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## **Research** Paper Numerical investigation of liquid methanol evaporation and oxy-combustion inside a button-cell ITM reactor

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HIGHLIGHTS

• Analysis of liquid methanol evaporation and oxy-combustion in an ITM reactor.

• A semi-empirical model is applied after fitting with the available LNO membrane data.

• Influences of inlet fuel fraction, inlet gas temperature and inlet sweep flux are studied.

• High combustion efficiency is encountered at moderate inlet gas temperatures.

• High fuel concentration at low inlet sweep flow resulted in high oxygen flux.

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#### ABSTRACT

A numerical study is conducted to investigate the performance of a button-cell LNO-ITM reactor utilizing the soot-free oxygenated liquid methanol under oxy-combustion condition. The Euler-Lagrange approach is utilized to solve discrete phase model. Taylor analogy breakup (TAB) model is used due to its convenience with the cases of low injection speed. A plain orifice atomizer is used for fuel atomization and CO<sub>2</sub> is used as sweep gas. A semi-empirical oxygen permeation model (ABn model) is validated with the available experimental data and is, then, applied in the present model. Over a wide range of inlet fuel concentrations, the results showed increase in oxygen permeation flux of about five times in cases of reacting conditions as compared to the cases of non-reacting cases. The results showed high oxygen permeation flux at low inlet fuel concentrations due to the improvement in the oxygen to fuel ratio toward the stoichiometric conditions. At inlet gas temperatures of 1223 K, 1123 K, 1023 K and 923 K, the combustion temperature approached 1423 K, 1347 K, 1284 K and 1231 K, respectively, indicating an average combustion efficiency of 43% at moderate inlet gas temperatures. High fuel concentration at low inlet sweep flow resulted in high oxygen flux and high combustion temperature.

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### 1. Introduction

For the time being, emission of greenhouse gases, mainly CO<sub>2</sub>, is the most critical challenge facing our existence on this globe. By the coming 2050, the emission levels of  $CO_2$  are expected to be increased by about 70% [1]. Most of the new technologies for power generation including renewable energy technologies are not economically visible yet [2]. In the present days, the oil price is reduced to very critical limits at which the developments of new clean technologies are significantly affected. Based on that, the world is forced to use fossil fuels for the production of energy which necessitates the development of technologies that can reduce the emissions of greenhouse gases while burning fossil fuels. Carbon capture (CC) technologies including precombustion, oxy-combustion and post combustion carbon capture are considered as the most effective tools regarding the control of CO<sub>2</sub> emissions [3]. Among these carbon capture technologies, oxycombustion technology is the lowest cost and the most efficient and promising carbon capture technology [4]. This technology can be used in existing power plants with slight modifications and, also, it can be used with the new power plants [5]. As well, oxy-combustion technology does not require any additional chemical separation process for the separation of CO<sub>2</sub>. Simply,





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2-D	two dimensional	Si	source/sink term
3-D	three dimensional	SIMPLE	semi-implicit method for pressure-linked equations
ABn	oxygen permeation model	SSE	sums of the squares of the error
ASU	air separation unit	TAB	taylor analogy breakup model
СС	carbon capture	$T_M$	membrane temperature
CFD	computational fluid dynamics	$T_{\infty}$	surrounding temperature
CH₃OH	methanol	UDF	user defined function
CO	carbon monoxide	VO	vegetable oil
CO <sub>2</sub>	carbon dioxide	CP	heat capacity of the particle (J/kg K)
DME	dimethyl ether	m <sub>P</sub>	mass of the particle (kg)
DNS	direct numerical simulations	Ap	surface area of the particle (m <sup>2</sup> )
DO	discrete ordinates	h	convective heat transfer coefficient (W/m <sup>2</sup> K)
EWBM	exponential wide band model	ε <sub>p</sub>	particle emissivity
FB	flow burning	σ	Stefan-Boltzmann constant ( $5.67 \times 10^{-8} \text{ W/m}^2 \text{ K}^4$ )
HC	hydrocarbons	$\theta_{R}$	radiation temperature (–)
$H_2O$	water vapor	d <sub>p</sub>	particle diameter (m)
ITM	ion transport membrane	$k_{\infty}$	thermal conductivity of the continuous phase (W/m K)
ITMR	ion transport membrane reactor	<i>Re</i> <sub>d</sub>	Reynolds number based on the particle diameter and
Jo2	oxygen permeation flux		the relative velocity (-)
LES	large eddy simulations	Pr	Prandtl number of the continuous phase (–)
LNO	La <sub>2</sub> NiO <sub>4</sub> ionic ceramic membrane	N <sub>v</sub>	$(\text{kg mol}/\text{m}^2 \text{ s})$
LSCF	lanthanum strontium cobaltite ferrite	k <sub>c</sub>	mass transfer coefficient (m/s)
NOx	nitric oxides	C <sub>v,D</sub>	vapor concentration at the droplet surface (kg mol/m <sup>3</sup> )
PDF	probability density function	$C_{v,\infty}$	vapor concentration in the bulk gas (kg mol/m <sup>3</sup> )
$P'_{O2}$	oxygen partial pressure in feed side	m <sub>D</sub>	mass of the droplet (kg)
$P_{02}''$	oxygen partial pressure in permeate side	A <sub>D</sub>	surface area of the droplet (m <sup>2</sup> )
RTE	radiative transfer equation	T <sub>D</sub>	surface temperature of the droplet (K)
sccm	standard cubic centimeter per minute	T <sub>R</sub>	radiation temperature (K)

