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Bis-spirolabdane diterpenoids from *Leonurus japonicus* and their anti-platelet aggregative activity

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1. Introduction

Leonurus japonicus Houtt. (Labiatae) is widely distributed and cultivated in China, particularly in Anhui, Jiangsu, Sichuan, and Zhejiang provinces. The aerial parts of *L. japonicus*, "vi mu cao" in Chinese, are often used to treat various diseases such as blood stasis, menstrual disturbance, dysmenorrhea, and amenorrhea [1,2]. Previous investigations on *L. japonicus* (synonyms Leonurus heterophyllus) have led to the isolation of many diterpenoids, particularly the spirolabdane diterpenoids [3–7]. However, their anti-platelet aggregation activities in relation to the traditional use of *L. japonicus* have rarely been reported [8]. In our search for bioactive diterpenoids, the EtOAc extract of L. japonicus was phytochemically investigated, affording six new bis-spirolabdane diterpenoids (1-6) and four known analogues (7-10) (Fig. 1). Their inhibitory activities against platelet aggregation were found to depend on the absolute configuration of the C-13 position in these compounds.

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ABSTRACT

Six bis-spirolabdane diterpenoids along with four known analogues were isolated from the aerial parts of *Leonurus japonicus*. Their structures and absolute configurations were elucidated by spectroscopic analyses, single-crystal X-ray diffraction, and a modified Mosher's method. The inhibitory activity of the compounds against the abnormal increase in platelet aggregation induced by adenosine diphosphate was investigated. Only the (13*R*)-bis-spirolabdane diterpenoids exhibited a significant effect.

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2. Experimental

2.1. General

Optical rotations were measured using a Perkin-Elmer 341 plus. IR spectra were recorded using a Nicolet 5700 FT-IR microscope instrument. NMR spectra were obtained using a Bruker-AVIIIHD-600 spectrometer with the solvent peaks used as the references. HRESIMS spectra were measured using a Waters Synapt G2 HDMS. Column chromatography (CC) was performed using silica gel (200-300 mesh, Yantai Institute of Chemical Technology, Yantai, China), MCI gel CHP 20P (75-150 µm, Mitsubishi Chemical, Co., Japan), and Sephadex LH-20 (Amersham Pharmacia Biotech AB, Uppsala, Sweden). HPLC separation was performed using an instrument equipped with a Cometro 6000LDS pump, a Cometro 6000PVW UV/VIS detector, and an Ultimate $(250 \times 10 \text{ mm}^2)$ preparative column packed with C18 (5 µm). TLC was performed using glass precoated silica gel GF254 plates (Qingdao Marine Chemical Inc., Qingdao, China).









Fig. 1. Chemical structures of compounds 1-10.

2.2. Plant material

The aerial parts of *L. japonicus* were collected in May of 2012 from a field in Wenjiang District, Chengdu City, Sichuan Province, China. The herb was identified by Prof. Min Li (Chengdu University of TCM, Sichuan, China). A voucher specimen (SYMC-0522) was deposited at the School of Pharmacy, Chengdu University of TCM, China.

2.3. Extraction and isolation

The air-dried aerial parts of L japonicus (20 kg) were extracted with 95% EtOH (3×160 L) at room temperature for 3×72 h. The EtOH extract was evaporated under reduced pressure to afford a dark brown residue (1.2 kg). The residue was suspended in H₂O and successively partitioned into EtOAc (400 g) and *n*-BuOH (160 g) fractions. The EtOAc fraction was subjected to silica gel CC using a gradient elution of petroleum ether/acetone (100:1–0:1, v/v) to afford 19 fractions (F_1 – F_{19}) based on TLC analyses. F_8 (8.2 g) was subjected to MCI with a gradient of increasing EtOH (50%-100%) in water to afford five subfractions ($F_{8-1}-F_{8-5}$). F_{8-3} (2.1 g) was further separated by RP-MPLC affording seven parts, F₈₋₃₋₁-F₈₋₃₋₇. F₈₋₃₋₂ was purified by RP semipreparative HPLC (73% MeOH in H₂O) to afford mixed crystals **9** and **10** (2.8 mg). F₈₋₃₋₃ was purified via PTLC (petroleum ether/acetone, 5:1 v/v) followed by RP semipreparative HPLC (72% MeCN in H₂O) purification to afford **4** (1.5 mg, t_R = 35 min) and **6** (2.4 mg, t_R = 39 min). The separation of F₈₋₃₋₄ using Sephadex LH-20 (petroleum ether/ CHCl₃/MeOH, 5:5:1 v/v/v) and RP semipreparative HPLC (76% MeCN in H₂O) successively afforded compounds **1** (13 mg, $t_R =$ 42 min), **2** (16.5 mg, $t_R = 26$ min), **3** (9 mg, $t_R = 29$ min), **5** (1.6 mg, $t_R = 47$ min), **7** (15 mg, $t_R = 55$ min), and **8** (22 mg, $t_R = 61$ min).

 $\begin{array}{l} (-)-(3R,5S,7R,8R,9R,10S,13R,15R)-3-Acetoxy-7-hydroxy-\\ 15-ethoxy-9,13;15,16-diepoxylabdan-6-one (1): Colourless crystals (acetone); [<math>\alpha$]_D²⁰ - 58.8 (*c* 0.10, MeOH); IR ν_{max} 3456, 2977, 2922, 2881, 1733, 1708, 1373, 1241, 1307, 904, 800 cm⁻¹; ¹H NMR (Me₂CO-*d*₆, 600 MHz) data see Table 1; ¹³C NMR (Me₂CO-*d*₆, 150 MHz) data see Table 2; and (+)-HRESIMS m/z 461.2520 [M + Na]⁺ (calcd. for C₂₄H₃₈O₇Na, 461.2515).

