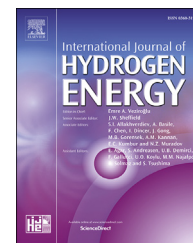


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Effect of different turbulence models on combustion and emission characteristics of hydrogen/air flames

Harun Yilmaz^a, Omer Cam^b, Selim Tangoz^b, Ilker Yilmaz^{b,*}

^a Department of Airframes and Powerplants, Civil Aviation College, Erzincan University, Erzincan, 24100, Turkey

^b Department of Airframes and Powerplants, Faculty of Aeronautics and Astronautics, Erciyes University, Kayseri, 38039, Turkey

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ABSTRACT

This paper aims to present modeling results of hydrogen/air combustion in a micro-cylindrical combustor. Modeling studies were carried out with different turbulence models to evaluate performance of these models in micro combustion simulations by using a commercially available computational fluid dynamics code. Turbulence models implemented in this study are Standard $k-\epsilon$, Renormalization Group $k-\epsilon$, Realizable $k-\epsilon$, and Reynolds Stress Transport. A three-dimensional micro combustor model was built to investigate impact of various turbulence models on combustion and emission behavior of studied hydrogen/air flames. Performance evaluation of these models was executed by examining combustor outer wall temperature distribution; combustor centerline temperature, velocity, pressure, species and NO_x profiles. Combustion reaction scheme with 9 species and 19 steps was modeled using Eddy Dissipation Concept model. Results obtained from this study were validated with published experimental data. Numerical results showed that two equation turbulence models give consistent simulation results with published experimental data by means of trend and value. Renormalization Group $k-\epsilon$ model was found to give consistent simulation results with experimental data, whereas Reynolds Stress Model was failed to predict detailed features of combustion process.

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Introduction

Hydrogen is the simplest and most abundant chemical element in existence and it can be utilized as an alternative fuel [1]. Using hydrogen as fuel offers many benefits which include a reduction in the pollution levels and a drop in greenhouse gas emissions [2]. It also reduces the dependency on crude oil. When hydrogen is used in its cleanest form, the

only product by combustion is water. This means exhaust gas emissions such as CO , CO_2 , and SO_x can be eliminated [1].

Considerable number of researchers investigated combustion and emission characteristics of H_2 /air flames both numerically and experimentally. Goulier et al. [3] built an experimental setup to investigate impact of turbulence on combustion behavior of H_2 /air mixtures. They found that turbulent flame speed is directly proportional to turbulence intensity but maximum combustion pressure is insensitive to

* Corresponding author. Fax: +90 352 437 5744.

E-mail address: iyilmaz@erciyes.edu.tr (I. Yilmaz).

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Nomenclature

ρ	Density
\vec{v}	Velocity vector
p	Static pressure
$\bar{\tau}$ and τ_{ij}	Stress tensor
$\rho \vec{g}$	Gravitational body force
\vec{F}	External body force
μ	Molecular viscosity
I	Unit tensor
h	Enthalpy of gas composition
F_{hj}	Energy flux of the x_j direction
u_j	Velocity component
λ_w	Thermal conductivity
R_0	Universal gas constant
M_i	Molar mass of species i
Y_i	Mass fraction of species i
R_i	Net production rate of species i via chemical reactions
\vec{J}_i	Diffusion flux of species i
Sc_t	Turbulent Schmidt number $\left(\frac{\mu_t}{\rho D_t}\right)$
μ_t	Turbulent viscosity
D_t	Turbulent diffusivity
$D_{i,m}$	Mass diffusion coefficient for species i
$D_{T,i}$	Thermal diffusion coefficient
τ^*	Time scale
C_e	Volume fraction constant (2.1377)
C_τ	Time scale constant (0.4082)
G_k	Generation of turbulence kinetic energy due to the mean velocity gradients
G_b	Generation of turbulence kinetic energy due to the buoyancy
Y_M	Contribution of the fluctuating dilatation in compressible turbulence to the overall dissipation rate
C_{1e}, C_{2e} and C_{3e}	Constants
σ_k and σ_ϵ	Turbulent Prandtl numbers for k and ϵ , respectively
α_k and α_ϵ	Inverse effective Prandtl numbers for k and ϵ (RNG model)
S_k and S_ϵ	User-defined source terms
μ_{eff}	Effective viscosity

