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N,C,N-chelated Antimony(III), Bismuth(III) and Tin(IV) Derivatives of 1,1'-Ferrocenedicarboxylic Acid.

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Abstract

N,C,N-intramolecularly coordinated antimony(III) and bismuth(III) oxides $(ArMO)_2$ [where $Ar = [2,6\text{-}(Me_2NCH_2)_2C_6H_3]$], M = Sb (1) or Bi (2)] reacted with 1,1'-ferrocenedicarboxylic acid $fc(COOH)_2$ (fc = ferrocene-1,1'-diyl) under formation of corresponding carboxylates $fc(COO)_2SbAr$ (4) and $[fc(COO)_2BiAr]_2$ (5). Similarly, the treatment of $fc(COOH)_2$ with fc(IV) carbonate $fc(COO)_2Sn(Ph)Ar$ (6). All compounds were characterized by the help of multinuclear NMR, Raman, IR and fc(IV) spectroscopy and in the case of 4 and 5 using single-crystal X-ray diffraction analysis.

The utilization of carboxylates as ligands in main group chemistry is unexceptional. Carboxylates derived from ferrocene constitute a special class of compounds. Especially the easily available 1,1'-ferrocenedicarboxylic acid provides a rich spectrum of coordination modes, thus being a promising ligand and(or) template for building of various types of carboxylates.[1] This field of chemistry is quite well studied in the case of organotin(IV) compounds, where both 1,1'-ferrocenedicarboxylic and also various mono-functionalized ferrocenecarboxylic acids were applied.[2] Surprisingly, very limited number of related structurally characterized carboxylates is known for the antimony and bismuth. In fact, the structure of only two antimony(V) compounds, i.e. Fc(COO)SbPh₄ and [Fc(COO)]₂Sb(ptolyl)₃, (Fc = ferrocenyl) were reported by Liu et al.[3] Only recently, Chandrasekhar used both ferrocene mono- and 1,1'-dicarboxylic acid in the reaction with Ph₃Bi leading to the formation of interesting coordination polymers.[4] Above mentioned facts inspired us to enrich the chemistry of heavier pnictogen ferrocene-based carboxylates by using the 1,1'ferrocenedicarboxylic acid as a reagent for our antimony(III) [5] and bismuth(III) [6] oxides $(ArMO)_2$ [where $Ar = [2,6-(Me_2NCH_2)_2C_6H_3]^T$, M = Sb (1) or Bi (2)]. These oxides showed an interesting reactivity toward acidic oxides (such as SeO₂, CO₂, SO₂, As₂O₃ etc.) [7] and

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