



# Development of a new LOPAP instrument for the detection of O<sub>3</sub> in the atmosphere

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## HIGHLIGHTS

- A simple instrument for the detection of O<sub>3</sub> is presented.
- Sensitive, selective and absolute quantification by Lambert–Beer's law.
- Single interferences studied can be negligible for atmospheric conditions.
- Successful validation in the atmosphere and in a complex smog chamber experiment.

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## ABSTRACT

A simple O<sub>3</sub>-LOPAP (Long Path Absorption Photometer) instrument for the selective detection of O<sub>3</sub> in the atmosphere is presented, which is mainly intended to be used as an extension of a recently developed NO<sub>2</sub>-LOPAP. O<sub>3</sub> is sampled in a stripping coil by a selective chemical reaction with the highly absorbing Indigo dye. The reduction of the optical absorption of the dye is detected in a liquid core waveguide. The instrument has a detection limit of 0.4 ppbv, an accuracy of 10%, a precision of 2% for 6 min time resolution and allows the absolute quantification of O<sub>3</sub> according to Lambert–Beer's law. Interferences towards nitrogen oxides (NO and NO<sub>2</sub>), N<sub>2</sub>O<sub>5</sub>, H<sub>2</sub>O<sub>2</sub> and several VOCs were quantified in the laboratory and found to be negligible for atmospheric conditions. The new instrument was successfully validated against a commercial UV-absorption instrument during an urban field campaign and against the FTIR technique in a smog chamber under complex photochemical conditions. For the UV-absorption instrument significant positive interferences towards aromatic species were observed in the smog chamber.

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## 1. Introduction

Due to its direct adverse effects on humans, animals and plants and due to its UV absorbing properties, ozone (O<sub>3</sub>) is of crucial importance for life on Earth. Especially, photochemical O<sub>3</sub> formation in the troposphere during “summer smog” episodes and its destruction in the stratosphere after polar winter (“ozone hole”) have attracted considerable public and scientific attention in the past. Due to its health effects, a threshold limit average daytime value of 120 µg m<sup>−3</sup> (~60 ppbv) not to be exceeded more than 25 days per year was introduced in the European Union in 2010 (Directive, 2002/3/EC, 2002).

The environmental importance of ozone has led to routine measurements in the atmosphere by different techniques which started with primary studies of the chemist Schönbein (1844).

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Today, ozone is typically measured by in-situ instruments using the UV-absorption of ozone at 254 nm. Standard UV-absorption instruments have a typical detection limit of 1–2 ppbv for a time response of <1 min (Dunlea et al., 2006). The simplicity of the method and the absolute quantification of O<sub>3</sub> with Lambert–Beer's law have resulted in this technique being typically recommended, by e.g. DIN, ISO, EPA, EU (ISO 13964, 1998; McElroy et al., 1997). However, it is well known that atmospheric trace gases, which absorb in the UV (e.g. aromatic hydrocarbons), can interfere in the detection of O<sub>3</sub>. For that reason, the instruments use two channels. In one channel, ozone is destroyed, e.g. on heated metal surfaces or manganese dioxide, and only the interfering species are quantified. In the other channel, the sum of ozone and the interferences are measured. From the difference between the two channels, the ozone concentration is determined. While this interference correction typically works well under atmospheric conditions (Kleindienst et al., 1993; Dunlea et al., 2006; Williams et al., 2006), interferences towards sticky aromatic species – also undesirably removed in the interference channel of the instrument – are still

observed under laboratory conditions, when high trace gas levels are used (Kleindienst et al., 1993).

In addition to UV-absorption, chemiluminescence techniques based on gas phase reactions of  $O_3$  with NO or ethene and liquid phase reactions, e.g. with Coumarine, Luminol or Indigo-disulfonate are also used for the detection of ozone in the atmosphere (Stedman et al., 1972; Gregory et al., 1983; Takeuchi et al., 1990; Ridley et al., 1992; Jiménez et al., 1997), even though absolute calibration is not possible for these techniques. However, the sensitivity and time response of these instruments can be very high, with detection limits down to 10 pptv for seconds time response (Ridley et al., 1992). Besides compact in-situ instruments,  $O_3$  is also measured by the path averaging FTIR and DOAS techniques (Dunlea et al., 2006). The detection limits and time responses of FTIR and DOAS instruments are typically a few ppbv and  $\sim 10$  min, respectively, depending on the optical path length, visibility of the atmosphere and the concentration levels (Williams et al., 2006; Dunlea et al., 2006).

