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## Technical note

# Oxidation characteristics of airborne carbon nanoparticles by NO<sub>2</sub>

Jungbum Choo<sup>a</sup>, Jae Hee Jung<sup>a</sup>, Woojin Kim<sup>a</sup>, Hyuncheol Oh<sup>a</sup>,  
Jinho Kim<sup>a</sup>, Hakjoon Kim<sup>b</sup>, Yong Jin Kim<sup>b</sup>, Sangsoo Kim<sup>a,\*</sup>

<sup>a</sup>School of Mechanical, Aerospace & Systems Engineering, Korea Advanced Institute of Science and Technology, 335 Gwahangno, Yuseong-gu, Daejeon, 305-701, Republic of Korea

<sup>b</sup>Eco-Machinery Research Division, Korea Institute of Machinery & Materials, 171 Jang-dong, Yuseong-gu, Daejeon, 305-343, Republic of Korea

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## ABSTRACT

The oxidation characteristics of airborne carbon nanoparticles were investigated at various temperatures and NO<sub>2</sub> concentrations. Airborne carbon nanoparticles were generated by spark-discharging method using nitrogen as a carrier gas. Monodisperse carbon particles classified using a differential mobility analyzer were introduced into a tube furnace with NO<sub>2</sub> for oxidation reaction. The size distributions of oxidized carbon aerosol particles were measured using a scanning mobility particle spectrometer system which consisted of a differential mobility analyzer and a condensation particle counter. The result was that as NO<sub>2</sub> concentrations and reaction temperatures increased, the surface oxidation rate of carbon aerosol particles increased. For NO<sub>2</sub> gas, the activation energy of the oxidation reaction was approximately 76.3 kJ/mol. The activation energy of the oxidation reaction for the mixture of NO<sub>2</sub> and O<sub>2</sub> gases was 65.9 kJ/mol, which is smaller than that for only NO<sub>2</sub> gas.

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## 1. Introduction

For over a century, diesel engines have proved reliable, economical and durable source of power for many applications. Currently, diesel vehicles are undergoing further development due to their higher fuel efficiency and lower CO<sub>2</sub> emissions as compared to gasoline vehicles because regulations concerning diesel vehicles are becoming more severe due to the harmful effects of diesel exhaust on human health and on the environment (Utell and Frampton, 2000; EPA600/8-90/057F, 2002). The particulate matter (PM) from diesel vehicles consists mostly of carbonaceous particle and a volatile organic fraction (VOF). The PM is often formed during combustion process in

locally fuel-rich regions and exhausted in the form of solid agglomerates condensed with VOC (Walker, 2004).

For diesel engine, the lowering of PM emission is generally based on a filtration system in which trapped PM is periodically removed by means of controlled oxidation reaction with pure O<sub>2</sub> or air. Accordingly, many researchers have investigated the oxidation characteristics of PM exhausted from the diesel engine. Higgins et al. (2002, 2003) investigated the oxidation characteristics of size-selected soot nanoparticles in air with the temperature range of 800–1140 °C. However, according to recent research, NO<sub>2</sub> has been found to be a more efficient oxidant than O<sub>2</sub> at lower temperatures in the range of 200–500 °C (Dorai et al., 2000; Ehrburger et al., 2002; Jeguirim et al., 2005; Kamm et al., 2004).

\* Corresponding author. Tel.: +82 42 350 2009; fax: +82 42 350 2204.  
E-mail address: [sskim@kaist.ac.kr](mailto:sskim@kaist.ac.kr) (S. Kim).

