



# Achieving nitrification in a continuous moving bed biofilm reactor at different temperatures through ratio control



Wei Bian, Shuyan Zhang, Yanzhuo Zhang, Wenjing Li, Ruizhe Kan, Wenxiao Wang, Zhaoming Zheng, Jun Li\*

College of Architecture and Civil Engineering, Beijing University of Technology, Beijing 100124, China

## HIGHLIGHTS

- A ratio control strategy was used to remain stable nitrification in MBBR.
- Oxygen-limiting conditions in biofilm reactor were determined in this study.
- Nitrification was recovered in low temperature of 6 °C through ratio control.
- Stable nitrification was achieved without complete wash-out of NOB.
- Choice of partial or full nitrification was recommended.

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## ABSTRACT

A ratio control strategy was implemented in a continuous moving bed biofilm reactor (MBBR) to investigate the response to different temperatures. The control strategy was designed to maintain a constant ratio between dissolved oxygen (DO) and total ammonia nitrogen (TAN) concentrations. The results revealed that a stable nitrification in a biofilm reactor could be achieved via ratio control, which compensated the negative influence of low temperatures by stronger oxygen-limiting conditions. Even with a temperature as low as 6 °C, stable nitrification could be achieved when the controlling ratio did not exceed 0.17. Oxygen-limiting conditions in the biofilm reactor were determined by the DO/TAN concentrations ratio, instead of the mere DO concentration. This ratio control strategy allowed the achievement of stable nitrification without complete wash-out of NOB from the reactor. Through the ratio control strategy full nitrification of sidestream wastewater was allowed; however, for mainstream wastewater, only partial nitrification was recommended.

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## 1. Introduction

Nitrification receives increasing interest due to its associated economic savings (e.g. organic carbon source and aeration energy) compared to classical nitrogen removal technologies (Turk and Mavinic, 1987). To achieve nitrification, several operational factors have been adopted, such as low dissolved oxygen (DO) (Hulle et al., 2010), high pH (Villaverde et al., 1997), high free ammonia (FA) (Hulle et al., 2010), high free nitrous acid (FNA) (Gabbro et al., 2012), high temperature, and suitable sludge retention time (Hulle et al., 2010). Recently, an oxygen-limiting ratio control (DO/TAN) strategy was successfully applied to biofilm airlift reactors to obtain full nitrification (almost 100% ammonia conversion to nitrite) under stable operating conditions (Carrera et al., 2010).

In theory and practice, nitrification with batch influent can easily be achieved compared to continuous influent (Zheng et al., 2016; Julio et al., 2009). With continuous influent, the closer the system is to plug-flow, the easier it is to obtain nitrification. In other words, the closer the system is to complete mixing of the liquid phase, the narrower the range of conditions under which nitrification is obtained (Julio et al., 2009). Plug-flow reactors perform better since the higher substrate concentration in the bulk liquid along the influent of the plug-flow reactor increases conversion although at the same time increasing any inhibition effects due to higher TAN concentrations (Julio et al., 2009; Watten and Sibrell, 2006). Therefore, it is both necessary and meaningful to test the applicability of ratio control in the continuous complete mixed moving bed biofilm reactor (MBBR). Undoubtedly, temperature is one of the main influencing factors for achievement and stable operation of nitrification. However, several studies have pointed out that the effect of temperature on nitrification in a biofilm reactor was limited

\* Corresponding author.

E-mail addresses: [jgljun@bjut.edu.cn](mailto:jgljun@bjut.edu.cn), [bjutlijun@126.com](mailto:bjutlijun@126.com) (J. Li).

## Nomenclature

MBBR	moving bed biofilm reactor	$r$	production rate
DO	dissolved oxygen	SOUR	specific oxygen uptake rate
TAN	total ammonia nitrogen	$D_{\text{NH}_4^+}$	diffusivity coefficients for ammonium
AOB	ammonia oxidizing bacteria	$D_{\text{O}_2}$	diffusivity coefficients for oxygen
NOB	nitrite oxidizing bacteria	$\gamma_{\text{O}_2/\text{NH}_4^+}$	ratio of the stoichiometric coefficients of ammonium and oxygen
FA	free ammonia	NRR	nitrogen removal rate
FNA	free nitrous acid	$\mu_{\text{max, AOB(NOB)}}^T$	maximum specific growth rate of AOB (NOB) at $T$ in $K$
HRT	hydraulic retention time	$T_0$	temperature at which $\mu_{\text{max, AOB(NOB)}}^T$ has been reported
TNN	total nitrite nitrogen	$E_{\text{a, AOB(NOB)}}$	free energy of AOB (NOB)
$C$	concentration	$R$	ideal gas constant
$T$	actual temperature		
NAR	nitrite accumulation rate		
OUR	oxygen uptake rate		

where competition for oxygen drives nitrification (Julio et al., 2009; Eduardo et al., 2015). Even so, nitrification with a temperature below 10 °C has not drawn much attention, especially in MBBR through ratio control. So far, most studies focus on nitrification as a treatment strategy of the sidestream (reject water) (Albert et al., 2010; Achlesh et al., 2013). Nitrification within the mainstream is comparatively difficult to achieve. Research on nitrification with a low-strength influent is beneficial for the application of nitrification. Considering economy and practicality, it is not recommended to elevate the temperature of the mainstream in a wastewater treatment plant. Therefore, an exploration of effective control methods to maintain the stable nitrification under low temperature for the treatment of the mainstream is crucial.

