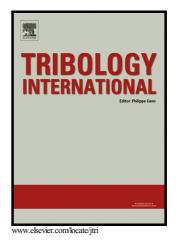
Author's Accepted Manuscript

Investigation of chemical tooth mechanism in chemical mechanical planarization of germanium

Apeksha Gupta, S. Noyel Victoria, R. Manivannan



 PII:
 S0301-679X(17)30146-9

 DOI:
 http://dx.doi.org/10.1016/j.triboint.2017.03.029

 Reference:
 JTRI4657

To appear in: Tribiology International

Received date: 26 December 2016 Revised date: 11 March 2017 Accepted date: 21 March 2017

Cite this article as: Apeksha Gupta, S. Noyel Victoria and R. Manivannan Investigation of chemical tooth mechanism in chemical mechanical planarizatio of germanium, *Tribiology International* http://dx.doi.org/10.1016/j.triboint.2017.03.029

This is a PDF file of an unedited manuscript that has been accepted fo publication. As a service to our customers we are providing this early version o the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting galley proof before it is published in its final citable form Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain

ACCEPTED MANUSCRIPT

Investigation of chemical tooth mechanism in chemical mechanical planarization of

germanium

Apeksha Gupta, S. Noyel Victoria, R. Manivannan*

Department of Chemical Engineering, National Institute of Technology Raipur, Chhattisgarh-

492010, INDIA.

*Corresponding author. Tel.: +91-771-2254 200; fax: +91-771-2254 600.

rmani.che@nitrr.ac.in

Abstract

The effect of pH on germanium (Ge) chemical mechanical planarization (CMP) removal rate (RR) using polymorphs of titania was investigated. Rutile and anatase polymorphs of titania were used to study the removal rate of germanium in absence of oxidizer. Polishing experiments performed using rutile titania showed highest removal rate at pH 3, and subsequently decreased with pH and became negligible at pH 11. However, no material removal was observed with anatase polymorph for the entire pH range investigated in this work. Higher Ge removal using rutile abrasive can be attributed to the formation of Ti-O-Ge bond due to the structural similarity of polishing surface and abrasive. Ge RR showed a linear relationship with pressure and table speed, following Prestonian behavior.

Graphical Abstract

Download English Version:

https://daneshyari.com/en/article/4985956

Download Persian Version:

https://daneshyari.com/article/4985956

Daneshyari.com