenium ion-like transition states leading to and from the glucosylated enzyme.

2. Materials and methods

Materials used have been described elsewhere [22]. α -D-Glucopyranosyl fluoride (Glc1F) was prepared by Zemplén deprotection from 2,3,4,6-tetra-O- α -D-glucopyranosyl fluoride that was obtained from TCI Europe.

2.1. Site-directed mutagenesis and enzyme preparation

Mutations resulting in substitution of Phe⁵² by Ala (F52A) or Asn (F52N) were introduced using a reported two-stage PCR protocol [28] in which a pQE30 expression vector encoding wild-type LmSPase was used as template [22]. Oligonucleotide primers and details of conditions used for PCR are summarized in [Supplementary data](#). For expression, we subcloned native and

mutated genes, via *Bam*HI and *Pst*I restriction sites, from pQE30 into pASK-IBA7+ (IBA GmbH). The pASK-IBA7+ constructs introduce *Strep*-tag II at the N-terminus of each enzyme. The new tag (8 amino acids) is comparable in length to the previously used His-tag (11 amino acid), which however did not facilitate protein isolation [22]. All inserts were confirmed by sequencing in sense and antisense directions. *Escherichia coli* Top10 cells harboring pASK-IBA7+ construct were cultivated under standard conditions ([Supplementary data](#)) using induction with 200 μ g/l anhydrotetracycline (6 h, 25 °C). Crude cell extract was prepared with a French press and protein was purified using chromatography on a Strep-Tactin Superflow column (bed volume 5 ml) (for details, see the [Supplementary data](#)). Purification was monitored by SDS-PAGE.

2.2. Assays

Enzyme activity in direction of phosphorolysis of sucrose was determined at 30 °C using a continuous coupled enzymatic assay with phosphoglucomutase and glucose-6-phosphate dehydrogenase [22]. Protein concentration was measured with the BioRad dye-binding method referenced against BSA. Glucose was determined using an assay with hexokinase and glucose-6-phosphate dehydrogenase [22]. Glucose in the presence of fructose was determined using an assay with glucose oxidase and peroxidase [29]. Glc1P was assayed using phosphoglucomutase and glucose-6-phosphate dehydrogenase [22]. Phosphate was determined colorimetrically at 850 nm [22]. Other analytical methods (NMR, HPLC)

Download English Version:

<https://daneshyari.com/en/article/2048164>

Download Persian Version:

<https://daneshyari.com/article/2048164>

[Daneshyari.com](https://daneshyari.com)