

## Influence of synthesis parameters on particle properties and catalytic activity of rice roll-like Au/SiO<sub>2</sub> nanocatalysts prepared in inverse miniemulsions

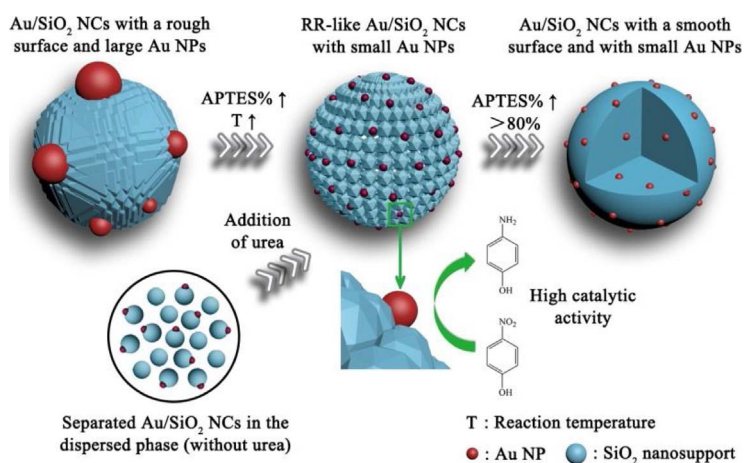


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### GRAPHICAL ABSTRACT



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### ABSTRACT

In our previous paper, we reported the synthesis of easily recyclable and highly active rice roll-like (RR-like) Au/SiO<sub>2</sub> nanocatalysts (NCs) in inverse miniemulsions [Colloids Surf. A 517, 2017, 52–62]. In the present contribution, the influences of three key synthesis parameters, including the 3-aminopropyltriethoxysilane (APTES) content, reaction temperature, and urea amount, on the particle properties of the Au/SiO<sub>2</sub> NCs were systematically investigated. The role of these synthesis parameters in the formation of SiO<sub>2</sub> nanosupports and Au nanoparticles (NPs) was discovered. The RR-like SiO<sub>2</sub> nanosupports could only be formed in the systems within the range of 20–80 mol% of APTES content, at reaction temperatures above 60 °C, and with more than 0.72 mg of urea. The AuCl<sub>4</sub><sup>−</sup> ions could be immobilized onto the aminopropyl-functionalized SiO<sub>2</sub> nanosupports through a coordination interaction. The immobilized ions were reduced by the aminopropyl groups at elevated temperatures. The APTES content and reaction temperature play an important role in controlling the particle size and its distribution of Au NPs. The catalytic activity of Au/SiO<sub>2</sub> NCs depends significantly on the APTES content.

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The RR-like Au/SiO<sub>2</sub> NCs with smaller immobilized Au NPs and higher specific surface area display higher catalytic activity towards hydrogenation of *p*-nitrophenol.

## 1. Introduction

Au nanoparticles (NPs) have become a kind of commonly-used nanocatalysts (NCs), as they may display an excellent catalytic activity towards many organic reactions, such as *p*-nitrophenol (*p*-NPh) reduction [1], alkyne hydrochlorination [2], alkene epoxidation [3], and alcohol oxidation [4]. However, in the application of Au NPs, they easily undergo agglomeration due to the high surface energy. Therefore, much effort has been spent to suppress the agglomeration of Au NPs. One effective strategy is to immobilize Au NPs onto various porous supports, such as SiO<sub>2</sub> [5,6], TiO<sub>2</sub> [7–9], C [10], and metal–organic frameworks [11–14]. SiO<sub>2</sub>, which could be easily synthesized through the (modified) Stöber method [15], has been often used as support for Au NPs taking advantage of its high specific surface area, tunable pore properties, good chemical and thermal stabilities, good mechanical strength, and good biocompatibility [16–18]. Versatile techniques have been devised to prepare Au/SiO<sub>2</sub> NCs, such as impregnation, colloidal deposition, and deposition-precipitation [19–22]. However, these synthetic techniques often suffer from the low immobilization efficiency of Au NPs to the SiO<sub>2</sub> nanosupports.

In recent years, inverse miniemulsion technique has been proven to be an effective method to prepare various hydrophilic nanoparticles or nanocomposite particles [23,24]. In a typical inverse miniemulsion system, polar nanodroplets are homogeneously dispersed in a hydrophobic continuous phase. Each polar nanodroplet can be regarded as a separated nanoreactor [23]. Landfester et al. prepared hydrophilic SiO<sub>2</sub> and TiO<sub>2</sub> NPs through a sol–gel process of hydrophilic inorganic precursors [25,26]. We prepared SiO<sub>2</sub> nanocapsules through a sol–gel process of commercially available hydrophobic silica precursors in inverse miniemulsions [27]. Recently, we further prepared various noble metal/inorganic support NCs, such as Ag/TiO<sub>2</sub>, Au/TiO<sub>2</sub>, Au/SiO<sub>2</sub>, and Pd/SiO<sub>2</sub> NCs, through a two-step process in inverse miniemulsions [28–31]. In detail, we first prepare inorganic nanosupports through a sol–gel process of inorganic precursors, and subsequently, noble metal NPs are formed through reduction of noble metal salts by post-addition of a strong reducing agent [28–31]. This technique holds many advantages including versatile combinations of the noble metal and inorganic supports and the high immobilization efficiency of noble metal NPs. However, the noble metal NPs synthesized by the two-step process display a relatively large particle size and broad particle size distribution. Very recently, we reported a one-step inverse miniemulsion-based technique for preparation of rice roll-like (RR-like) Au/SiO<sub>2</sub> NCs through introduction of 3-aminopropyltriethoxysilane (APTES) to the reaction system [32]. In addition to participation in the formation of SiO<sub>2</sub> nanosupports, APTES could reduce gold salts to Au NPs, simultaneously. Finally, RR-like Au/SiO<sub>2</sub> NCs containing narrowly size-distributed sub-10 nm Au NPs were prepared through this technique. The RR-like Au/SiO<sub>2</sub> NCs displayed excellent catalytic activity, recycling catalytic performance, and reversible pH-dependent colloidal stability.

