## Accepted Manuscript

Synthesis of Epichlorohydrin from 1,3-Dichloropropanol Using Solid Base

Yangcheng Lu, Tianyang Li, Rui Wang, Guangsheng Luo

PII: S1004-9541(16)30638-3

DOI: doi: 10.1016/j.cjche.2016.08.019

Reference: CJCHE 656

To appear in:

Received date: 30 June 2016 Revised date: 18 August 2016 Accepted date: 22 August 2016



Please cite this article as: Yangcheng Lu, Tianyang Li, Rui Wang, Guangsheng Luo, Synthesis of Epichlorohydrin from 1,3-Dichloropropanol Using Solid Base, (2016), doi: 10.1016/j.cjche.2016.08.019

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Catalysis, kinetics and reaction engineering

Synthesis of Epichlorohydrin from 1,3-Dichloropropanol Using Solid Base $^{\star}$ 

Yangcheng Lu\*, Tianyang Li, Rui Wang, Guangsheng Luo

State Key Laboratory of Chemical Engineering, Department of Chemical Engineering, Tsinghua University,

Beijing 100084, China

\*Supported by the National Natural Science Foundation of China (21176136, 21422603) and the National Science

and Technology Support Program of China (2011BAC06B01).

**Abstract** 

In this work, the synthesis of epichlorohydrin (ECH) from 1,3-dichloropropanol (DCP) by using solid sodium

hydroxide (NaOH) is carefully investigated. Inert organic solvent, 1-octanol, is introduced to ensure reaction

intensity under control. The reaction performances with respects of apparent kinetics and selectivity are determined

to explore optimized reaction conditions and confirm potentials for enhancing productivity in one batch. The

dissolution and liquid phase reaction mechanism and instant reaction assumption are proposed and verified through

process analysis. A process design towards free additional water is schematically figured out to manipulate solid

NaOH, by-product, and unreacted starting materials to realize a nearly closed circuit. This process allows high

selectivity over 97 % and complete DCP conversion at 323.2 K within a reaction time less than 20 min. Other

advantages includes near-zero wastewater emission, economically possible NaOH regeneration from NaCl, and

robust operating condition window.

Keywords: Clean processes, Reaction, Process design, Kinetics, Multiphase

1

## Download English Version:

## https://daneshyari.com/en/article/4764263

Download Persian Version:

https://daneshyari.com/article/4764263

<u>Daneshyari.com</u>