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## Triterpenoids from the roots of Rubus parvifolius

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**[ABSTRACT]** Two new oleanane-type triterpenoids, parvifolactone A (1) and rubuside P (2), together with 11 known triterpenoids, fupenzic acid (3), 18,19-seco, $2\alpha$ , $3\alpha$ -dihydroxyl-19-oxo-urs-11,13(18)-dien-28-oic acid (4), euscaphic acid (5), maslinic acid (6), 1 $\beta$ -hydroxyeuscaphic acid (7),  $2\alpha$ , $3\alpha$ , $19\alpha$ ,23-tetrahydroxyolean-12-en-28-oic acid (8),  $2\alpha$ , $3\beta$ , $19\alpha$ ,23-tetrahydroxyurs-12-en-28-oic acid (9), glucosyl pinfaensate (10), rubuside J (11),  $2\alpha$ , $3\alpha$ , $19\alpha$ ,23-tetrahydroxyurs-12-en-24,28-dioic acid (12), and  $2\alpha$ , $3\beta$ , $19\alpha$ -trihydroxyurs-12-en-23,28-dioic acid (13), were isolated from the roots of *Rubus parvifolius*.

[KEY WORDS] Triterpenoid; Parvifolactone A; Rubuside P; Rubus parvifolius[CLC Number] R284.1[Document code] A[Article ID] 2095-6975(2016)05-0377-05

## Introduction

*Rubus parvifolius* belonging to the family Rosaceae is a small shrub widely distributed in China. The roots of *R. parvifolius* are widely used for the treatment of rheumatism, hepatitis, and abdominal pain caused by postpartum stasis <sup>[1]</sup>. Previous investigations on the chemical constituents of the roots of *R. parvifolius* led to the isolation and identification of triterpenoids and flavonoids <sup>[2-10]</sup>. As part of ongoing effort to search for bioactive compounds from natural plants, the chemical constituents of *R. parvifolius* were investigated in the present study. Herein, we report the isolation and structural elucidation of two new triterpenoids, parvifolactone A (1) and rubuside P (2), together with 11 known triterpenoids, from the roots of *R. parvifolius*. All compounds but **5** and **13** were firstly isolated from the roots of *R. parvifolius*.

## **Results and Discussion**

Compound 1 was obtained as an amorphous powder. The

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positive HR-ESI-MS gave a  $[M + H]^+$  ion peak at m/z469.337 6. in accordance with an empirical molecular formula  $C_{30}H_{44}O_4$ , which was supported by the <sup>13</sup>C NMR spectroscopic data (Table. 1). The IR absorptions at  $3 440 \text{ cm}^{-1}$  and  $1 774 \text{ cm}^{-1}$  indicated the presence of hydroxyl group and carbonyl group, respectively. The <sup>1</sup>H NMR spectrum of 1 indicated the presence of seven methyl singlets at  $\delta_{\rm H}$  0.77, 0.87, 0.93, 0.94, 1.04, 1.07, and 1.29, three oxymethine protons at 4.42 (1H, m), 3.82 (1H, d, J = 2.5 Hz), and 4.93 (1H, s), and two olefinic protons at  $\delta_{\rm H}$  5.96 (1H, dd, J = 10.0, 2.0 Hz) and 6.26 (1H, d, J = 10.0 Hz). The <sup>13</sup>C NMR spectrum of 1 showed the presence of 30 carbons, including two tertiary olefinic carbons at  $\delta_{\rm C}$  123.3 and 130.3, two quaternary olefinic carbons at  $\delta_C$  133.4 and 135.3, two oxymethine carbons at  $\delta_{\rm C}$  66.1 and 79.6, one oxygen-bearing tertiary carbon at  $\delta_C$  85.1, and one carbonyl carbon at  $\delta_C$ 178.0. The assignment of protons and carbons of 1 was achieved by HSQC, HMBC, and NOESY experiments (Table 1).

In the HMBC spectrum of **1** (Fig. 1), the observation of the long-range correlations between H-19 ( $\delta_{\rm H}$  4.93) and C-28 ( $\delta_{\rm C}$  178.0) indicated the occurrence of a five-membered lactone ring in **1**. The four unsaturated carbons ( $\delta_{\rm C}$  123.3, 130.3, 133.4, and 135.3) contributed to two conjugated double bonds were unambiguously assigned by the HMBC correlations from H-11 ( $\delta_{\rm H}$  5.96, dd, J = 10.0, 2.0 Hz) to C-8 ( $\delta_{\rm C}$  41.7) and C-13 ( $\delta_{\rm C}$  135.3), from H-12 ( $\delta_{\rm H}$  6.26, d, J = 10.0 Hz)