turbulent intensity. Thinking that laminar burning velocities are very important for design and performance of combustion systems, Ilbas et al. [4] carried out an experimental study to measure laminar flame velocities of H_2 /air and H_2 /CH₄/air mixtures. They concluded that an increment in H_2 amount in the H_2 /CH₄ mixture increases resultant flame velocity and expands flammability limits. Jiang et al. [5] simulated hydrogen/air combustion using 3D DNS (Direct Numerical Simulation) method. Simulations were carried out in two different ways, with taking into consideration preferential diffusion and with unity Lewis number assumption. Comparisons between two different computational cases showed that preferential diffusion makes flow more turbulent. Hariharan et al. [6] performed an experimental study to understand effect of burner geometry (circular and elliptic) on performance of turbulent H_2 /air flames. They also made some

measurements to investigate flame stability characteristics and NO_x emissions. They reported that because liftoff velocity is lower for elliptic burner flames, elliptic burner flames are shorter and radiate less. They also reported that for both type burners, an increment in equivalence ratio decreases NO levels. Choudhuri and Gollahalli [7] investigated characteristics of H_2 /HC diffusion flames. It was founded that increasing hydrogen amount in fuel/oxidizer mixture increases reactivity hence combustion time and flame length reduce; an increment in hydrogen concentration rises NO and NO_x emission indices but decreases CO emission index and axial soot concentration. Hwang et al. [8] investigated impact of turbulence intensity of an air stream on characteristics of hydrogen/air jet flames. Results showed that increasing turbulence intensity shortens flame length. Skottene and Rian [9] simulated NO_x generation in H_2 /air flames. For this purpose, eight laminar flames were tested with 4 different reaction mechanisms and results obtained from this numerical study were validated with experimental data in literature. They reported that generation of NO through NNH radicals is dominant NO generation mechanism for such flames. To establish dynamics of H_2 /air flames at elevated temperatures and pressures, Kwon [10] numerically investigated laminar burning velocities of H_2 /air flames and effect of stretch on such flames at fuel lean equivalence ratios, high pressures (5–50 atm) and high temperatures (298–1000 K). They observed that flame stretch increases laminar burning velocity up to 28%.

Increasing demand for micro scale power sources is driven by recent developments in micro devices. Compared to conventional batteries, energy intensity of these power sources is at massive levels. Micro power sources exploit from conductive and radiative heat emanating from hydrogen or hydrocarbon combustion in a micro combustor. Aside from large or mesoscale combustion applications, the subject of using hydrogen as a fuel for micro combustors is investigated by many researchers [11].

Hua et al. [12] carried out simulation studies on premixed H_2 /air flames in a number of chambers by varying chamber dimensions and keeping aspect ratio at a constant value. They varied combustion chamber dimensions from millimeter to micron level to understand micro-combustion mechanism. They also analyzed impact of heat transfer conditions and found that the smaller the combustor size, the more effect on combustion characteristics. Li et al. [13] simulated premixed H_2 /air combustion in micro channels. They investigated respective effects of different physical and boundary conditions on flame temperature. Numerical results showed that flame temperature is directly proportional to combustor size and at a certain inlet velocity, a flame with least temperature can be maintained. Yang et al. [14] numerically analyzed effect of reduced pressures on combustion characteristics of fuel lean hydrogen/air flames in a micro combustor with cavity and concluded that combustion efficiency follows a non-monotonic trend depending on the pressure value. When pressure decreases from 1.0 atm to 0.8 atm; combustion efficiency increases but further decrement (from 0.8 to 0.5 atm) reduces combustion efficiency. At 0.8 atm, the amount of heat released from fuel increases with rising burning velocity. But at 0.5 atm, pressure decrement has a negative effect on burning rates, reaction intensity and temperature levels. Wan

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