



## Contents

## Articles

### Effect of support on $V_2O_5$ catalytic activity in chlorobenzene oxidation

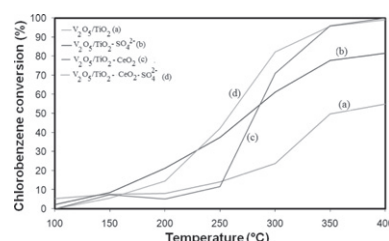
Chiraz Gannoun<sup>a</sup>, Romain Delaigle<sup>b</sup>, Damien P. Debecker<sup>b</sup>, Pierre Eloy<sup>b</sup>, Abdelhamid Ghorbel<sup>a</sup>, Eric M. Gaigneaux<sup>b</sup>

<sup>a</sup>Laboratoire de Chimie des Matériaux et Catalyse, Département de Chimie, Faculté des Sciences de Tunis, Campus universitaire, 2092 El Manar Tunis, Tunisia

<sup>b</sup>Université catholique de Louvain, Institute of Condensed Matter and Nanosciences (IMCN), Division «Solids, Molecules and Reactivity (MOST)», Croix du Sud 2/17, B-1348 Louvain-la-Neuve, Belgium

►  $TiO_2$ ,  $TiO_2-SO_4^{2-}$ ,  $TiO_2-CeO_2$  and  $TiO_2-SO_4^{2-}-CeO_2$  are prepared by sol gel route. ► Active vanadia is deposited on these aerogel supports by impregnation route. ► Vanadia based catalysts are tested in chlorobenzene oxidation. ► Cerium containing catalysts considerably improved catalytic properties at 400 °C. ► Sulfate containing samples improved catalytic properties in the range 200–300 °C.

Applied Catalysis A: General, 447–448 (2012) 1



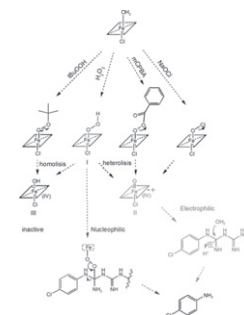
### Metalloporphyrins as cytochrome P450 models for chlorhexidine metabolite prediction

Vinicius Palaretti, Joicy Santamalvina dos Santos, Débora Fernandes Costa Guedes, Luiz Alberto Beraldo de Moraes, Marilda das Dores Assis

Departamento de Química, Faculdade de Filosofia, Ciências e Letras de Ribeirão Preto, Universidade de São Paulo, Av. Bandeirantes 3900, 14040-901 Ribeirão Preto, SP, Brazil

► Chlorhexidine (CHX) is prone to degradation in the presence of metalloporphyrins. ► *p*-Chloroaniline (pCA) is the main product from CHX catalytic degradation. ► The results show the strong possibility of pCA formation when CHX is ingested.

Applied Catalysis A: General, 447–448 (2012) 7



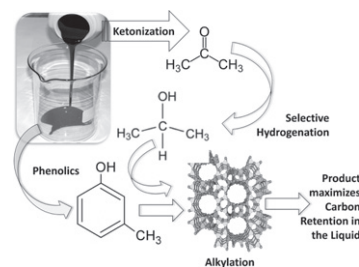
### Improving carbon retention in biomass conversion by alkylation of phenolics with small oxygenates

Lei Nie, Daniel E. Resasco

Center for Biomass Refining, School of Chemical, Biological and Materials Engineering, University of Oklahoma, Norman, OK 73019, USA

► Alkylation is an efficient method to retain C from small oxygenates in liquid fraction of bio-oil. ► *m*-Cresol alkylation activity follows the sequence 2-propanol > propylene > 1-propanol. ► A hydrogenation/alkylation two-stage process is proposed to utilize small aldehydes and ketones. ► The combination Pt–Fe/SiO<sub>2</sub> + H-beta was found to be effective for the two-stage process.

Applied Catalysis A: General, 447–448 (2012) 14



## Alloyed Ni-Fe nanoparticles as catalysts for NH<sub>3</sub> decomposition

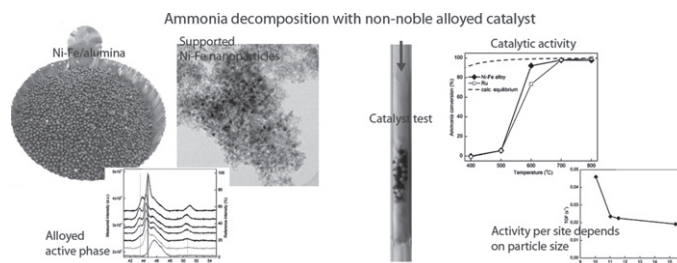
Applied Catalysis A: General, 447–448 (2012) 22

Søren Bredmose Simonsen<sup>a</sup>, Debasish Chakraborty<sup>b</sup>, Ib Chorkendorff<sup>a</sup>, Søren Dahl<sup>a</sup>

<sup>a</sup>CINF, Department of Physics, Technical University of Denmark, Fysikvej, Building 307, DK-2800 Lyngby, Denmark

<sup>b</sup>Amminex A/S, Gladsaxevej 363, DK-2860 Søborg, Denmark

► Non-noble metal Ni-Fe alloy catalysts for NH<sub>3</sub> decomposition. ► Promising catalysts for high temperature NH<sub>3</sub> decomposition. ► Structural sensitivity: increasing activity for decreasing particle sizes. ► Of various support oxides: Al<sub>2</sub>O<sub>3</sub>-based support materials give the best performance.



