



Enhancement in thermal property and mechanical property of phase change microcapsule with modified carbon nanotube



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HIGHLIGHTS

- Carbon nanotubes was grafted and used to enhance the thermal conductivities of the microcapsules.
- The average particle size of the prepared MicroPCMs/CNTs-SA is 0.1 μm.
- The thermal conductivity of MicroPCMs/CNTs-SA with 4% of CNTs increased by 79.2% compared with MicroPCMs.
- MicroPCMs/CNTs-SA has better durability and thermal stability compared to the original MicroPCMs.

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ABSTRACT

Carbon nanotubes grafted with stearyl alcohol (CNTs-SA) was used in synthesizing phase change microcapsules (MicroPCMs) in order to enhance the thermal conductivities of the microcapsules. Urea-formaldehyde resin (UFR) was used as wall material. Scanning Electron Microscope (SEM), laser particle size analyzer, Fourier Transform Infrared Spectroscopy (FTIR), Differential Scanning Calorimeter (DSC) are employed to characterize the prepared MicroPCMs containing the grafted CNTs (MicroPCMs/CNTs-SA). The results indicated that CNTs improved the performance of microcapsules. The average particle diameter of MicroPCMs/CNTs-SA is much smaller than that of MicroPCMs. There was no chemical reaction among paraffin, CNTs and UFR. The phase change temperature and latent heat of MicroPCMs/CNTs-SA was 26.2 °C and 47.7 J/g, respectively. The thermal conductivity of MicroPCMs/CNTs-SA with 4% of CNTs increased by 79.2% compared with MicroPCMs. The initial decomposition temperature of MicroPCMs/CNTs-SA is 38 °C higher than that of MicroPCMs. After 100 heating and cooling cycles, MicroPCMs/CNTs-SA still has good durability and thermal stability.

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1. Introduction

Research on microencapsulated phase change materials (PCMs) began in the 1970s. PCMs are encapsulated by natural or synthetic macromolecular to form spherical particles with diameter from 1 to 1000 μm [1]. Phase change occurs within microcapsules and heat are absorbed or released. Consequently, the purpose of adjusting temperature or storing thermal energy was achieved [2]. Microencapsulation can not only solve the volume change in solid–liquid phase transition, but also prevent PCMs from being exposed to the external environment [3].

Phase change microcapsules (MicroPCMs) have drawn greater attention than traditional PCMs. Ahmet Sari used methyl meth-

acrylate as wall material to prepare microcapsules with latent heat of 86.4 J/g and coating rate of 43% [4]. Zhang found that microcapsules using styrene maleic anhydride copolymer (SMA) as emulsifier exhibited good phase change characteristics, high encapsulation efficiency and excellent stability [5]. Wei used melamine–formaldehyde resin as wall material to prepare microcapsules through in situ polymerization. The microcapsules had an average diameter of 2.2 μm, the latent heat of 144 J/g and the encapsulation efficiency of 59% [6]. Ma et al. [7] prepared a series of MicroPCMs with butyl stearate and paraffin as a binary core material and poly(methyl methacrylate-co-divinylbenzene) copolymer as shell material. The binary core material content in MicroPCMs was in the range between 50% and 85%. The MicroPCMs decomposed above 200 °C. Li et al. [8] fabricated a series of MicroPCMs with polyurethane shell. The fusion heat of polyurethane MicroPCMs was less than 60 J/g.

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