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# Dual Functional Three-Dimensional LnMOFs for Luminescence Sensing of Nitrobenzene and Fe<sup>3+</sup> Ions

Xiaohai Guo<sup>a</sup>, Yongsheng Li<sup>a</sup>, Qingyuan Peng<sup>a</sup>, Zhiming Duan<sup>a</sup>, Mingxing Li<sup>a</sup>, Min Shao<sup>d</sup>, Xiang He<sup>a,b,c\*</sup>

<sup>a</sup> Department of Chemistry, Innovative Drug Research Center, Shanghai University, Shanghai 200444, China

<sup>b</sup> State Key Laboratory of Rare Earth Resource Utilization, Changchun Institute of Applied Chemistry, Chinese Academy of Sciences, 5625 Renmin Street, Changchun, Jilin 130022, P.R. China

<sup>c</sup> State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China

<sup>d</sup> Laboratory for Microstructures, Shanghai University, Shanghai 200444, PR China

## Abstract:

Two lanthanide MOFs, namely  $\{\{\text{Me}_2\text{NH}_2\}[\text{Tb}^{\text{III}}(\text{L})]\cdot 3\text{H}_2\text{O}\cdot \text{DMF}\}_n$  (**1**) and  $\{\{\text{Me}_2\text{NH}_2\}[\text{Eu}^{\text{III}}(\text{L})]\cdot 3\text{H}_2\text{O}\cdot \text{DMF}\}_n$  (**2**), based on the flexible ligand 5-(bis(4-carboxybenzyl)amino)isophthalic acid (H<sub>4</sub>L) have been synthesized under solvothermal conditions, where DMF = N,N-dimethylformamide. The two compounds exhibit a 3D structure with rectangle channels encapsulating  $[\text{Me}_2\text{NH}_2]^+$  cations. The microporous structure as well as the confinement of cations inside the MOFs pores allows **1** and **2** to act as dual-functional luminescent materials to detect nitroaromatics and metal ions. The experiments show that the luminescence of sample **1** in DMF solution was quenched when the nitrobenzene (NB) content reached  $2.3 \times 10^{-2}$  mol/L and concentration of Fe<sup>3+</sup> reached  $5 \times 10^{-4}$  mol/L.

Keywords: Lanthanide MOFs; Sensor; Crystal structure; Luminescence

## 1 Introduction

Metal-organic frameworks (MOFs) have been developed quickly because of their potential applications in gas storage and separation [1], ion exchange [2], dye adsorption and desorption [3], catalysis [4] and so on. Compared with transition metal-based MOFs, lanthanide MOFs (Ln-MOFs) have more prominent fluorescence properties for sensing due to their unique optical properties [5-7], which are derived from f–f transitions, that are known as the “antenna effect” [8-10]. So far, various Ln-MOFs have been prepared for detecting pesticides [11-13], explosives [14-18], cations [19-24], anions [25-28] and volatile organic compounds [29,30].

It is well known that iron is one of the indispensable elements for life, but the existence of a

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