

Short communication

## Synthesis of biofunctionalized silica nanospheres to separate GST-tagged proteins



Xueyan Zou<sup>a</sup>, Liangliang Li<sup>b</sup>, Haitao Lu<sup>a,c</sup>, Yutao Zhang<sup>a,c</sup>, Yanbao Zhao<sup>a</sup>, Yu Zhang<sup>b</sup>,  
Quanhui Guo<sup>d,\*</sup>

<sup>a</sup> Engineering Research Center for Nanomaterials, Henan University, Kaifeng, 475004, PR China

<sup>b</sup> Key Laboratory of Plant Stress Biology, Henan University, Kaifeng, 475004, PR China

<sup>c</sup> Henan University Minsheng College, Kaifeng, 475004, PR China

<sup>d</sup> College of Chemistry and Chemical Engineering, Henan University, Kaifeng, 475004, PR China

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### ABSTRACT

A Thiol-functionalized silica hollow nanospheres (denoted as SiO<sub>2</sub>-SH NSs) were prepared through a hydrothermal route. The SiO<sub>2</sub>-SH NSs were conjugated with glutathione group (denoted as-GSH) to afford SiO<sub>2</sub>-GSH NSs. The as-prepared SiO<sub>2</sub>-GSH sample has hollow structure and exhibits an average diameter of about 45 nm and a wall thickness of 10 nm. The SiO<sub>2</sub>-GSH NSs were used to separate three kinds of GST-tagged proteins (GST-tagged GPX, GST-tagged LOV and GST-tagged 210-6P). These SiO<sub>2</sub>-GSH NSs exhibit negligible non-specific adsorption, high binding capacity (89.9 μmol/g), the low detection limit (1.0 × 10<sup>-6</sup> mol/L) and reuse property, showing great potentiality in purifying GST-tagged proteins.

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## 1. Introduction

Glutathione S-transferase (denoted as GST) represents a major group of detoxification isoenzymes which participate in a wide range of processes including xenobiotic biotransformation, drug metabolism, degradation of aromatic amino acids and so on [1]. In the meantime, GST can be used as a marker of early phase tumors such as brain tumor, gangliogliomas, prostate cancer and so on [2–4]. This could partly account for the significance of the separation and purification of GST and GST-tagged proteins in biological medicine and disease detection. Recently many materials, such as microfiber, magnetic microparticles and gels, have been employed to separate and purify GST and GST-tagged fusion proteins [5,6]. These materials, although adaptable to many protein expression systems, have some limitations, such as the need for pretreatment to remove cell debris and colloid contaminants, a relatively long operation time, and poor protein solubility. Today, nanoparticles or nanorods as promising nano-devices for the separation of target proteins could be advantageous over conventional counterparts [7–10]; however, surface group density of solid nano-adsorbents was low due to the

multi-step surface modification reaction. In the present research, we focus on the preparation of functionalized silica hollow NSs to enhance the surface group density. As-synthesized silica NSs can directly enrich and separate GST-tagged proteins from the *E. coli* cell lysate, showing great potential as novel adsorbents.

## 2. Experimental

### 2.1. Materials

3-Mercaptopropyltrimethoxysilane (MPS) was purchased from Alfa-Aesar, America. Tetraethyl orthosilicate (TEOS) was from Tianjin Fuchen Chemicals; Hexadecyltrimethylammonium Bromide (CTAB) was purchased from Sinopharm Chemicals; Glutathione (GSH) was purchased from Amresco Company. Triethanolamine (TEA) was purchased from Tianjin Kermel Chemicals. Ethylene diamine tetraacetic acid (denoted as EDTA), absolute alcohol and aqueous ammonia (28 wt%) were purchased from Tianjin Kermel Chemical Reagent Company, China. Phosphate buffer saline (abridged as PBS; concentration 0.1 mol/L, pH 8.0; concentration 0.01 mol/L, pH 7.4) and 5,5'-dithiobis-(2-nitrobenzoic acid) (abridged as DTNB, purity ≥98%) were purchased from Sigma-Aldrich, America. All the reagents are of analytical grade and used as-received.