As a successive work of the preparation of RR-like Au/SiO<sub>2</sub> NCs through one-step process in inverse miniemulsions, the influences of critical parameters on the particle properties of SiO<sub>2</sub> nanosupports and Au NPs were systematically investigated to understand their role in the formation of Au/SiO<sub>2</sub> NCs and finally to controllably synthesize Au/SiO<sub>2</sub> NCs with a RR-like SiO<sub>2</sub> nanosupports and narrowly size-distributed small Au NPs. The APTES content and reaction temperature display a significant influence on the particle morphology of the SiO<sub>2</sub> nanosupports and the particle size and particle size distribution of Au NPs. Urea plays an important role in the formation of the RR-like morphology of SiO<sub>2</sub> nanosupports. The catalytic activity significantly

depends on the particle properties of the Au/SiO<sub>2</sub> NCs.

## 2. Experimental section

### 2.1. Materials

Tetrachloroauric acid (HAuCl<sub>4</sub>, 48%–50%), tetramethoxysilane (TMOS, 98%), APTES (99%), dimethyl sulfoxide (DMSO, 99%), urea (99%), *p*-nitrophenol (*p*-NPh, 99%), and sodium borohydride (NaBH<sub>4</sub>, 98%) were purchased from Aladdin Chemistry Co. Ltd. and used as received. *n*-Hexadecane (HD, 98%) was purchased from Macklin. Poly(ethylene-*co*-butylene)-*b*-poly(ethylene oxide) (P(E/B)–PEO) with number average molecular weight of 7100 g mol<sup>−1</sup> was synthesized according to the literature [33]. The molecular weights of the hydrophobic (E/B) and hydrophilic (EO) blocks are 4000 and 3100 g mol<sup>−1</sup>, respectively. These weights resulted in a P(E/B)–PEO hydrophilic–lipophilic balance of 8.7. Cyclohexane (99.5%) and ethanol (99.7%) were purchased from Hangzhou Gaojing Fine Chemical Co. Ltd. Demineralized water was used in all experiments.

### 2.2. Preparation of Au/SiO<sub>2</sub> NCs

0.3 g aqueous solution of urea with various concentrations (0–0.14 mol L<sup>−1</sup>) was mixed with 1.0 g of DMSO, and then 0.043 g of HAuCl<sub>4</sub> was added to the mixed polar solution, which was used as the dispersed phase. 4 wt% of P(E/B)–PEO relative to the dispersed phase was dissolved in 12.5 g of HD to form a hydrophobic solution, which was used as the continuous phase. These two solutions were mixed with an agitation of 700 rpm at 40 °C for 15 min to form a crude emulsion. The crude emulsion was sonicated by using a pulse mode (work 12 s, break 6 s) at 42% maximum power of the Scientz JY92-IIDN sonifier for 9 min to form an inverse miniemulsion.

Two grams of the prepared inverse miniemulsion was added to a glass vessel, and then the mixtures of TMOS and APTES with various molar contents of APTES (0–100 mol%) were dropwise added to the inverse miniemulsion, respectively. The sol–gel process of two silica precursors ran for 24 h with an agitation of 400 rpm at various temperatures (40–90 °C) to obtain dispersions of the Au/SiO<sub>2</sub> NCs. The recipes of the prepared Au/SiO<sub>2</sub> NCs are listed in Table 1.

The obtained Au/SiO<sub>2</sub> NCs were purified through two centrifugation–redispersing cycles in cyclohexane and three centrifugation–redispersing cycles in ethanol to obtain purified Au/SiO<sub>2</sub> NC powders. The Au/SiO<sub>2</sub> NCs were dried in a vacuum oven at 50 °C overnight. The dried powders for catalytic tests were further calcined at 500 °C for 1 h in a muffle furnace to remove organic components.

### 2.3. Evaluation of the complexation of AuCl<sub>4</sub><sup>−</sup> with APTES

The interaction between the AuCl<sub>4</sub><sup>−</sup> ions and APTES was evaluated by ultraviolet–visible (UV–vis) spectroscopic measurements. In detail, 0.04 g of HAuCl<sub>4</sub> was dissolved into 10 g of ethanol, and then various amounts of APTES (0–0.07 g) were added to the solution of HAuCl<sub>4</sub>. After mixing by magnetic agitation of 700 rpm for 1 min, the UV–vis spectra of the mixed solutions were recorded on a Shimadzu UV-2600 UV–vis spectrometer.

### 2.4. Evaluation of the catalytic activity of the Au/SiO<sub>2</sub> NCs

Catalytic activity of the Au/SiO<sub>2</sub> NCs was evaluated by hydrogenation of *p*-NPh with NaBH<sub>4</sub>. Five milligrams of the Au/SiO<sub>2</sub> NCs was

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