position	1		2	
	$\delta_{\rm H} (J \text{ in Hz})^*$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})^*$	$\delta_{C}$
1	1.83, m; 2.25, dd (11.5, 4.0)	43.1 (CH <sub>2</sub> )	1.38, m; 2.29, dd (13.0, 4.0)	47.7 (CH <sub>2</sub> )
2	4.42, m	66.1 (CH)	4.24, m	68.3 (CH)
3	3.82, d (2.5)	79.6 (CH)	4.04, m	77.3 (CH)
4	-	39.2 (C)	-	56.9 (C)
5	1.75, m	48.5 (CH)	2.09, m	48.6 (CH)
6	1.59, m	18.4 (CH <sub>3</sub> )	1.05, m; 1.55, m	21.2 (CH <sub>2</sub> )
7	1.48, m; 1.54, m	33.6 (CH <sub>2</sub> )	1.36, m; 1.56, m	32.8 (CH <sub>2</sub> )
8	-	41.7 (C)	-	40.6 (C)
9	2.37, brs	53.4 (CH)	1.66, m	48.5 (CH)
10	-	38.5 (C)	-	38.8 (C)
11	5.96, dd (10.0, 2.0)	130.3 (CH)	2.09, m	24.5 (CH <sub>2</sub> )
12	6.26, d (10.0)	123.3 (CH)	5.51, brs	123.7 (CH)
13	_	135.3 (C)	_	144.7 (C)
14	-	41.1 (C)	_	42.4 (C)
15	1.15, m; 1.40, m	26.1 (CH <sub>2</sub> )	1.05, m; 1.22, m	29.2 (CH <sub>2</sub> )
16	2.46, m	24.9 (CH <sub>2</sub> )	2.13, m	28.2 (CH <sub>2</sub> )
17	-	44.5 (C)	_	46.7 (C)
18	-	133.4 (CH)	3.54, br.s	44.9 (CH)
19	4.93, s	85.1 (CH)	3.59, m	81.2 (CH)
20	-	36.1 (C)	_	35.8 (C)
21	1.28, m; 1.40, m	33.1 (CH <sub>2</sub> )	1.05, m; 1.22, m	29.2 (CH <sub>2</sub> )
22	1.51, m; 1.75, m	35.0 (CH <sub>2</sub> )	1.96, m; 2.06, m	33.3 (CH <sub>2</sub> )
23	1.29, s	29.5 (CH <sub>3</sub> )	9.66, s	206.7 (C)
24	0.94, s	22.0 (CH <sub>3</sub> )	1.45, s	10.9 (CH <sub>3</sub> )
25	1.04, s	19.7 (CH <sub>3</sub> )	1.08, s	17.3 (CH <sub>3</sub> )
26	0.77, s	17.4 (CH <sub>3</sub> )	1.15, s	17.8 (CH <sub>3</sub> )
27	0.93, s	19.4 (CH <sub>3</sub> )	1.62, s	25.2 (CH <sub>3</sub> )
28	-	178.0 (C)	-	177.5 (C)
29	1.07, s	28.0 (CH <sub>3</sub> )	0.99, s	24.9 (CH <sub>3</sub> )
30	0.87, s	23.3 (CH <sub>3</sub> )	1.15, s	29.0 (CH <sub>3</sub> )
Glc1			6.40, d (8.0)	96.1 (CH)
2			4.24, m	74.4 (CH)
3			4.04, m	79.6 (CH)
4			4.38, m	71.3 (CH)
5			4.30, m	79.2 (CH)
6			4.44, m	62.4 (CH <sub>2</sub> )

Table 1 NMR data of Compounds 1 and 2 (500 MHz for <sup>1</sup>H , 125 MHz for <sup>13</sup>C, in pyridine-d<sub>5</sub>)

to C-9 ( $\delta_C$  53.4), from H-27 ( $\delta_C$  0.93, s) to C-13 ( $\delta_C$  135.3), and from H-19 ( $\delta_H$  4.93, s) to C-13 ( $\delta_C$  135.3) and C-18 ( $\delta_C$  133.4), respectively. The two oxymethine carbons at  $\delta_C$  66.1 (C-2) and 79.6 (C-3) were assigned by the HMBC correlations of H-2/C-1, H-2/C-3, and H-23/C-3. Comparison of the NMR data of **1** with those of the known compound  $2\alpha$ ,  $3\beta$ -dihydroxyolean-11, 13(18)-dien-19 $\beta$ , 28-olide suggested that these two compounds shared a similar skeleton [<sup>11</sup>].

However, the notable difference of the coupling constants between H-2 and H-3 of these two compounds ( $J_{2,3}$  = 2.5 Hz and 9.6 Hz, respectively) suggested that the configuration of the hydroxy groups of these two compounds might be different. In the NOESY spectrum of **1**, the cross-peaks of H-2/H-25, H-2/H-24, H-2/H-3, and H-19/H-29 allowed the assignment of  $\beta$ -orientation of H-2 and H-3, and the  $\alpha$ -orientation of H-19. Therefore, the structure of **1** was Download English Version:

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