

Accepted Manuscript

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PII: S0021-9797(17)30248-5
DOI: <http://dx.doi.org/10.1016/j.jcis.2017.03.008>
Reference: YJCIS 22105

To appear in: *Journal of Colloid and Interface Science*

Received Date: 20 December 2016
Revised Date: 27 February 2017
Accepted Date: 1 March 2017

Please cite this article as: W. Konicki, A. Hełminiak, W. Arabczyk, E. Mijowska, Removal of anionic dyes using magnetic Fe@graphite core-shell nanocomposite as an adsorbent from aqueous solutions, *Journal of Colloid and Interface Science* (2017), doi: <http://dx.doi.org/10.1016/j.jcis.2017.03.008>

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Removal of anionic dyes using magnetic Fe@graphite core-shell nanocomposite as an adsorbent from aqueous solutions

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Keywords: Anionic dyes, Adsorption, Magnetic nanocomposite, Kinetics, Thermodynamics.

Abstract

In this study, magnetic Fe@graphite nanocomposite (Fe@G-N) with a core-shell structure was prepared by chemical vapor deposition CVD process for the adsorptive removal of anionic dyes from aqueous solutions. Fe@G-N was characterized by XRD, HRTEM, HAADF-STEM, FTIR, Raman spectroscopy, BET and zeta potential measurements, and then applied in adsorption of two kinds of anionic dyes, Acid Red 88 (AR88) and Direct Orange 26 (DO26). The effect of parameters like initial dye concentration (5-40 mg L⁻¹), pH solution (4-10) and temperature (20-60°C) on the adsorption process was studied. The pseudo-first-order, pseudo-second-order kinetic models and the intraparticle diffusion model were used to describe the kinetic data. A kinetic study indicated that a pseudo-second-order model agreed

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