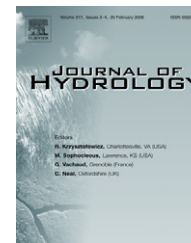




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# Evidence for terrigenous SF<sub>6</sub> in groundwater from basaltic aquifers, Jeju Island, Korea: Implications for groundwater dating

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**Summary** Measurements of the concentrations of dichlorodifluoromethane (CFC-12), tritium (<sup>3</sup>H), and sulfur hexafluoride (SF<sub>6</sub>) in groundwater from basaltic aquifers in Jeju Island, Korea, demonstrate a terrigenous source of SF<sub>6</sub>. Using a lumped-parameter dispersion model, groundwater was identified as young water (<15 years), old water with negligible CFC-12 and <sup>3</sup>H, and binary mixtures of the two. Model calculations using dispersion models and binary mixing based on <sup>3</sup>H and CFC-12 concentrations demonstrate a non-atmospheric excess of SF<sub>6</sub> relative to CFC-12 and <sup>3</sup>H concentrations for more than half of the samples. The non-atmospheric excess SF<sub>6</sub> may have originated from terrigenous sources in relict volcanic fluids, which could have acquired SF<sub>6</sub> from granites and basement rocks of the island during volcanic activity. Local excess anthropogenic sources of SF<sub>6</sub> are unlikely. The SF<sub>6</sub> age is biased young relative to the CFC-12 age, typically up to 20 years and as high as 30 years. This age bias is more pronounced in samples of groundwater older than 15 years. The presence of terrigenous SF<sub>6</sub> can affect the entire dating range for groundwater in mixtures that contain a fraction of old water.

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## Introduction

Sulfur hexafluoride (SF<sub>6</sub>) is a colorless, odorless, incombustible, non-toxic, stable gas with excellent electrical insulating and arc-quenching properties (Maiss and Brenninkmeijer,

1998). SF<sub>6</sub> has been used as a conservative artificial gaseous tracer in many groundwater studies through deliberate introduction of SF<sub>6</sub>-charged water. Upstill-Goddard and Wilkins (1995) used SF<sub>6</sub> to evaluate bulk water flow in an artificial geothermal reservoir. In a coastal karst aquifer with freshwater–seawater interfaces, SF<sub>6</sub> was used as an artificial tracer to determine groundwater transport rates (Dillon et al., 1999; Dillon et al., 2003). Gamlin et al. (2001) used SF<sub>6</sub> to trace surface water from a 9 km reach of a sinking

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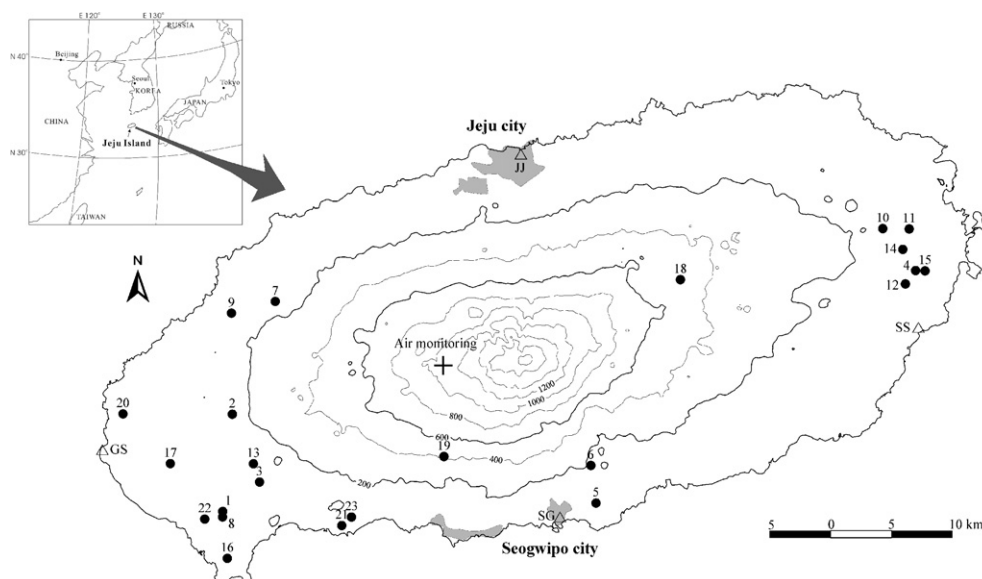
river to nearby wells. Clark et al. (2005) traced movement of water in a permeable alluvial aquifer from artificial recharge ponds using the gaseous tracers SF<sub>6</sub> and <sup>3</sup>He.

