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Analysis of the hygrothermal functional properties of stabilised rammed earth materials

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ABSTRACT

Suitable experimental methodologies for determining the hygrothermal properties of stabilised rammed earth (SRE) materials have been presented along with comparative experimental data for three different SRE mix designs with parametric analysis of the influence of these variables on material function. Higher bulk porosity corresponds to reduced volumetric heat capacity (*C*), but increased sorptivity (*S*) and vapour permeance (*W*). Since bulk porosity and void size distribution (VSD) are interdependent variables, it follows that for constant particle size distribution (PSD) and compaction energy an increase in porosity results in an increase in the mean pore radius, \bar{r} for a material. This explains why the magnitude of liquid/vapour transfer (*S* and *W*) terms are inversely related to the hygroscopic moisture capacity, ξ since the capillary potential, Ψ will increase when the mean pore diameter decreases. The implications are that the hygrothermal properties of SRE materials can be designed and predicted by manipulating particle size distribution and compaction energy.

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1. Introduction

Approximately one half of the world's population are said to live or work in an earth building [1] and it has long been mooted that earth materials have the ability to provide superior levels of indoor thermal comfort, e.g. Refs. [2-4]. Stabilised rammed earth (SRE) is the modern practice of a traditional construction technique and is typically achieved by the addition of <10% Portland cement to subsoil and dynamic compaction (ramming) of the mixture into temporary formwork. It provides a low embodied energy material for rapid construction on-site. Its modern day use is widespread across Australasia, North America, Asia, and parts of mainland Europe such as Spain, Germany and France and increasingly Great Britain. Since earth is a porous, hygroscopic material that contains active clay minerals, it is expected that the walls absorb water vapour from the air when relative humidity increases and release this moisture when the humidity falls. As it is usual practice to leave earth walls exposed to the interior of the building, the availability of combined thermal and hygric buffering (i.e. passive air conditioning) will be maximised when compared with other materials such as brick, timber or concrete that are often covered over with more insulating and less permeable coverings, e.g. varnish and paints.

The functional properties of SRE that can be used to describe its hygrothermal behaviour are the moisture storage function, vapour permeability, liquid conductivity, thermal conductivity, and specific heat capacity. Samples of three SRE mix designs were manufactured and tested to determine these functional properties and the results compared.

The objective of this paper is to investigate how the hygrothermal properties of SRE materials can be experimentally measured and how these functional properties are influenced by the variable mix design parameters of the material. The aim is to test the hypothesis that hygrothermal properties can then be predicted and designed through correspondence with mix design parameters. SRE materials can be characterised as multiphase granular composites whose particle size distribution (PSD) and particle packing efficiency largely determine the geometry of their matric structure. The term stabilisation refers to the application of a process and/or additive component that enhances the cohesion, Young's modulus or another physical property [5–7]. By far the most common forms of stabilisation are (i) dynamic compaction and (ii) addition of hydraulic binders, e.g. cementicious materials. As with most granular soils, dynamic compaction close to the Proctor optimum moisture content increases the inter-particle friction/interlock whilst reducing the bulk porosity. The addition of hydraulic binders (commonly <10% Portland cement) increases the





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Nomenclature		D	diffusion coefficient for water vapour in still air (m^2/s) .
		$C_{\rm v}$	mass concentration of water vapour (kg/m ³).
$V_{\rm T}$	total volume.	M_{w}	molar mass of water (kg/kg mol).
Vs	volume of solid.	Ro	the universal gas constant (J/kg mol K).
$V_{\rm v}$	volume of void.	$m_{\rm w}$	mass of water (kg).
Va	volume of air.	Α	area.
Vm	molar volume.	G	water vapour flow rate (kg/s).
p_{v}	vapour pressure.	W	water vapour permeance (kg/m ² s Pa).
p_{sat}	saturation vapour pressure.	i	cumulative volume of absorbed water per unit inflow
p_{sat}^*	saturation vapour pressure above a flat surface.		surface area (mm³/mm²).
Pa	total air pressure.	t	time.
P_{w}	total water pressure.	S	sorptivity (mm/min ^{-0.5}).
r	pore radius.	ξ	hygroscopic moisture storage function.
r _{crit}	critical pore radius.	Ψ	capillary potential.
r	mean pore radius.	λ	thermal conductivity (W/m K).
Т	thermodynamic temperature (K).	λ^*	moisture content-dependent thermal conductivity (W/
η	viscosity.		m K).
γ	surface tension.	<i>q</i> _{sens}	sensible heat flow (W/m^2) .
φ	relative humidity ($0 \le \varphi \le 1$).	$q_{\rm lat}$	latent heat flow (W/m ²).
RH	relative humidity % ($\phi imes$ 100).	w	reference moisture content (kg/m ³).
θ	relative moisture (0 $\leq \theta \leq$ 1).	$g_{\rm v}$	rate of water vapour transfer (kg/s m ²).
$\theta_{\rm r}$	residual moisture content.	h_{e}	specific latent enthalpy of evaporation/condensation
θ_{c}	capillary saturation moisture content.		(J/kg).
θ_{AEV}	air entry value moisture content.	Sr	degree of saturation (or saturation ratio).
θ_{s}	saturation moisture content.	$m_{ m f}$	moisture factor.
$ ho_{w}$	density of water (kg/m ³).	Cp	specific heat capacity at constant pressure (J/kg K).
$ ho_{ m v}$	density of water vapour (kg/m ³).	$c_{\rm p}^{*}$	moisture content-dependant specific heat capacity at
g	gravitational constant.		constant pressure (J/kg K).
h	height.	$\overline{\omega}$	relative mass fraction.
$g_{ m v,\ air}$	rate of water vapour diffusion through air.		

