

Thermodynamically constrained averaging theory approach for modeling flow and transport phenomena in porous medium systems: 4. Species transport fundamentals

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Abstract

This work is the fourth in a series of papers on the thermodynamically constrained averaging theory (TCAT) approach for modeling flow and transport phenomena in multiscale porous medium systems. The general TCAT framework and the mathematical foundation presented in previous works are built upon by formulating macroscale models for conservation of mass, momentum, and energy, and the balance of entropy for a species in a phase volume, interface, and common curve. In addition, classical irreversible thermodynamic relations for species in entities are averaged from the microscale to the macroscale. Finally, we comment on alternative approaches that can be used to connect species and entity conservation equations to a constrained system entropy inequality, which is a key component of the TCAT approach. The formulations detailed in this work can be built upon to develop models for species transport and reactions in a variety of multiphase systems.

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1. Introduction

This paper is the fourth in a series of efforts intended to yield complete, rigorous, closed models that describe transport phenomena in multiscale porous medium systems using the thermodynamically constrained averaging theory (TCAT) approach. The first paper [18] provides an overview of the general TCAT approach, which is built on averaged conservation and thermodynamic equations that constrain an entropy inequality. The second paper provides the mathematical fundamentals and theorems that are used to generate needed macroscale equations [25]. The third paper illustrates the application of the method for single-fluid-phase, single-species flow in a porous medium [19]. In the present work, we develop additional fundamental

components of the theory to enable the subsequent building of rigorous, closed models for multispecies systems.

The composition of a phase is of central importance for modeling many porous medium systems. Example applications include saltwater intrusion; contaminant fate, transport, and remediation; irrigation and fertilization; aquifer storage and recovery of treated drinking water; and analysis of the effects of nuclear waste disposal. Models used to describe such systems are typically based upon the advective–dispersive equation, assuming a Fickian form of the dispersion process [11]. These models are typically posited directly at the macroscale and are not usually thermodynamically constrained. It is also commonplace for the variables that appear in such macroscale equations to lack precise definitions and connections to microscale quantities.

Of further significance is the general consensus that heterogeneity at the macroscale, typical of most natural systems, leads to the limited usefulness of the advective–dispersive equation for many problems of interest (e.g.

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Nomenclature

Roman letters

b	entropy source density	\mathcal{J}_{Pt}	index set of all microscale points in the REV that describes Ω_i
\mathbf{C}	Greens' deformation tensor	\mathcal{J}_{Pts}	index set of common point entities that include an intersection with the solid phase volume entity
\mathbf{d}	rate of strain tensor	$\mathcal{J}_{\text{Pt/s}}$	index set of common point entities that do not include and intersection the solid phase volume entity
E	internal energy density	\mathcal{J}_s	index set of species
E_T	total energy density	\mathcal{J}_{s_i}	index set of species in entity i
\mathcal{E}	the set of entities	j_s	solid-phase Jacobian
\mathcal{E}_C	the set of common curve entities	K_E	macroscale kinetic energy per unit mass due to microscale velocity fluctuations
\mathcal{E}_I	the set of interface entities	\mathbb{M}	mass
\mathcal{E}_P	the set of phase volume entities	M	variational constant subject to mass conservation constraint
\mathcal{E}_{Pt}	the set of common point entities	\mathcal{M}	conservation of mass equation
\mathcal{E}	conservation of energy equation	${}^{i\kappa \rightarrow i}M$	transfer of mass of species i in the κ entity to the i species in the i entity per unit volume per unit time
\mathbb{E}	internal energy	${}^{i\kappa \rightarrow i}M_{E_i}$	transfer of energy from the κ entity to the i entity due to inter-entity mass transfer of species i per unit volume per unit time
\mathcal{E}_c	connected set of entities	${}^{i\kappa \rightarrow i}\mathbf{M}_{v_i}$	transfer of momentum from the κ entity to the i entity due to inter-entity mass transfer of species i per unit volume per unit time
\mathbf{e}	unit vector tangent to a common curve and oriented positive outward	${}^{i\kappa \rightarrow i}M_{\eta_i}$	transfer of entropy from the κ entity to the i entity due to inter-entity mass transfer of species i per unit volume per unit time
e_{ii}	microscale intra-entity internal energy transfer rate from all other species in entity i to the i species per unit measure of the i entity	\mathbf{n}_i	outward unit normal vector from entity i
e_{Tii}	total microscale energy transferred intra-entity from all other species in entity i to the i species per unit measure of the i entity	\mathcal{P}	conservation of momentum equation
$e_{T_i}^{ii}$	total microscale energy transferred intra-entity from all other species in entity i to the i species per unit measure of the i entity	\mathcal{P}_i	general microscale property
F	thermodynamic functional to be minimized	p	fluid pressure
f	general variable	\mathbf{p}_{ii}	microscale intra-entity momentum transfer rate from all other species in entity i to the i species per unit measure of the entity
\mathbf{f}	general vector variable	$\bar{\mathbf{p}}^{ii}$	macroscale intra-entity momentum transfer rate from all other species in entity i to the i species per unit measure of the entity
\mathbf{f}'	general microscale vector tangent to an interface	${}^{j\kappa \rightarrow i}Q$	transfer of energy from species in the κ entity to the i species in the i entity resulting from heat transfer and deviation from mean processes per unit volume per unit time
\mathbf{f}''	general microscale vector tangent to a common curve	\mathbf{q}	non-advective heat flux density vector
\mathbf{g}	acceleration vector due to an external force, such as gravity	r_{ii}	microscale intra-entity reaction rate resulting in the production of species i in entity i from all other species per unit measure of the entity
h	heat source density	r^{ii}	macroscale intra-entity reaction rate resulting in the production of species i in entity i from all other species per unit measure of the entity
\mathbf{I}	identity tensor	\mathcal{S}	entropy balance equation
\mathbf{I}'	surface identity tensor	\mathbb{S}	entropy
\mathcal{J}	index set of entities	\mathcal{S}	set of all species
\mathcal{J}_C	index set of common curve entities		
\mathcal{J}_{Cs}	index set of common curve entities that include an intersection with the solid phase volume		
$\mathcal{J}_{C/s}$	index set of common curve entities that do not include an intersection with the solid phase volume		
\mathcal{J}_c	index set of connected entities		
\mathcal{J}_f	index set of fluid-phase volumes		
\mathcal{J}_I	index set of interface entities		
\mathcal{J}_{Is}	index set of interface entities that include an intersection with the solid phase volume		
$\mathcal{J}_{I/s}$	index set of interface entities that do not include an intersection with the solid phase volume		
\mathcal{J}_P	index set of phase volume entities		
\mathcal{J}_{Pt}	index set of common point entities		

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