



Highly efficient electro-optically tunable smart-supercapacitors using an oxygen-excess nanograin tungsten oxide thin film

Akbar I. Inamdar^{a,*}, Jongmin Kim^a, Yongcheol Jo^a, Hyeonseok Woo^a, Sangeun Cho^a, Sambhaji M. Pawar^a, Seongwoo Lee^a, Jayavant L. Gunjekar^a, Yuljae Cho^b, Bo Hou^b, SeungNam Cha^b, Jungwon Kwak^c, Youngsin Park^d, Hyungsang Kim^{a,*}, Hyunsik Im^{a,*}

^a Division of Physics and Semiconductor Science, Dongguk University, Seoul 04620, South Korea

^b Department of Engineering Science, University of Oxford, Parks Road, OX1 3PJ, UK

^c Medical Physics Department, Asan Medical Center, Seoul, South Korea

^d School of Natural Science, Ulsan National Institute of Science and Technology, Ulsan 44919, South Korea

ARTICLE INFO

Keywords:

Multi-functional electrode
Oxygen-excess tungsten oxide
Nanograin
Electrochromism
Supercapacitor

ABSTRACT

A smart supercapacitor shares the same electrochemical processes as a conventional energy storage device while also having electrochromic functionality. The smart supercapacitor device can sense the energy storage level, which it displays in a visually discernible manner, providing increased convenience in everyday applications. Here, we report an electro-optically tunable smart supercapacitor based on an oxygen-rich nanograin WO₃ electrode. The nanostructured WO₃ electrode is dark blue in color in the charged state and becomes transparent in its discharged state with a high optical modulation of 82%. The supercapacitor has a specific capacitance of 228 F g⁻¹ at 0.25 Ag⁻¹ with a large potential window (1.4 V). It is highly durable, exhibits good electrochemical stability over 2000 cycles, retains 75% of its initial capacitance, and exhibits high coloration efficiency (~170 cm²/C). The excellent electrochromic and electrochemical supercapacitor properties of the electrode is due to the synergetic effect between nanograin morphology and excess oxygen. A smart-supercapacitor fabricated with an oxygen-rich nanograin WO₃ electrode exhibits a superb combination of energy storage and highly-efficient electrochromic features in one device that can monitor the energy storage level through visible changes in color.

1. Introduction

As the demand for advanced electronic devices increases, it is inevitable that greater functionality is integrated into energy storage devices in order to support the growing range of applications in portable electronics. While using an electronic device, it would certainly be useful and advantageous if the user could determine the amount of energy that has been consumed before the device stops working. Various transition metal oxides, such as WO₃ [1–14], TiO₂ [15–17], Nb₂O₅ [18–20], V₂O₅ [21,22] and NiO [23–33] show electrochromic behaviors that are determined largely by ion intercalation and deintercalation processes. This electrochemical mechanism is similar to the operating principle of an electrochemical supercapacitor. Thus, electrochromic (or smart) supercapacitors fill the niche of advanced electronic devices by functioning as a normal supercapacitor to store energy, and also by sensing the energy storage level through a change in the visual color (visualization of the device charge state as the

degree of color saturation) [34–39]. In general, in order to maximally enhance electrochromic and supercapacitive performances, more complex electrode materials need to be used. However, for complex compound electrodes with high performance such as W₁₈O₄₉/PANI composite [35], WO₃/ZnWO₄ compounds [4], and PANI/WO₃ nanocomposite [41], reproducibility and reliability in fabrication and performance are challenging issues. Thus, it is desirable to obtain smart supercapacitor electrode materials that are reliable, earth-abundant, and easy to fabricate with a conventional technique.

Pure WO₃ is a well-known electrochromic material showing fast redox reaction, good chemical stability, and strong adherence to the substrate [33,35,40,41]. However, its potential application for a smart supercapacitor has not yet been fully explored. This is presumably because its electrochemical energy storage properties are anticipated to be unsatisfactory. The nanostructured morphologies of electrode materials can be very beneficial to enhance electrochemical reactions and efficiency. Herein, we employ a nanograin WO₃ film as a bifunc-

* Corresponding authors.

E-mail addresses: akbarphysics2002@gmail.com (A.I. Inamdar), hskim@dongguk.edu (H. Kim), hyunsik7@dongguk.edu (H. Im).