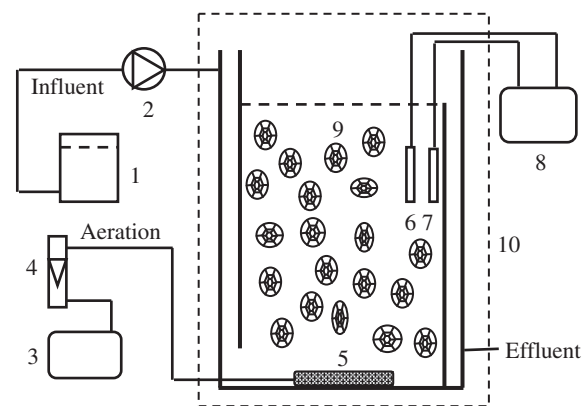
In addition, following the discovery of anaerobic ammonia oxidation (anammox) (Strous et al., 1999) and its later application, a two-step treatment consisting of nitrification plus anammox to autotrophically denitrify to nitrogen gas has probably become the most advanced and economical alternative for nitrogen removal (Henze et al., 2008; Van der Star et al., 2007). In a two-step system, nitrification is alternative with activated sludge, granular and biofilm reactors, and others. In summary, nitrification with granular and biofilm reactors has less effect on the subsequent anammox process compared to nitrification with activated sludge, i.e., no sludge settling is needed (Eduardo et al., 2015; Ma et al., 2011; Malovany et al., 2015).

In this study, a MBBR that utilized ratio control obtained nitrification of a low-strength ammonium concentration wastewater. Here, the main goals were to (i) decouple oxygen-limiting conditions from low DO concentrations, (ii) test the feasibility of ratio control, allowing the achievement and maintenance of nitrification in MBBR, and then explore the range of operating conditions, (iii) explore the nitrification performance of MBBR through ratio control at different temperatures, especially for temperature below 10 °C.

## 2. Materials and methods

### 2.1. Reactor set-up

A cubic reactor (about 180 × 120 × 260 mm) with working volume of about 4.4 L was used (see a detailed diagram in Fig. 1). The reactor was placed in a temperature control cabinet in which water temperature could be controlled according to requirements (26, 16, and 6 °C). Compressed air was supplied through an air diffuser placed at the bottom of the reactor. The DO concentration in the bulk liquid was measured via an online electrode. The aeration rate was manually controlled to maintain DO concentration at different set-point values. The hydraulic retention time (HRT) was adjusted via influent pump to obtain different setting TAN concentrations in



**Fig. 1.** Schematic diagram of the MBBR showing peripheral instrumentation: (1) Influent tank; (2) influent pump; (3) air pump; (4) air flowmeter; (5) air diffuser; (6) DO electrode; (7) pH electrode; (8) WTW; (9) pall ring; (10) temperature control cabinet. DO: dissolved oxygen.

the effluent. No pH adjustment was carried out in the reactor and in the effluent it ranged between approximately 7.3 and 7.4. Pall rings were used as biomass carriers. The volume of the carriers was approximately 22% of the working volume of the reactor. Inoculated activated sludge used for biofilm culturing was taken from the aeration tank of a wastewater treatment plant, which had a good nitrification performance, and its  $f$  (MLVSS/MLSS) and SVI were approximately 0.75 and 90, respectively.

### 2.2. Wastewater

The MBBR was fed with synthetic wastewater, which contained approximately 5 mg COD/L, added as  $\text{CH}_3\text{COONa}$ ; 2.5 mg P/L, added as  $\text{KH}_2\text{PO}_4$ ; 60 mg  $\text{NH}_4^+\text{-N/L}$ , added as  $\text{NH}_4\text{Cl}$ .  $\text{NaHCO}_3$  was added for alkalinity, which was approximately 720 mg/L. The pH of the influent ranged between 7.7 and 7.8. The synthetic wastewater contained 1 mL of trace elements solution per L of influent (Guerrero et al., 2011).

### 2.3. Analytical methods

All samples were filtered with a 0.45  $\mu\text{m}$  filter prior to analyzing.  $\text{NH}_4^+\text{-N}$ ,  $\text{NO}_2^-\text{-N}$ ,  $\text{NO}_3^-\text{-N}$ , MLSS, MLVSS, and alkalinity were measured according to standard methods (APHA, 1998). Biofilms were respectively detached from pall rings by ultrasound (45 kHz, 120 W, 2–3 min) and subsequently centrifuged (10000g, 15 min) for MLSS and MLVSS measurements. The detached pall rings were

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