Recently, a sensitive and selective  $NO_2$ -LOPAP instrument (Long Path Absorption Photometer) has been developed and successfully validated in our group (Villena et al., 2011, 2012). The only significant interference towards  $O_3$  is suppressed using a scrubber, in which  $O_3$  is quantitatively sampled in a stripping coil by the selective reaction with the blue Indigo dye, while  $NO_2$  is almost unaffected. This reaction was originally used to quantify  $O_3$  in the liquid phase (Bader and Hoigné, 1981) and was later applied in passive samplers (Grosjean and Hisham, 1992; Scherren and

Adema, 1996; Garcia et al., 2010), in a discontinuous bubbler method (Bergshoeff et al., 1984) and in a stopped flow approach (Toda et al., 2003) to measure  $O_3$  in the atmosphere. In the present study, the Indigo method is used for the continuous sensitive quantification of  $O_3$  in the gas phase also by the LOPAP technique. A two channel instrument was developed to characterize and optimize the instrument's performance. However, later it is planned to use only the one channel ozone scrubber from the  $NO_2$ -LOPAP instrument for the sensitive and selective quantification of  $O_3$  and  $NO_2$  in the atmosphere.

## 2. Experimental section

The  $O_3$ -LOPAP instrument consists of two separate units. (1) The external sampling unit, directly situated at the sampling site (i.e. in the atmosphere), which avoids the use of any sampling lines. (2) The detection unit (19" instrument) where the Indigo absorption is quantified using a liquid core wave guide. Both units are connected by a temperature controlled sampling line, through which liquid reagents, zero gas and the liquid from the thermostat are transferred (Fig. 1).

### 2.1. Sampling unit

In the external sampling unit two similar vertical glass stripping coils are connected in series (see Fig. 1). Each of the two stripping coils has 25 coils with 40 mm outer coil diameter, an inner glass

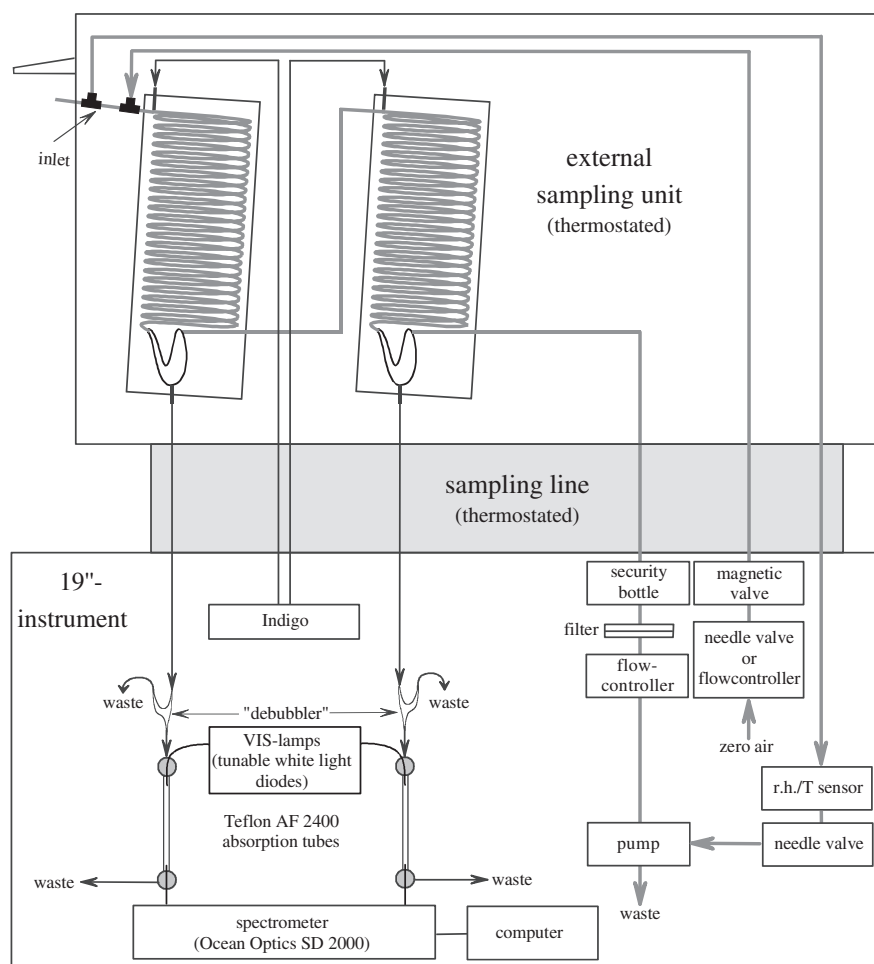


Fig. 1. Schematic setup of the  $O_3$ -LOPAP instrument.

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