\* Corresponding author.

E-mail address: [1528540836@qq.com](mailto:1528540836@qq.com) (Q. Guo).

## 2.2. Preparation of SiO<sub>2</sub>-GSH

Briefly, 0.03 g CTAB, 2.8 mL of ethanol and 1.1 mL of TEA were dissolved in 17 mL of H<sub>2</sub>O to form a transparent solution. Then mixture of TEOS and MPS (different volume ratios) was slowly dropped into above solution at 60 °C. After the resultant solution heated at 60 °C for 3 h, the solution was transferred into a Teflon-lined stainless-steel autoclave, sealed and heated at 110 °C for 24 h. Upon completion of the reaction, the solution was cooled, centrifuged, washed and dried at 60 °C in an oven to afford SiO<sub>2</sub>-SH product.

0.3 g of SiO<sub>2</sub>-SH product was dispersed in 10 mL 80 mg/mL of GSH solution. The dispersion was oscillated at 37 °C for 24 h (60 rev/min), followed by centrifuging to afford desired SiO<sub>2</sub>-GSH product. Subsequently, the obtained SiO<sub>2</sub>-GSH product was fully washed with PBS solution to remove physically adsorbed GSH, followed by dispersion in ethanol (25%, v/v) and storing at 4 °C.

## 2.3. Separation and detection of GST-tagged proteins

The to-be-tested mixed proteins were collected from the cell lysate of *Escherichia coli*, which is by water lysis. (concentration 0.01 mol/L, pH 7.4). 1 mL 0.02 g/mL of SiO<sub>2</sub>-GSH sample fully washed with PBS solution was directly introduced into 1000 μL of the cell lysate and shaken at a rotary speed of 90 rev/min at 4 °C for 1 h to capture GST-tagged proteins. Upon completion of capturing GST-tagged proteins, the SiO<sub>2</sub>-GSH sample with captured GST-tagged proteins was isolated by centrifugation and fully washed with PBS solution to remove residual uncaptured proteins, followed by washing with 300 μL of 50 mmol/L GSH solution to

disassociate the GST-tagged proteins from the surface. Then the separated GST-tagged proteins were detected with SDS-PAGE, with the pre-concentration voltage of 70 V and the separation voltage of 120 V. The binding proteins concentration was analyzed at 280 nm and 400 nm by UV-vis spectrophotometer.