Environmental sulfur hexafluoride (SF<sub>6</sub>) from atmospheric sources was employed as a dating tool of young groundwater by Busenberg and Plummer (2000). They developed a very sensitive gas chromatography system with electron capture detector to detect low levels of SF<sub>6</sub> in groundwater of atmospheric origin and determined the natural background for SF<sub>6</sub>. Because the SF<sub>6</sub> atmospheric history is well known, SF<sub>6</sub> has been used as an environmental tracer in multi-tracer studies to estimate ages and mixing properties of groundwater (Plummer et al., 2001). In their study of discharge from springs in the Blue Ridge Mountains of Virginia, USA, reliable ages for young groundwater were obtained by SF<sub>6</sub> dating in igneous and metamorphic terrain where terrigenous SF<sub>6</sub> had not accumulated because of the young age. CFC ages could not be resolved because of the turnover in air input functions. The <sup>3</sup>H/<sup>3</sup>He ages of waters from the shallow springs were typically biased 1–2 years younger than the SF<sub>6</sub> ages. Because the aqueous diffusion coefficient of He ( $6.30 \times 10^{-5}$  cm<sup>2</sup>/s at 15 °C from Jahne et al., 1987) is seven times larger than that of SF<sub>6</sub> ( $9.20 \times 10^{-6}$  cm<sup>2</sup>/s at 15 °C from King and Saltzman, 1995), the waters may have lost some <sup>3</sup>He (Plummer et al., 2001) due to incomplete He confinement. Zoellmann et al. (2001) used environmental SF<sub>6</sub> to delineate nitrate contamination of groundwater from tertiary basalts in conjunction with <sup>3</sup>H. Zuber et al. (2005) studied different hydrochemical water types and groundwater mixing in a sandy aquifer in southern Poland using SF<sub>6</sub> and <sup>3</sup>H where CFCs could not be used for groundwater dating due to contamination of atmospheric CFCs by local sources. Plummer et al. (2003) and Plummer (2005) reviewed applications of SF<sub>6</sub> and other environmental tracers to dating young groundwater. Goody et al. (2006) studied the Chalk aquifer consist-

ing of microporous and fractured carbonate rocks located in a rural area of southern England using CFC-12 and SF<sub>6</sub>. They characterized groundwater flow regimes and evaluated the mixing of tracer-free old water and recent recharge, and that of groundwater and surface water near a river.

SF<sub>6</sub> has many advantages as an environmental tracer in comparison to CFCs. SF<sub>6</sub> does not react with the substrate or sorb onto aquifer organic matter and apparently does not biodegrade, even in highly reducing environments as is the case with chlorofluorocarbons (Busenberg and Plummer, 2000). Contamination of groundwater from non-atmospheric anthropogenic sources of SF<sub>6</sub> is less likely than contamination with CFCs. SF<sub>6</sub> is mainly used as an electrical insulator in high-voltage switches and transformers and as a blanket gas in the melting operations of magnesium metal production (Maiss and Brenninkmeijer, 1998), but SF<sub>6</sub> is not present in any commonly used household products. In addition, SF<sub>6</sub> does not readily sorb into rubber and polymers; therefore, groundwater is less likely to be contaminated with SF<sub>6</sub> from contact with these materials during sampling. Atmospheric concentrations of SF<sub>6</sub> are expected to continue to increase into the 21st century (Ko et al., 1993). This enables SF<sub>6</sub> to be continuously used as a hydrologic tracer in the future, which is not the case for CFCs that have atmospheric mixing ratios that have decreased from the 1990s. The dating range with SF<sub>6</sub> applies to water recharged in 1970 to the present.

Dating of young groundwater with SF<sub>6</sub> can, in some instances, be complicated by natural terrigenous sources of SF<sub>6</sub>, such as from some diagenetic, igneous and volcanic fluids, presumably within fluid inclusions in mineral grains (Harnisch and Eisenhauer, 1998; Harnisch et al., 2000; Busenberg and Plummer, 2000), and by addition of excess air during recharge (Busenberg and Plummer, 2000). Furthermore, urban air can be elevated in SF<sub>6</sub> from local and regional sources (Ho and Schlosser, 2000; Santella et al., 2003).



**Figure 1** Location of groundwater sampling sites. A cross shows the site of air monitoring for CFCs and SF<sub>6</sub>. Numbers on the contour lines are altitudes in meters above mean sea level (Korean geodetic vertical datum was established in 1916 based on measured sea level at Incheon Port). Triangles are meteorological stations of Korea Meteorological Agency, where JJ = Jeju, GS = Gosan, SG = Seogwipo, and SS = Seongsanpo. Shaded areas locate the two major cities of Jeju and Seogwipo on the island.

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