

## Review

## Antimicrobial and photocatalytic disinfection mechanisms in silver-modified photocatalysts under dark and light conditions



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## ABSTRACT

The modification of photocatalysts by silver addition or deposition can be used to increase photocatalytic efficiencies by preventing photogenerated electron–hole recombination through electron trapping mechanisms, and by increasing visible light absorption of the composite materials through the surface plasmon resonance enhancement of silver nanoparticles. Nanosilver also possesses excellent antimicrobial activity, and can be used as a biocidal agent when incorporated into TiO<sub>2</sub> photocatalysts. Alternatively, the host photocatalyst may also contribute to antimicrobial activity observed in the absence of irradiation, such as for AgX (X = Cl, Br, I) and ZnO. These silver-modified composites present a novel class of hybrid photocatalysts, which possess antibacterial and/or antiviral action in both dark and light conditions, and are discussed in detail in this review. In addition, other antimicrobial photocatalysts such as those based on copper are examined. Further work should be performed on these materials to distinguish the roles of acting mechanisms in the light-induced disinfection processes.

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## Contents

1. Introduction .....	63
1.1. Photocatalytic disinfection .....	63
1.2. Challenges in TiO <sub>2</sub> photocatalysis .....	63
1.3. Silver-modified photocatalysts .....	64
1.4. Silver-based disinfection .....	64
1.5. Silver-modified photocatalysts as synergistic disinfection agents .....	64
2. Silver-TiO <sub>2</sub> .....	64
2.1. Overview of photocatalytic enhancement .....	64
2.2. Mechanisms of photocatalytic enhancement .....	64
2.3. Photocatalytic disinfection .....	65
2.4. Silver ion release behaviour .....	65
2.5. Role of silver under irradiation .....	65
2.6. Changes to bacterial adhesion properties .....	67
3. Silver–silver halides (Ag/AgX) .....	68
3.1. Overview of Ag/AgX photocatalysts .....	68
3.2. Mechanism of photocatalytic enhancement .....	68
3.3. Photocatalytic disinfection .....	68
3.4. Ag/AgX as bactericidal and photocatalytic materials .....	68
3.5. Silver ion release behaviour from antibacterial AgX .....	68
3.6. Silver ion release behaviour from antibacterial and photocatalytic Ag/AgX in dark and light conditions .....	69
4. Silver-ZnO .....	69
4.1. Overview of Ag-ZnO photocatalysts .....	69
4.2. Bactericidal activity of ZnO materials .....	69
4.3. Photocatalytic disinfection .....	70

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4.4.	Silver ion release behaviour of antibacterial and photocatalytic Ag-ZnO materials .....	70
4.5.	Changes to bacterial adhesion properties .....	70
5.	Other antimicrobial photocatalysts .....	70
5.1.	Copper-based materials .....	70
5.1.1.	Copper-based disinfection .....	70
5.1.2.	Cupreous biocidal photocatalysts .....	70
5.2.	Other antimicrobial photocatalysts .....	71
6.	Conclusions .....	71
	Acknowledgement .....	72
	References .....	72



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## 1. Introduction

### 1.1. Photocatalytic disinfection

Disinfection plays an important role in the control of pathogens and microbial species in water, and can prevent waterborne epidemics and the spread of infectious disease. Adequate sterilization is also crucial to ensure the safety of medical instruments, food production processes, and environments such as health care facilities. Issues with conventional disinfectants such as chlorine, chloramines, and ozone have been recently identified, since these chemicals may be linked to the formation of harmful disinfection byproducts [1]. Additionally, highly resistant pathogens such as *Cryptosporidium* and *Giardia* cannot be effectively inactivated at normal dosages used for water treatment applications. Alternate disinfection methods using UV-induced processes have also been found to suffer from a lack of residual effect, highlighting the need for further development of disinfection techniques to address these shortcomings.

Since the discovery of the photocatalytic water splitting effect of titanium dioxide by Fujishima and Honda in 1972 [2], research in photocatalysis has been carried out to exploit this process for use in a wide variety of applications, including: hydrogen generation by solar water splitting [3], environmental remediation and purification of contaminated air, water, and soil [4,5], self-cleaning applications [6], and photocatalysis-assisted organic chemical synthesis [7], among others. Matsunaga et al. first investigated the inactivation of microbial cells in water using photochemical sterilization, and they found the complete inactivation of *Lactobacillus acidophilus*, *Saccharomyces cerevisiae*, and *Escherichia coli* could be

achieved using Pt-TiO<sub>2</sub> under irradiation [8]. Photocatalytic disinfection has since been investigated for a number of applications in the contexts of indoor air and environmental health, biological and medical applications, laboratory and hospital applications, pharmaceutical and food production, plant protection applications, wastewater and effluents treatment, and potable water production, as reviewed by Gamage and Zhang [9].

### 1.2. Challenges in TiO<sub>2</sub> photocatalysis

TiO<sub>2</sub> is the most widely used photocatalyst due to its availability, effectiveness, and low cost. TiO<sub>2</sub> can absorb electromagnetic radiation in the ultraviolet (UV) range, causing the photoexcitation of electrons in its valence band to be promoted to its conduction band, creating an electron-hole pair. This electron-hole pair can then undergo further reactions with dissolved oxygen and water to form reactive radical species. The process is often represented schematically for the degradation of a pollutant (ex. organic) according to Fig. 1. The generation of superoxide anions at the cathodic sites and of hydroxyl radicals at the anodic sites can also lead to the production of other reactive species such as hydrogen peroxide. Interaction of these photocatalysis-produced reactive oxygen species (ROS) with biological microorganisms can induce inactivation and cell death, for example, through reaction with functional components in the microbial cell envelope [10].

A main issue arising in photocatalysis lies in the inability of TiO<sub>2</sub> to efficiently use solar light, which is composed of only 3–5% UV. However, solar irradiation consists of approximately 43% visible light, so more efficient utilization of this portion is desirable. Efforts to address this issue by increasing visible light absorption have been made through a number of catalyst modifications such as impurity doping [11–13], metals deposition [14–16], and sensitization [17,18]. In addition, the rate of recombination of the photoexcited electrons and holes is a major factor limiting the efficiency of photocatalytic processes [4,19], and as such research in photocatalyst

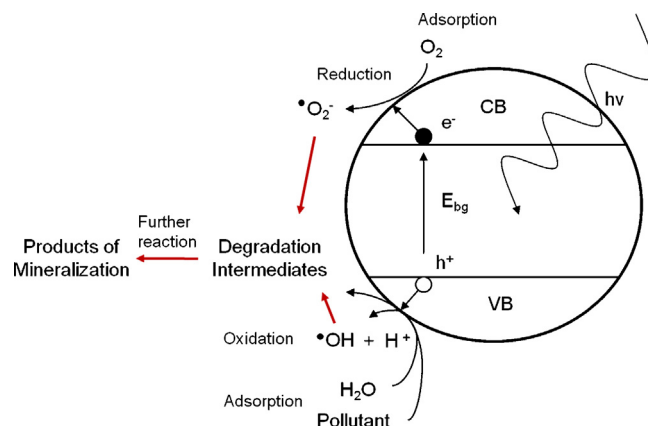


Fig. 1. Photocatalytic degradation by a semiconductor photocatalyst.

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