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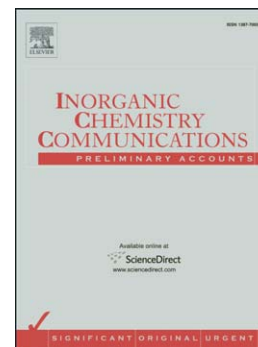
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A cyclopalladated phosphine selenide with an anionic acylselenourea ligand

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^b Chemische Kristallographie, Max-Planck-Institut für Kohlenforschung, 45470 Mülheim an der Ruhr, Germany**Keywords:** Selenium; Palladium; Cyclometallation; Phosphine selenide; Selenourea; X-ray structure**Abstract**

The chloro-bridged cyclometallated palladium(II) complex $[\text{PdCl}\{\{\kappa\text{C},\text{Se}-\text{C}_6\text{H}_4\text{P}(\text{Se})\text{Ph}_2\}\}_2]$ reacts with 4-MeC₆H₄C(O)NHC(Se)NEt₂ in the presence of base to give $[\text{Pd}\{\{\kappa\text{C},\text{Se}-\text{C}_6\text{H}_4\text{P}(\text{Se})\text{Ph}_2\}\}\{\{\kappa\text{O},\text{Se}-4\text{-MeC}_6\text{H}_4\text{C}(\text{O})\text{NC}(\text{Se})\text{NEt}_2\}\}]$ in high yield. The complex was fully characterised by spectroscopic methods and also X-ray diffraction. This compound represents one of the few examples of a metal complex containing two different selenium-containing ligands.

Triphenylphosphine selenide (Ph₃P=Se) typically acts as a neutral, monodentate Se-donor ligand in complexes with transition metals including palladium, platinum and gold. A number of such compounds have been reported and several have also been structurally characterised. Examples include $[\text{MX}_2(\text{Ph}_3\text{PSe})_2]$ (M = Pt, Pd; X = Cl, Br, SCN), $[\text{AuX}(\text{Ph}_3\text{PSe})]$ (X = Cl, Br, I, CN) and also $[\text{Au}(\text{Ph}_3\text{PSe})_2]\text{SbF}_6$ [1-9]. In several cases, cyclometallation can occur, forming compounds in which Ph₃PSe acts as a monoanionic C,Se-chelating ligand. To date cyclometallated triphenylphosphine selenide derivatives are known for Mn, Au, Pt and Pd [10-12]. It is worth noting, that while triarylphosphines can undergo cyclometallation reactions directly [13], triphenylphosphine selenide has to be converted into the Hg- Li- or Sn-derivatives to transmetallate the $[\{\kappa\text{C},\text{Se}-\text{C}_6\text{H}_4\text{P}(\text{Se})\text{Ph}_2\}]^-$ group. We have previously investigated the chemistry and applications of heteroleptic metal complexes containing acylchalcogenourea ligands, which may either act as chelating O,Se⁻ or monodentate Se⁻ ligands [14-21]. Given that there are very few examples of metal complexes containing two different selenium donor centres, we thought the combination of a cyclometallated phosphine selenide with an acylselenourea might allow us to isolate such a compound. The reaction of the chloro-bridged palladium(II) dimer $[\text{PdCl}\{\{\kappa\text{C},\text{Se}-\text{C}_6\text{H}_4\text{P}(\text{Se})\text{Ph}_2\}\}_2]$ with 4-MeC₆H₄C(O)NHC(Se)NEt₂ in the presence of base afforded a yellow-orange compound (**1**) in high yield (Scheme 1) [22].

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