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Data Article

## Data on physicochemical properties of active films derived from plantain flour/PCL blends developed under reactive extrusion conditions



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#### ARTICLE INFO

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#### ABSTRACT

The data given below relates to the research paper entitled: "Ecofriendly films prepared from plantain flour/PCL blends under reactive extrusion conditions using zirconium octanoate as a catalyst", recently published by our research group [1]. This article provides information concerning the physicochemical properties of the above-mentioned film systems: thickness, density, opacity, moisture content and surface moisture.

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#### **Specifications Table**

Subject area	Polymers.
More specific sub-	Active eco-friendly films derived from plantain flour/PCL blends using zirconium
ject area	octanoate $(Zr(Oct)_4)$ as a catalyst under reactive extrusion (REx) conditions.
Type of data	Table.

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How data was acquired	Thickness was determined with a digital micrometer. Density and moisture content were calculated gravimetrically. Opacity and surface moisture were estimated with the aid of a UV-vis spectrophotometer (u-2001) and a Moisture Analyzer (Model MA150), respectively.
Data format	Raw, calculated and analyzed.
Experimental	The films were conditioned at ~ 57% relative humidity in desiccators at 25 °C for
factors	7 days using a saturated NaBr solution.
Experimental	Film thickness was determined from eighteen random positions on each sample.
features	Film density and moisture contents were determined at 105 $\pm$ 1 °C for 24 h.
	Film opacity was measured at 600 nm. Surface moisture content of the different
	films was measured after drying at 105 °C for 120 s.
Data source location	Mar del Plata, Argentina.
Data accessibility	Data are presented in this article.

#### Value of the data

- The data gives detailed descriptions of the physicochemical properties of active eco-friendly films based on plantain flour/PCL blends using Zr(Oct)<sub>4</sub> as a catalyst under REx conditions.
- The data provides information to researchers about the effects of the catalyst and the PCL-containing blends on the opacity of the materials.

#### 1. Data

Tabla 1

 $\rho$  (g/cm<sup>3</sup>)

Opacity

MC (%)

(%)

Surface moisture

 $1.2~\pm~0.1^a$ 

 $0.9 \pm 0.6^{a}$ 

The data given in this study (Table 1) shows the physicochemical properties (thickness, density, opacity, moisture content and surface moisture) of active films derived from plantain flour/PCL blends under REx conditions, described in the article by Gutiérrez and Alvarez [1]. These characteristics add to the properties previously investigated [1]. This data widens the knowledge we have about the physicochemical properties of plantain flour/PCL systems cross-linked under REx conditions using Zr (Oct)<sub>4</sub> as a catalyst. Thickness, density, opacity, moisture content as well as surface moisture were evaluated.

Equal letters in the same row indicate no statistically significant differences ( $p \leq 0.05$ ).

Film systems: plantain flour (TPPF), plantain flour + PCL ( $M_w = 10,000 \text{ g/mol}$ ) (TPPF/PCL(10,000)), plantain flour + PCL ( $M_w = 10,000 \text{ g/mol}$ ) + catalyst (TPPF/PCL(10,000)+CAT), plantain flour + PCL  $(M_w = 80,000 \text{ g/mol})$  (TPPF/PCL(80,000)) and plantain flour + PCL  $(M_w = 80,000 \text{ g/mol})$  + catalyst (TPPF/PCL(80,000)+CAT).

Thickness (e), density ( $\rho$ ), opacity, moisture content ( <i>MC</i> ) and surface moisture of the different films.								
Parameter	TPPF	TPPF/PCL (10,000)	TPPF/PCL(10,000)+ CAT	TPPF/PCL (80,000)	TPPF/PCL(80,000)+ CAT			
<i>e</i> (mm)	1.23 ± 0.	$04^{b}$ 1.03 $\pm$ 0.04 <sup>a</sup>	$1.04~\pm~0.04^{a}$	$1.17~\pm~0.06^{\mathrm{b}}$	$1.14~\pm~0.05^{b}$			

 $1.19~\pm~0.04^a$ 

 $0.22 ~\pm~ 0.01^{b,c}$ 

 $16.64\ \pm\ 0.08^{b}$ 

 $0.7 \pm 0.1^{a}$ 

 $1.1~\pm~0.1^a$ 

 $16~\pm~1^{b}$ 

 $0.8 \pm 0.1^{a}$ 

 $0.16~\pm~0.01^a$ 

 $1.16 \pm 0.09^{a}$ 

 $0.19 ~\pm~ 0.01^{\rm b}$ 

 $13.3\ \pm\ 0.4^a$ 

 $0.72 \pm 0.02^{a}$ 

Thickness (e), density ( $\rho$ ), opacity, moisture content (MC) and surface moisture of	the different films.

 $1.24~\pm~0.03^a$ 

 $0.8 \pm 0.2^{a}$ 

 $0.32~\pm~0.01^{d}~~0.20~\pm~0.01^{b}$ 

 $17.2~\pm~0.5^{\rm b,c}~~16.1~\pm~0.6^{\rm b}$ 

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