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Sonochemical synthesis of SrMnO₃ nanoparticles as an efficient and new catalyst for O₂

evolution from water splitting reaction

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Abstract

The principal focus of this investigation is to prepare the SrMnO₃ nanostructures by different chemical

methods such as ultrasonic, co-precipitation, microwave, and hydrothermal methods. The influence of

calcination temperature, and ultrasound irradiation power, and the presence of surfactant investigated on

morphology and size of SrMnO₃ nanostructures. As-prepared nanoparticles were characterized by X-ray

diffraction (XRD), scanning electron microscope (SEM), transmission electron microscopy (TEM), Fourier

transform infrared spectroscopy (FT-IR), X-ray energy dispersive spectroscopy (EDS) and ultraviolet-visible

(UV-Vis) spectroscopy. The results indicated that by changing in method and reaction condition, product

appeared in different size, morphology, and uniformity. The morphology and size of nanostructures have been

influenced on the properties of nano-SrMnO₃. For investigation of properties, the SrMnO₃ was used in

catalytic water splitting for O₂ evolution in presence of (NH₄)₂Ce(NO₃)₆. The effect of nano-catalysts and the

concentration of (NH₄)₂Ce(NO₃)₆ have been studied on O₂ evolution reaction. Results show that the efficiency

of water splitting increased by enhancement in the size and uniformity of catalysts and introduced the SrMnO₃

as a new and efficient catalyst for O_2 evolution reaction.

Key words: Nanostructures; Chemical synthesis; Catalytic properties; Water splitting; O₂ evolution reaction.

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