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Dual-functional catalytic materials: Magnetically hollow porous Ni-manganese oxides microspheres/cotton cellulose fiber

Danfeng Zhang^{a,b}, Guozhen Zhang^a, Qiong Wang^a, Lei Zhang^{a,*}

- ^a College of Chemistry, Liaoning University, Shenyang 110036, China
- ^b College of Sciences, Heilongjiang Bayi Agricultural University, Heilongjiang 163319, China

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ABSTRACT

An eco-friendly magnetically hollow porous NiMn₂O₄ microspheres (HPS-NiMn₂O₄)/cotton cellulose fiber (CCF) composite (HPS-NiMn₂O₄/CCF) was designed by using regenerated cotton as the potential cotton cellulose fibers (CCF) source. Magnetically HPS-NiMn2O4 microspheres were in-situ grown on the CCF surface uniformly by using hydrothermal method with subsequent calcination for removing the cellulose matrix to obtain magnetic HPS-NiMn₂O₄/CCF. HPS-NiMn₂O₄/CCF exhibited dual-functional catalytic properties on reduction of 4-nitrophenol and photodegradation of dye pollutants compared with NiMn₂O₄ nanoparticles (NPs-NiMn₂O₄) and HPS-NiMn₂O₄. The mechanisms for the dual-functional catalytic reactions over HPS-NiMn₂O₄/CCF were inferred. The special morphology of HPS-NiMn₂O₄, the presence of NiMn₂O₄ with 'd⁸' (Ni element) and 'd⁵' (Mn element) electronic configurations and the introduction of CCF facilitated positively the catalytic reaction and efficient electron-hole separation. Because of its green and magnetic separability, HPS-NiMn₂O₄/CCF can be potentially applied in catalysis, water purification and green chemistry.

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1. Introduction

Nowadays, dual-functional metal oxides catalysts designed from renewable and eco-friendly resource have aroused much attention because it provided the promising sustainable development technologies. Due to their chemically and thermally stable magnetic properties [1-3], spinel type metal oxides $(AB_2O_4; A, B=Co, Ni,$ Zn, Mn, Fe) [4], combination of two simple low-cost transition metal oxides, were used for many applications, such as drug delivery, electronic devices, magnetic resonance imaging, and information storage [5-8]. Also, because of prominent electrochemical performance and easier to large-scale synthesis, they are emerging in electrode materials for both electrochemical capacitors and lithium ion batteries [9,10]. More applications of AB₂O₄ materials include adsorbents for removing toxic substances [11,12], treatment of heavy metal waste [13] and chemical sensors [14]. It is worth noting that spinel type AB2O4 containing 3d metal ion is developing prospects to be the most hopeful economical catalysts [15,16]. For example, Ali [17] studied the catalytic activity of NiFe₂O₄ for hydroxylation of benzene into phenol. Mathew [18] synthesized orthoalkyl phenols by using Cu_{0.5}Co_{0.5}Fe₂O₄ as

E-mail address: zhanglei63@126.com (L. Zhang).

the catalyst. Moreover, because of a narrow band gap, AB2O4 also has excellent photocatalytic performance [19,20]. AB₂O₄ may be used alone, or together with other photocatalysts as photocatalysts [21,22]. When used as composite photocatalysts, AB₂O₄ exhibited enhanced degradation efficiency [23,24]. As an important spinel binary metal oxide, NiMn2O4 has significant applications because of its magnetic and catalytic properties [2,25], such as magnetism [26], catalysis [27], super capacitor [28], sensors [29], etc.

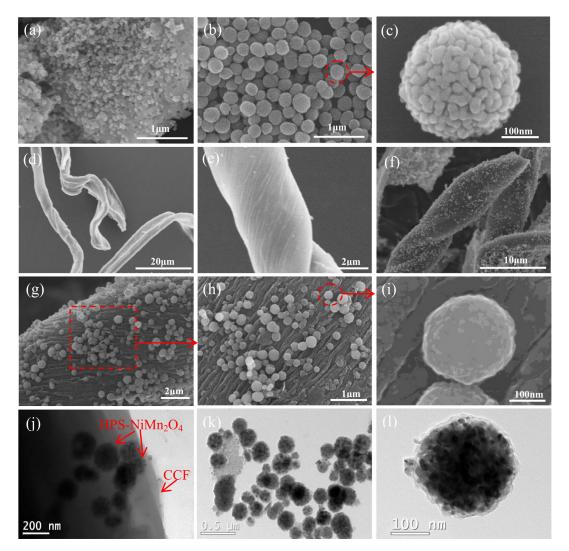
Along with the particle size decreasing and the increase of the surface energy for the catalyst, the interparticle aggregation will deteriorate and hinder seriously the catalytic activity. It is regarded a beneficial approach anchoring metal oxide particles on the substrates to prevent the interparticle aggregation and improve the catalytic efficiency. Recently, carbon-based materials, such as graphite [30], carbon black [31], carbon nanotubes [32,33], mesoporous carbon [34], graphene [35,36], and carbon spherules [37] have drawn increasing attention as catalyst carrier materials. Compared with these carbon-based materials, reproducible cotton as the potential cotton cellulose fibers (CCF) matrix are low cost, abundant, pollution-free, and renewable, also possess desired fibrous structure. Meanwhile, cotton cellulose fibers can endow the bare particles with more high active sites and offer a favorable platform to anchor the particles. The cotton cellulose fibers made of reproducible cotton has a great interest toward sustainable future and controlling the white pollution.

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Corresponding author.

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 $\textbf{Fig. 1.} \hspace{0.5cm} \textbf{SEM images of (a) NPS-NiMn}_2O_4, (b \ and \ c) \ HPS-NiMn}_2O_4/CCF, (l) \ monospheres \\ \textbf{HPS-NiMn}_2O_4/CCF, \ tEM \ images \ (j \ and \ k) \ of \ HPS-NiMn}_2O_4/CCF, (l) \ monospheres \\ \textbf{HPS-NiMn}_2O_4/CCF, \ with \ more \ details.}$



Scheme 1. Schematic illustration of the formation of HPS-NiMn $_2\mathrm{O}_4/\text{CCF}.$

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