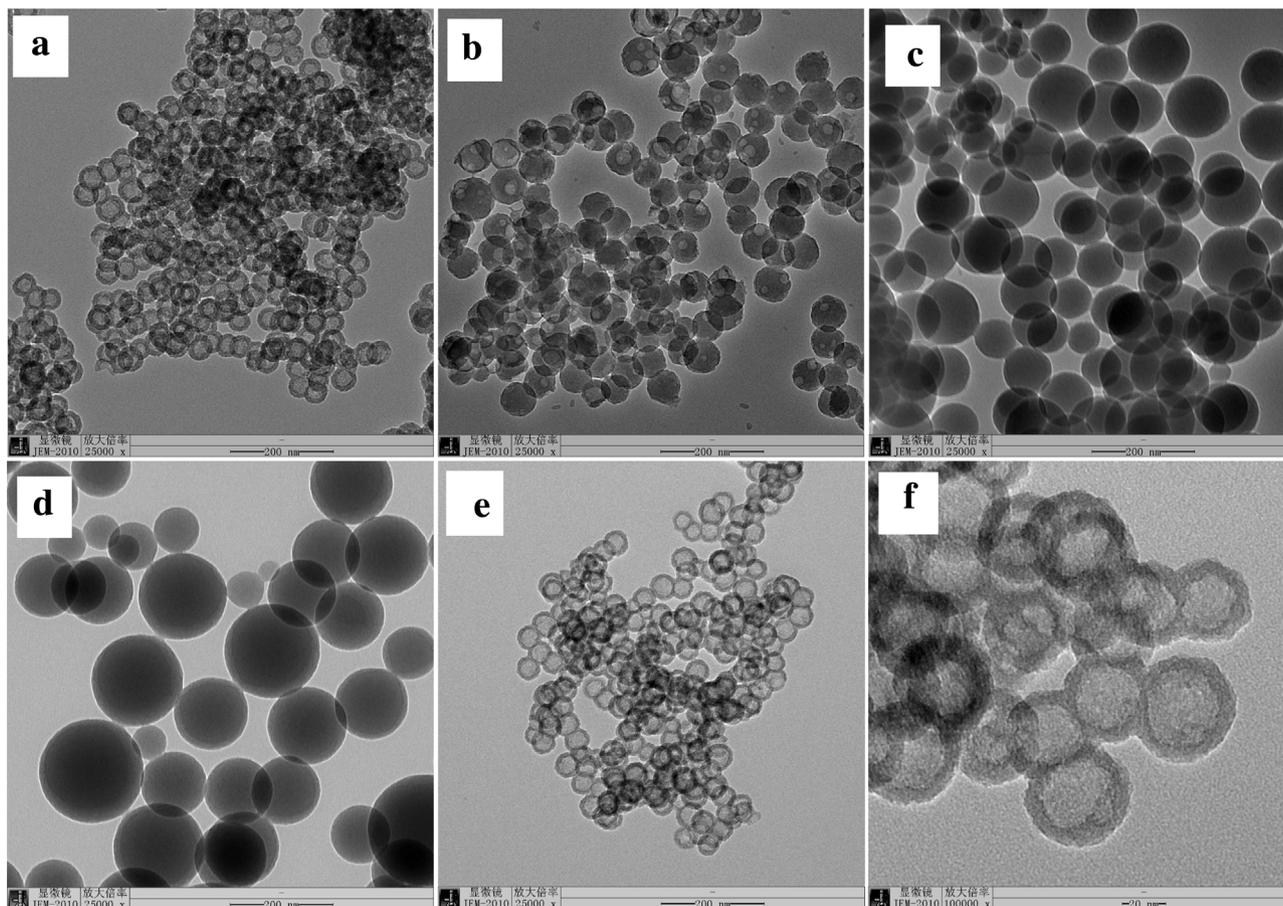
## 2.4. Characterization

The morphology and composition were characterized by transmission electron microscopy (TEM, JEM-2010, Japan), Fourier transform infrared (FT-IR, AVATAR360, America) and thermogravimetric analysis (TG, EXSTAR 6000, Japan), respectively. The surface area was measured by the Brunauer-Emmett-Teller (BET) method (QUADRASORB, American). The separated GST-tagged proteins were detected with SDS-PAGE (Power PAC 300, China). The content of -SH group of the prepared nanospheres and the binding proteins concentration was analyzed by UV-vis spectrophotometer (nanodrop 2000c, America).

## 3. Results and discussion

### 3.1. TEM images of SiO<sub>2</sub>-SH NSs<

A series of SiO<sub>2</sub>-SH samples were prepared under different TEOS/MPS volume ratios so as to investigate the effects of reaction condition on the morphology of the SiO<sub>2</sub>-SH products. As shown in Table 1 and Fig. 1, SiO<sub>2</sub>-SH sample obtained at a TEOS/MPS volume ratio of 10:1 (TEOS 1.4 mL, MPS 0.14 mL) exhibits hollow spherical shape as well as an average diameter of about 45 nm and a wall



**Fig. 1.** TEM images of SiO<sub>2</sub>-SH and SiO<sub>2</sub>-GSH product obtained under different reaction conditions: a- SiO<sub>2</sub>-SH at TEOS/MPS volume ratio 10:1, b- SiO<sub>2</sub>-SH at TEOS/MPS volume ratio 1:1, c- SiO<sub>2</sub>-SH at TEOS/MPS volume ratio 1:10, d- SiO<sub>2</sub>-SH with 1.4 mL of MPS alone, e,f- the SiO<sub>2</sub>-GSH NSs prepared from the SiO<sub>2</sub>-SH obtained at a TEOS/MPS volume ratio of 10:1.

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