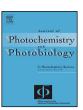
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#### Review

# Visible-light activation of TiO<sub>2</sub> photocatalysts: Advances in theory and experiments



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#### ABSTRACT

The remarkable achievement by Fujishima and Honda (1972) in the photo-electrochemical water splitting results in the extensive use of TiO2 nanomaterials for environmental purification and energy storage/conversion applications. Though there are many advantages for the TiO2 compared to other semiconductor photocatalysts, its band gap of 3.2 eV restrains application to the UV-region of the electromagnetic spectrum ( $\lambda \leq 387.5$  nm). As a result, development of visible-light active titanium dioxide is one of the key challenges in the field of semiconductor photocatalysis. In this review, advances in the strategies for the visible light activation, origin of visible-light activity, and electronic structure of various visible-light active TiO2 photocatalysts are discussed in detail. It has also been shown that if appropriate models are used, the theoretical insights can successfully be employed to develop novel catalysts to enhance the photocatalytic performance in the visible region. Recent developments in theory and experiments in visible-light induced water splitting, degradation of environmental pollutants, water and air purification and antibacterial applications are also reviewed. Various strategies to identify appropriate dopants for improved visible-light absorption and electron–hole separation to enhance the photocatalytic activity are discussed in detail, and a number of recommendations are also presented.

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acquiring more than €3 million direct R&D funding. Prof. Pillai is a recipient of a number of awards for research accomplishments including the 'Industrial Technologies Award 2011' from Enterprise Ireland for commercialising nanomaterials for industrial applications. He was also the recipient of the 'Hothouse Commercialisation Award 2009' from the Minister of Science, Technology and Innovation and also the recipient of the 'Enterprise Ireland Research Commercialization Award 2009'. He has also been nominated for the 'One to Watch' award 2009 for commercialising R&D work (Enterprise Ireland). One of the nanomaterials based environmental technologies developed by his research team was selected to demonstrate as one of the fifty 'innovative technologies' (selected after screening over 450 nominations from EU) at the first Innovation Convention organised by the European Commission on 5–6th December 2011. He is the national delegate and technical expert for ISO standardization committee and European standardization (CEN) committee on photocatalytic materials.

#### 1. Introduction

Photocatalysis refers to the acceleration of a chemical reaction in the presence of substances called photocatalysts, which can absorb light quanta of appropriate wavelengths depending on the band structure [1–4]. Usually semiconductors including  $TiO_2$ ,  $Fe_2O_3$ ,  $WO_3$ , ZnO,  $CeO_2$ , CdS,  $Fe_2O_3$ , ZnS,  $MoO_3$ ,  $ZrO_2$ , and  $SnO_2$  are selected as photocatalysts due to their narrow band gap and distinct electronic structure (unoccupied conduction band and occupied valence band) [5–24]. In semiconductor photocatalysis, electrons from the valence band of a semiconductor are excited to the conduction band by light of higher energy than the respective band gap, resulting in the formation of  $e^-_{CB}/h^+_{VB}$  pairs (Fig. 1). Conduction band electrons are good reducing agents (+0.5 to  $-1.5\,V$  vs. NHE) whereas the valence band holes ( $h^+_{VB}$ ) are strong oxidizing

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