(-)-(5*S*,7*R*,8*R*,9*R*,10*S*,13*R*,15*R*)-7-Hydroxy-15-ethoxy-9,13;15,16-diepoxylabdan-6-one (**2**): Colourless oil; $[\alpha]_D^{20}$ -92.2 (*c* 0.15, MeOH); IR ν_{max} 3471, 2978, 2924, 2873, 1707, 1472, 1391, 1368, 1324, 1269, 1111, 1041, 995, 951, 906 cm⁻¹; ¹H

NMR (Me₂CO- d_6 , 600 MHz) data see Table 1; ¹³C NMR (Me₂CO- d_6 , 150 MHz) data see Table 2; and (+)-HRESIMS m/z 403.2455 [M + Na]⁺ (calcd. for C₂₂H₃₆O₅Na, 403.2460).

(+)-(5*S*,7*R*,8*R*,9*R*,10*S*,13*S*,15*R*)-7-Hydroxy-15-ethoxy-9,13;15,16-diepoxylabdan-6-one (**3**): Colourless oil; $[\alpha]_D^{20}$ + 114.9 (*c* 0.20, MeOH); IR ν_{max} 3466, 2979, 2925, 2872, 1707, 1468, 1391, 1368, 1324, 1256, 1107, 1044, 997, 952, 932 cm⁻¹; ¹H NMR (Me₂CO-*d*₆, 600 MHz) data see Table 1; ¹³C NMR (Me₂CO-*d*₆, 150 MHz) data see Table 2; and (+)-HRESIMS *m*/*z* 403.2456 [M + Na]⁺ (calcd. for C₂₂H₃₆O₅Na, 403.2460).

 $\begin{array}{l} (-) - (5S,7R,8R,9R,10S,13S,15S) - 7 - Hydroxy - 15 - methoxy - 9,13;15,16 - diepoxylabdan - 6 - one (4): Colourless oil; <math display="inline">[\alpha]_D^{20} - 69.4~(c~0.05,~MeOH);~IR~\nu_{max}~3471,~2926,~2857,~1708,~1461,~1376,~1204,~1127,~1105,~1040,~992~cm^{-1};~^{1}H~NMR~(Me_2CO-d_6,~600~MHz)~data~see~Table~1;~^{13}C~NMR~(Me_2CO-d_6,~150~MHz)~data~see~Table~2;~and~(+) - HRESIMS~m/z~389.2310~[M + Na]^+~(calcd.~for~C_{21}H_{34}O_5Na,~389.2304). \end{array}$

(-)-(3*R*,5*S*,7*R*,8*R*,9*R*,10*S*,13*S*,15*S*)-3-Acetoxy-7-hydroxy-15-methoxy-9,13;15,16-diepoxylabdan-6-one (**5**): Colourless oil; $[\alpha]_{20}^{D}$ - 38.3 (*c* 0.04, MeOH); IR ν_{max} 3454, 2976, 2923, 2856, 1733, 1707, 1439, 1371, 1250, 1102, 1039, 989 cm⁻¹; ¹H NMR (Me₂CO-*d*₆, 600 MHz) data see Table 1; ¹³C NMR (Me₂CO-*d*₆, 150 MHz) data see Table 2; and (+)-HRESIMS *m*/*z* 447.2366 [M + Na]⁺ (calcd. for C₂₃H₃₆O₇Na, 447.2359).

(+)-(5*S*,7*R*,8*R*,9*R*,10*S*,13*S*,15*R*)-7-Hydroxy-15-methoxy-9,13;15,16-diepoxylabdan-6,16-dione (**6**): Colourless oil; $[\alpha]_D^{20} + 45.2$ (*c* 0.05, MeOH); IR ν_{max} 3370, 2926, 2852, 1784, 1708, 1461, 1372, 1246, 1093, 1041, 937 cm⁻¹; ¹H NMR (Me₂CO-*d*₆, 600 MHz) data see Table 1; ¹³C NMR (Me₂CO-*d*₆, 150 MHz) data see Table 2; and (+)-HRESIMS *m*/*z* 403.2094 [M + Na]⁺ (calcd. for C₂₁H₃₂O₆Na, 403.2097).

2.4. X-ray crystallography data of compound 1

 $C_{24}H_{38}O_7$, *M* = 438.54, orthorhombic, *P*2₁2₁2₁, *a* = 10.8395(3) Å, *b* = 12.9176(4) Å, *c* = 17.2745(5) Å, *V* = 2418.76(12) Å³, *Z* = 4, *T* = 293(2) K, μ(CuKα) = 0.713 mm⁻¹, *D*_{calcd} = 1.204 g·cm⁻³, 21,236 reflections measured (8.548 ≤ 2Θ ≤ 134.616), and 4323 unique (*R*_{int} = 0.0598, *R*_{sigma} = 0.0276), which were used in all calculations; final *R*₁ = 0.0483, *wR*₂ = 0.1428.

The data were collected using an Xcalibur Atlas Gemini ultra diffractometer with CuK α radiation. The crystal structures were solved by direct methods using the SHELXS-97 programme and refined anisotropically by least-squares method using the SHELXL-97 refinement package [9]. The absolute

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