Accepted Manuscript

Smart Polymeric Nanoparticles for Boron Scavenging

Sérgio Alves, Catarina Santos, André P. da Costa, Mara Silva, Carlos Baleizão, José Paulo S. Farinha

PII:	S1385-8947(17)30320-0
DOI:	http://dx.doi.org/10.1016/j.cej.2017.02.139
Reference:	CEJ 16578
To appear in:	Chemical Engineering Journal

Received Date:17 October 2016Revised Date:23 February 2017Accepted Date:26 February 2017



Please cite this article as: S. Alves, C. Santos, A.P. da Costa, M. Silva, C. Baleizão, J.P.S. Farinha, Smart Polymeric Nanoparticles for Boron Scavenging, *Chemical Engineering Journal* (2017), doi: http://dx.doi.org/10.1016/j.cej. 2017.02.139

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

ACCEPTED MANUSCRIPT

Smart Polymeric Nanoparticles for Boron Scavenging

Sérgio Alves, Catarina Santos, André P. da Costa, Mara Silva, Carlos Baleizão *, José Paulo S. Farinha *

CQFM – Centro de Química-Física Molecular and IN – Institute of Nanoscience and Nanotechnology, Instituto Superior Técnico, Universidade de Lisboa, 1049-001 Lisboa, Portugal

* Corresponding authors: farinha@tecnico.ulisboa.pt; carlos.baleizao@tecnico.ulisboa.pt

Abstract

Boron is biologically beneficial but only in trace quantities. It is hard to detect and remove from water, as sometimes required in the treatment of residual waters. We have designed an approach based on thermo-responsive polymer nanoparticles containing vicinal diol groups for boron scavenging. The particles have a core of poly(methyl methacrylate) (PMMA) and a thermosensitive shell with a brush composed of a copolymer of *N*-isopropylacrylamide (NIPAM), 2-aminoethyl methacrylate (AEMH), and either D-gluconoamidoethyl methacrylate (GAEM) or monodiol methacrylate (MDM) boron chelating monomers. The nanoparticles show boron removal efficiencies of up to 96%, with removal of phenylboronic acid being more efficient than that of boric acid. The best boron scavenging efficiency was obtained for the higher specific surface area particles containing D-gluconoamidoethyl groups. At temperatures above *ca*. 35°C the particle shell collapses, inducing particle aggregation which allows simple separation and recovery of the nanoparticles. These can be subsequently used for boron scavenging, without loss of efficiency.

Keywords

Boron; Nanoparticles; Scavenging; Thermo-responsive; Water treatment

Download English Version:

https://daneshyari.com/en/article/4763123

Download Persian Version:

https://daneshyari.com/article/4763123

Daneshyari.com