Accepted Manuscript

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| PII: | \$1572-6657(16)30742-1 |
|----------------|--|
| DOI: | doi: 10.1016/j.jelechem.2016.12.040 |
| Reference: | JEAC 3036 |
| To appear in: | Journal of Electroanalytical Chemistry |
| Received date: | 10 November 2016 |
| Revised date: | 13 December 2016 |
| Accepted date: | 23 December 2016 |
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Please cite this article as: Cheng Fang, Mallavarapu Megharaj, Ravi Naidu , Electrochemical switch on-off response of a self-assembled monolayer (SAM) upon exposure to perfluorooctanoic acid (PFOA). The address for the corresponding author was captured as affiliation for all authors. Please check if appropriate. Jeac(2016), doi: 10.1016/j.jelechem.2016.12.040

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Electrochemical switch on-off response of a self-assembled monolayer (SAM) upon exposure to perfluorooctanoic acid (PFOA)

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Abstract:

The switch on-off response of a self-assembled monolayer (SAM) on a stimulating moiety such as a surfactant is interesting but not well known. Priority research attention has been paid to a special family of surfactant, fluorosurfactants, because they have been recently listed as emerging contaminants and persistent organic pollutants. Herein we investigate the electrochemical switch on-off response of a SAM of 6 (ferrocenyl, FC) hexanethiol (C_6 -FC) upon exposure to perfluorooctanoic acid (PFOA), the most common fluorosurfactant. FC is employed as a redox probe to monitor the switch on-off response of the SAM formed on gold electrode surface. In an effort to understand the hydrophilic terminate group's influence from the surfactant, we dope the pristine C_6 -FC SAM with 6-amino-1-hexanethiol (C_6 -NH₂), 1-hexanethiol (C_5 -CH₃), or 6-mercaptohexanoic acid (C_5 -COOH) via a thiol-gold bond. We also dope it with 1-dodecanethiol (C_{11} -CH₃) or 12-mercaptododecanoic acid (C_{11} -COOH) to investigate the length effect of the carbon chain. We observe there is a similar switch on-off response either from the presence of PFOA or from the doping C_5 -COOH/ C_{11} -COOH, suggesting the insertion of PFOA into the C_6 -FC SAM, which is also evidenced by electrochemical

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