the products of combustion of a hydrocarbon fuel with oxygen consist mainly of  $CO_2$  and  $H_2O$ , which can be passed through a simple condenser to separate  $H_2O$  and, then,  $CO_2$  is captured [6].

There are two main approaches regarding the application of oxy-combustion technology. The first is through the use of air separation unit (ASU) to separate oxygen from air, and then the produced oxygen is used in combustion in conventional combustion systems. However, this ASU requires additional powering energy for its operation in terms of gas heating (to activate the membrane for oxygen separation) and compression [7]. The second approach is through the use of what's called ion transport membrane reactor (ITMR) technology. In this technology, both oxygen separation and oxy-combustion processes are performed inside the same unit. An ion transport membrane (ITM) is used to separate oxygen from the flowing air in the feed side to the permeate side. In the permeate side, permeated oxygen across the membrane is burned with the provided fuel in a medium of recirculate  $CO_2$  [8]. The resultant design of the ITMR should result in a compact size of the system due to the combination of both the separation unit and the combustion unit in one common unit. Also, the resultant system will not require any additional powering for oxygen separation because the membrane is heated using part of the released heat of combustion.

In the industrial field, the use of liquid fuels is preferred due to many reasons including low volume as compared to gaseous fuels which results in easy handling and transportation and high energy density. This forces the research towards the design of new systems that can handle easily the liquid fuels. However, care should be taken while designing any combustion system utilizing liquid fuel in order to reduce soot formation, reduce NOx emissions, and properly vaporize the fuel [9]. The liquid fuel combustion is utilized in many industrial devices including diesel engines and gas turbine engines. The subject of liquid fuel evaporation and combustion has been studied numerically by Fujita et al. [10] using 2-D direct numerical simulations (DNSs) and by Moin and Apte [11] using large eddy simulations (LESs). Yin [12] performed numerical modeling of *n*-heptane droplets heating and evaporation aiming at the development of a generic model for the conversion of fuel droplet. They were able to develop a computer code for droplet heating and evaporation taking into account droplet dynamics and the interaction between the droplet and the free-stream. Kitano et al. [13] numerically investigated the influence of difference in fuel composition on the droplet evaporation and combustion. They reported that the evaporation rate becomes slower for multicomponent fuel as compared to single fuel. Targeting reduced emission and clean combustion, Jiang et al. [14] experimentally investigated the combustion process of different liquid fuels including diesel, biodiesel and straight vegetable oil (VO) using a novel design flow burning (FB) injector. The measurements of combustion temperature and NOx and CO emissions were recorded in their work. The results showed the capability of the FB injector to produce clean blue flames indicating mainly premixed combustion for all the tested fuels.

Emission control, especially soot formation, is another critical issue in case of liquid fuel combustion. Park and Yoon [15] proposed a two-stage injection strategy for simultaneous reduction of NOx and soot emissions in dimethyl ether (DME) fueled engine. The results showed low NOx, HC, CO and soot emissions while using the pilot injection strategy with advanced main injection. Speth et al. [16] conducted a study on the effect of using alternative jet fuels that are low in aromatic content on the reduction of black carbon emissions from gas turbines. They established a relationship between the reduction in black carbon emissions and the aromatic volume fraction of the alternative fuel. Choi et al. [17] studied the effect of liquid fuel doping on the formation of soot and polycyclic aromatic hydrocarbon in counter flow ethylene

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