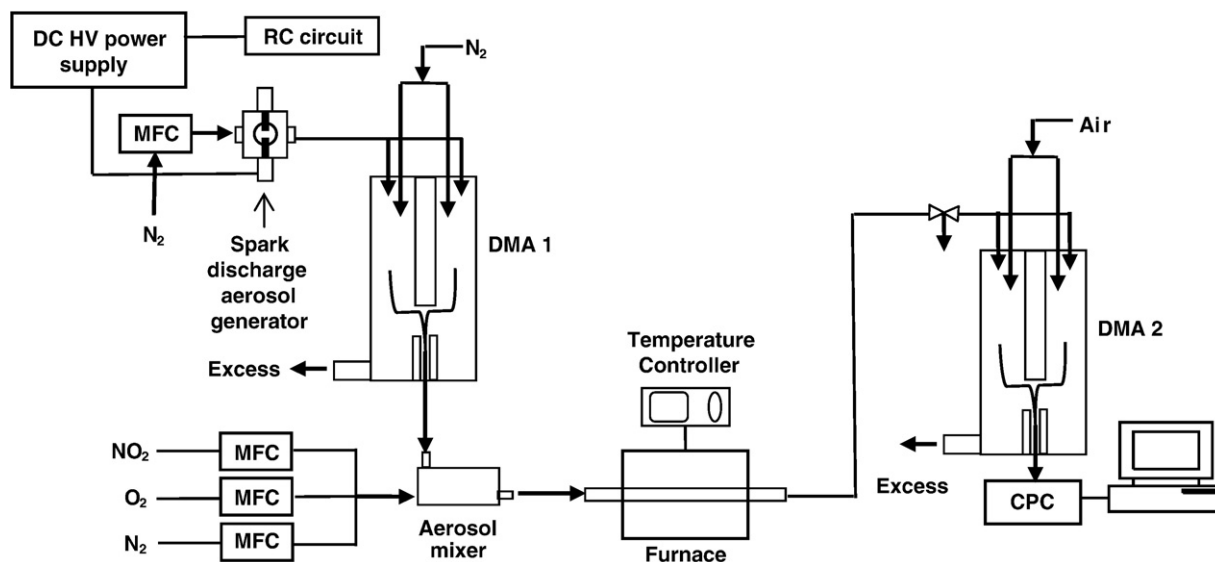


Fig. 1 – Schematic diagram of experimental setup.

Small amounts of NO<sub>2</sub> in the range of a few hundred ppm by volume can promote the continuous oxidation of carbon particulates. In most regenerative diesel particulate filters (DPF) using NO<sub>2</sub>, NO<sub>2</sub> is produced by catalytic oxidation reactions or non-thermal plasma reactions of NO, which is previously present in the engine exhaust stream (Dorai et al., 2000; Ehrburger et al., 2002). Jeguirim et al. (2005) studied the oxidation of carbon black deposited on a quartz plate by NO<sub>2</sub> regarding the mass change of carbon black and the concentration of generated CO<sub>2</sub> and CO. Kamm et al. (2004) investigated the gasification of soot deposited on a quartz wool filter by O<sub>3</sub> and NO<sub>2</sub>.

Most investigations of carbon oxidation by NO<sub>2</sub> have been on the carbon deposited on filters or plates. However, little is known about the oxidation characteristics of the carbon

nanoparticles in aerosol state. In this study, we have focused on the oxidation of airborne carbon nanoparticles which are generated by a spark discharge generator. During the regeneration process of a DPF, the NO<sub>2</sub> can oxidize not only the soot collected on a DPF but also airborne PM emitted from the engine and redispersed from a DPF. This oxidation effect of PM by NO<sub>2</sub> in the airborne state could reduce the total PM emission as a concurring way for mitigation of PM emission.

In our experiment, the oxidation characteristics of nano-sized carbon aerosol particles in NO<sub>2</sub> gas and in the mixture of NO<sub>2</sub> and O<sub>2</sub> gases were investigated at various temperatures. Additionally, the enhancement of carbon oxidation by NO<sub>2</sub> in the presence of O<sub>2</sub> was investigated because in the exhaust stream of diesel engines, a large excess of O<sub>2</sub> is always present compared to NO<sub>2</sub>.

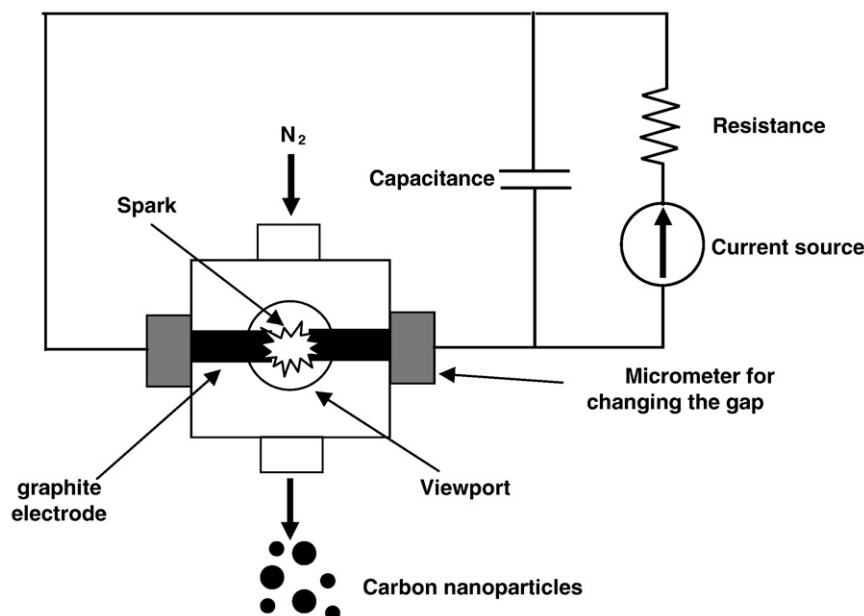


Fig. 2 – Schematic diagram of the spark discharge aerosol generator.

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