internal cohesion of the material and enhances durability and toughness [5–9]. Since hardened cement paste bonds particles together by surface adhesion between the paste and particle surfaces (interfacial transition zone), cement stabilisation is most effective on granular soils [8–10] where the specific surface area per unit mass is lower and the greater absence of cohesive materials is less able to interfere with the interfacial transition zone.

2. Specimen preparation

The characterisation of SRE material composition can easily be approximated using the geotechnical soil model. The particle size distribution for SRE must normally fall within designated upper and lower limits resulting in a wide range of achievable particle packing efficiencies and associated bulk porosity/dry density and void size distribution (VSD), as explained extensively in the authors' previous research [11-13]. By using an established technique of blending characterised soil constituents (14-6.3 mm gravel, 5 mm down medium grit sand, silty clay) and matching the net PSD to the defined upper/lower limits, the authors can maintain the parameters of aggregate mineralogy, particle angularity and clay mineralogy as constants whilst keeping PSD as the single variable. SRE specimens were prepared as 11 cylinders (3 compaction layers), 1/31 discs (1 compaction layer), and 300×300 mm slabs (1 compaction layer). All were stabilised by compacting at OMC using a constant energy of 596 kJ/m³ and the addition of 6% wt CEM IIa class Portland cement, followed by air curing for a minimum of 28 days at 20 $^{\circ}C\,(\pm2\,^{\circ}C)$ and 75% RH $(\pm5\%)$ in an environmental chamber.

3. Moisture sorption and storage

The assumption here for hygrothermal materials is that a representative microstructure of porous media has a total volume $(V_{\rm T})$ which consists of solid state matter $(V_{\rm S})$ and fluid-filled void (V_v) , where $V_T = V_s + V_v$. Under atmospheric conditions, when the material is dry the voids are filled with air $(V_v = V_a)$ and when saturated the voids are filled with water $(V_v = V_w)$. Clearly, when moisture enters a dry or unsaturated continuum it must displace air from the moment it crosses the boundaries defined by $V_{\rm T}$; a process referred to as 'absorption'. Absorbed moisture vapour may also be 'adsorbed' to the internal surfaces by van der Waals forces. Absorbed moisture may be classified into one of three domains (hygroscopic, capillary, and gravitational) depending upon (i) its phase when it enters $V_{\rm T}$ and (ii) its strength of electrostatic attraction. Hygroscopic moisture is absorbed in the vapour phase, capillary moisture is absorbed in the liquid phase, and gravitational moisture is liquid that is absorbed when the capillary potential in the pore network is zero (i.e. super saturation).

Although the classification of absorbed moisture (determined by its phase upon entry) cannot change, its phase once inside $V_{\rm T}$ can change, e.g. condensing from vapour to liquid. This is chiefly governed by internal void geometry and electrostatic surface charge for a given temperature and partial vapour pressure. Theoretically, the saturation vapour pressure above a flat surface of liquid water ($p_{\rm sat}^*$, when $r = \infty$) is dependent upon the pressure applied ($\Delta P_{\rm a}$) to that liquid by a surrounding gas, in this case air. For a given pressure differential at the liquid/gas interface, $\Delta P_{\rm a}$, the saturation vapour pressure becomes $p_{\rm sat} = p_{\rm sat}^* e^{V_{\rm m} \Delta P/R_o T}$ [14]. By curving the surface of the water, as in a meniscus in a capillary or pore, to a known radius, r

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