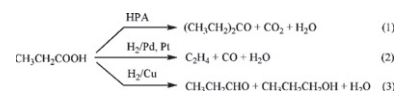
## Deoxygenation of propionic acid on heteropoly acid and bifunctional metal-loaded heteropoly acid catalysts: Reaction pathways and turnover rates

Applied Catalysis A: General, 447–448 (2012) 32

Mshari A. Alotaibi, Elena F. Kozhevnikova, Ivan V. Kozhevnikov

Department of Chemistry, University of Liverpool, Liverpool L69 7ZD, UK

► Propionic acid deoxygenation studied at 250–400 °C in H<sub>2</sub> or N<sub>2</sub>. ► HPA catalysed ketonisation to 3-pentanone at 250–300 °C in N<sub>2</sub>. ► Pd and Pt decarbonylated propionic acid to ethene in H<sub>2</sub> but had no effect in N<sub>2</sub>. ► C=O bond to yield propanal and 1-propanol. ► The turnover rate of propionic acid conversion follows the order: Pd > Pt > Cu.



## A comparative study of bulk and 3DOM-structured Co<sub>3</sub>O<sub>4</sub>, Eu<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3</sub>, and Co<sub>3</sub>O<sub>4</sub>/Eu<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3</sub>: Preparation, characterization, and catalytic activities for toluene combustion

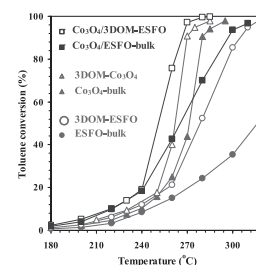
Applied Catalysis A: General, 447–448 (2012) 41

Kemeng Ji<sup>a</sup>, Hongxing Dai<sup>a</sup>, Jiguang Deng<sup>a</sup>, Xinwei Li<sup>a</sup>, Yuan Wang<sup>a</sup>, Baozu Gao<sup>a</sup>, Guangmei Bai<sup>a</sup>, Chak Tong Au<sup>b</sup>

<sup>a</sup>Laboratory of Catalysis Chemistry and Nanoscience, Department of Chemistry and Chemical Engineering, College of Environmental and Energy Engineering, Beijing University of Technology, Beijing 100124, PR China

<sup>b</sup>Department of Chemistry and Center for Surface Analysis and Research, Hong Kong Baptist University, Kowloon Tong, Kowloon, Hong Kong, China

► 3DOM-structure and bulk Co<sub>3</sub>O<sub>4</sub>, Eu<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3</sub>, and Co<sub>3</sub>O<sub>4</sub>/Eu<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3</sub> are prepared. ► Porous samples are higher than bulk samples in O<sub>ads</sub> content and better reducibility. ► The porous samples exhibit higher catalytic performance than the bulk samples. ► There is a synergistic effect between Co<sub>3</sub>O<sub>4</sub> and Eu<sub>0.6</sub>Sr<sub>0.4</sub>FeO<sub>3</sub>. ► Catalytic activity is governed by O<sub>ads</sub> content, reducibility, and synergistic action.



## New palladium catalysts on polyketone prepared through different smart methodologies and their use in the hydrogenation of cinnamaldehyde

Applied Catalysis A: General, 447–448 (2012) 49

Anna Maria Raspolli Galletti<sup>a</sup>, Luigi Toniolo<sup>b</sup>, Claudia Antonetti<sup>a</sup>, Claudio Evangelisti<sup>c</sup>, Claudia Forte<sup>d</sup>

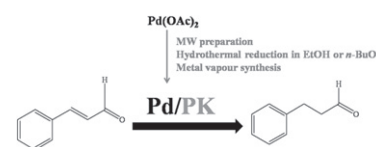
<sup>a</sup>Department of Chemistry and Industrial Chemistry, University of Pisa, Via Risorgimento 36, 56126 Pisa, Italy

<sup>b</sup>Department of Molecular Sciences and Nanosystems, University Ca' Foscari of Venice, Dorsoduro 2137, 30123 Venice, Italy

<sup>c</sup>ICCOM CNR, Via G. Moruzzi 1, 56124 Pisa, Italy

<sup>d</sup>CNR ISTM, Via Fantoli 16, 20138 Milano,

► Polyketone has been employed for the first time as support for Pd nanoparticles. ► Three different synthetic approaches were adopted. ► The hydrogenation of cinnamaldehyde to hydrocinnamaldehyde was studied. ► The catalysts showed high stability and recyclability. ► The performances are related to Pd nanoparticles morphology.



Download English Version:

<https://daneshyari.com/en/article/40611>

Download Persian Version:

<https://daneshyari.com/article/40611>

[Daneshyari.com](https://daneshyari.com)