

## A QSAR study of acute toxicity of *N*-substituted fluoroacetamides to rats

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### Abstract

Acute toxicity *in vivo* toward rats, of nineteen *N*-alkyl and *N*-cycloalkyl fluoracetamides [F–CH<sub>2</sub>–C(O)–NH–R] was correlated with their structure-dependent properties. Used descriptors are: molecular weights ( $M_w$ ) and heat of formation ( $\Delta H_f$ ) of compounds; molar refractivity (CMR), lipophilicity (Clog *P*), Broto lipol values, virtual log *P*, molecular lipophilic potential (MLP), Van der Waals surfaces (VdW SAS) and hydropathicity surface (ILM) of whole molecules; Taft steric parameters ( $E_s$ );  $E_s$  values with Hancock corrections ( $E_s^{CH}$ ) and Verloop sterimol ( $B_5$ ) and ( $L$ ) parameters of alkyl and cycloalkyl residues; superdelocalizabilities and electron densities on the [NH–C(O)–CH<sub>2</sub>–F] fragment. Strong quantitative structure–activity relationships were assessed. Obtained correlation suggested that lipophilicity, shape and bulkiness of the alkyl and cycloalkyl substituents, particular nearest vicinity of the amide nitrogen, as well charges on the amide moiety are the main factors that influence on the acute toxicity of studied compounds toward rats. Mechanism of toxic action was proposed.

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### 1. Introduction

One of the current interests in medicinal chemistry and toxicology is the classification of chemical substances with the respect to their toxicity toward living systems. Quantitative structure–activity relationships

(QSAR) have provided a valuable tool in research on the toxicity of organic chemicals.

The toxicity of derivatives of fluoroacetic acid to insects and rodents is well known (Metcalf, 1966; Zhu et al., 2002). Fluoroacetamide is an active insecticide, but it is less toxic and acts more slowly than sodium fluoroacetate (Alekseev and Turov, 1967). In addition, various *N*-substituted and *N,N*-disubstituted fluoroacetamides (Takeuchi and Ishida, 1962) and *N*-methylenefluoroacetamide derivatives (Pianka and Polton, 1965) have been

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tested as insecticides and rodenticides. Related compounds have been also studied (Ishii, 1976).

*N*-alkyl fluoroacetamides also exert antischistosomal activity (Chen et al., 1982a,b). It was shown that *N*-ethyl fluoroacetamide inhibit the aconitase (E.C. 4.2.1.3) from *Schistosoma japonicum* and exerts antischistosomiasis activity (Huang et al., 1980).

It is also known that *N*-alkyl haloacetamides act as alkylating agents (Kanstrup et al., 1993; Jablonkai, 2003). Structure–activity relationships of fifteen *N*-alkyl bromoacetamides in their action toward *S. aureus* were described previously (Hansch and Lien, 1971). Minimum bactericidal concentration (MBC) was correlated with lipophilic ( $\log P$ ), steric  $E_s$  and electronic  $\sigma'$  values. Very good correlation was obtained ( $r = 0.980$ ).

The aim of this work was to correlate the acute toxicity in vivo toward rats, of group of nineteen *N*-alkyl and *N*-cycloalkyl fluoracetamides with their structure related properties. The toxicity results will also complement the toxicity database for the risk assessments of the studied compounds.

## 2. Materials and methods

### 2.1. Chemistry

Nineteen *N*-alkyl and *N*-cycloalkyl fluoroacetamides (listed in Table 1) were synthesized, using the known

Schotten–Baumann reaction by acylation of the corresponding amines with fluoroacetyl chloride in the presence of a concentrated aqueous solution of potassium hydroxide, purified by recrystallization/microdistillation and characterized by melting point, elemental analysis,  $^1\text{H}$  NMR,  $^{13}\text{C}$  NMR spectroscopy and mass spectrometry (Mišćević et al., 1992; Jeremić et al., 1995).

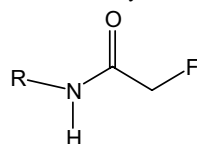
### 2.2. Animals

The male adult Wistar rats, average mass 200–250 g, were used. Animals were kept in cages (10 rats per cage) at room temperature under a 12-h light/dark cycle with food and water available ad libitum.

### 2.3. Acute toxicity

The acute toxicity ( $\text{LD}_{50}$ ) was evaluated as described by Miller and Tainter (1994). In brief, freshly prepared aqueous solutions of *N*-substituted fluoroacetamides were used. The method involved the administration of five different doses of the aqueous solutions to five groups of rats (six rats per group). The mortality in each group was recorded in 24 h. The  $\text{LD}_{50}$  was than estimated and the obtained results are listed in Table 1.

Table 1  
Acute toxicity of *N*-alkyl and *N*-cycloalkyl fluoroacetamides



Compound no.	R–	C (mg/kg)	C (M/kg)	$\log(1/C)$
1	<i>n</i> -Propyl–	7	$5.88 \times 10^{-05}$	4.2310
2	<i>n</i> -Butyl–	6	$4.51 \times 10^{-05}$	4.3462
3	<i>n</i> -Pentyl–	8	$5.44 \times 10^{-05}$	4.2648
4	<i>n</i> -hexyl–	7	$4.34 \times 10^{-05}$	4.3623
5	(1-Methyl)ethyl–	72	$6.04 \times 10^{-04}$	3.2187
6	(1-Methyl)propyl–	104	$7.81 \times 10^{-04}$	3.1073
7	(1,2,2-Trimethyl)propyl–	150	$9.30 \times 10^{-04}$	3.0313
8	(1-Methyl)butyl–	118	$8.02 \times 10^{-04}$	3.0960
9	(1,4-Dimethyl)pentyl–	250	$1.43 \times 10^{-04}$	2.8457
10	(1,1-Dimethyl)ethyl–	132	$9.91 \times 10^{-04}$	3.0038
11	(2-Methyl)propyl–	29	$2.18 \times 10^{-04}$	3.6620
12	(2,2-Dimethyl)propyl–	70	$4.76 \times 10^{-04}$	3.3228
13	(3-Methyl)butyl–	13	$8.83 \times 10^{-05}$	4.0539
14	(1,1,3,3-Tetramethyl)butyl–	300	$1.59 \times 10^{-03}$	2.8000
15	Cyclopropyl–	9	$7.68 \times 10^{-05}$	4.1144
16	Cyclobutyl–	10	$7.62 \times 10^{-05}$	4.1178
17	Cyclopentyl–	31	$2.14 \times 10^{-04}$	3.6705
18	Cyclohexyl–	130	$8.17 \times 10^{-04}$	3.0880
19	Cycloheptyl–	200	$1.15 \times 10^{-04}$	2.9376

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