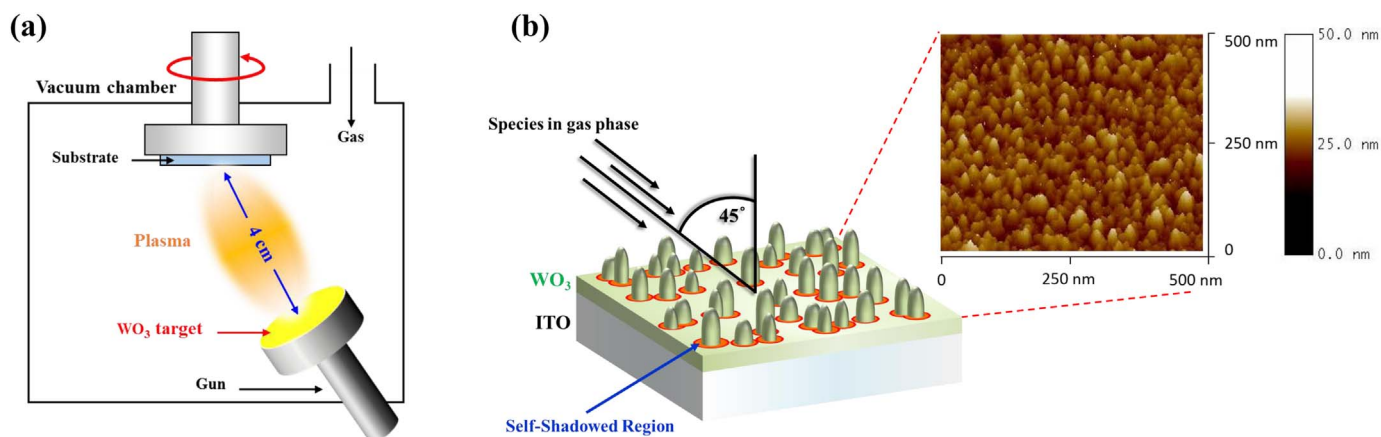


Fig. 1. Nanogranular growth of tungsten oxide during sputtering. (a) Schematic illustrations of RF-magnetron sputtering in which the angle and distance between target and substrate is 45° and 4 cm respectively. (b) 3-dimensional view showing growth of nanograin WO_3 electrode film together with an actual AFM image of a sample.

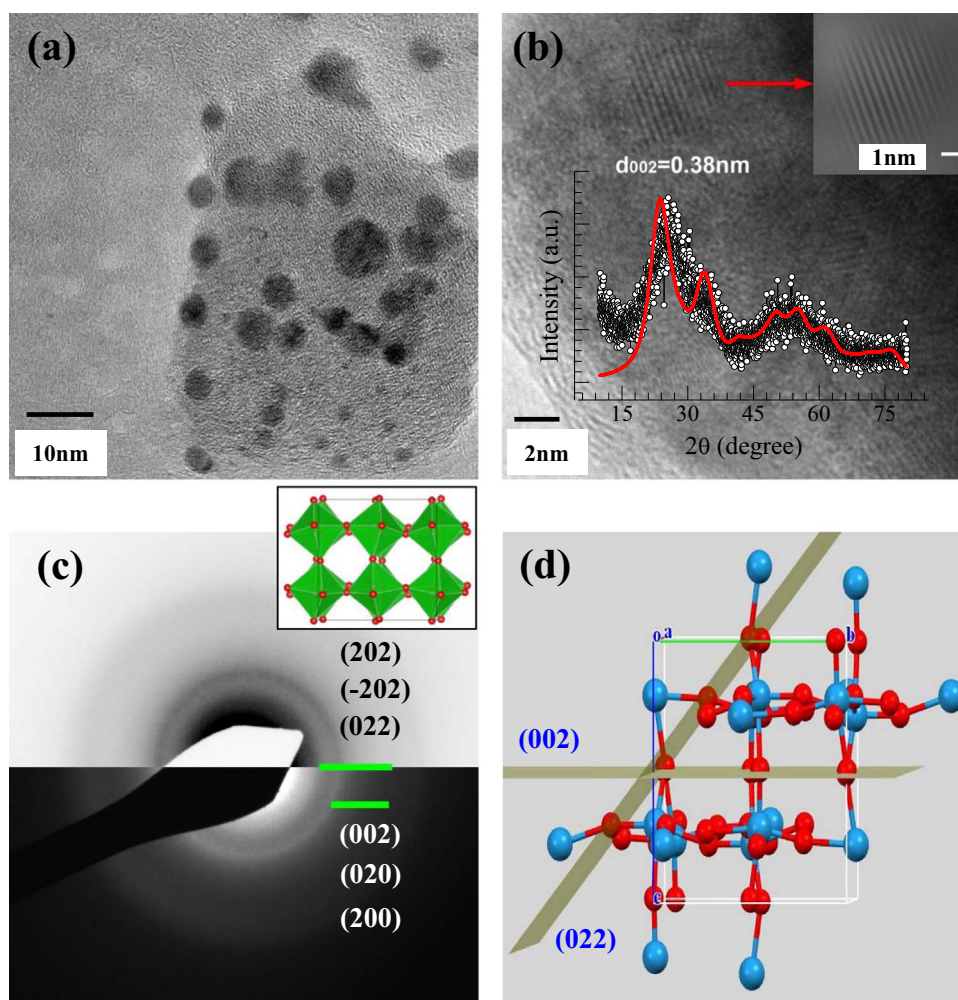


Fig. 2. (a) TEM and (b) HR-TEM images with 10 and 2 nm scale bars, respectively. The inset shows the X-ray nature of the synthesized film. The SAED pattern of the corresponding film is shown in (c), and the inset shows that the monoclinic WO_3 phase is mostly comprised of corner-linked octahedra of 4-membered channels along all crystallographic directions. The 002 and 022 crystallographic planes obtained from the SAED patterns are projected in (d). All crystallographic projections are made with the VESTA structure drawing software.

tional electrode for smart supercapacitor devices. The nanograin WO_3 layer was fabricated via oblique-angle sputtering and exhibited excellent electrochromic and supercapacitive properties owing to the large specific surface area as well as the monoclinic phase which facilitates electrochemical ion-insertion/extraction reactions and efficiency. Using a nanograin WO_3 electrode in a half cell configuration, excellent coloration efficiency ($\sim 170 \text{ cm}^2 \text{ C}^{-1}$) and good specific capa-

citance (228 F g^{-1} at 0.25 A g^{-1}) are achieved.

2. Experimental section

Oxygen-excess nanograin WO_3 thin film electrodes were fabricated on ITO-coated conducting glass substrates with a sheet resistance of $27 \Omega \text{ cm}^{-2}$ using conventional radio frequency (RF) magnetron sput-

Download English Version:

<https://daneshyari.com/en/article/4758835>

Download Persian Version:

<https://daneshyari.com/article/4758835>

[Daneshyari.com](https://daneshyari.com)