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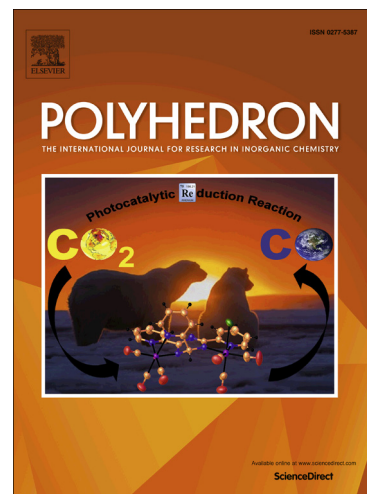
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**Syntheses, crystal structures and luminescent properties of three metal
coordination polymers based on aromatic carboxylic acids and
2-(pyridine-4-yl)-(1*H*)-benzoimidazole**

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Abstract

Three new metal coordination complexes, namely [Co(**IPA**)(**4PBI**)·H₂O] (**1**), [Cd(**IPA**)(**4PBI**)(H₂O)] (**2**) and [Cd(**TPA**)(**4PBI**)(H₂O)] (**3**) [**4PBI** = 2-(pyridin-4-yl)-(1*H*)-benzoimidazole], **IPA** = isophthalate and **TPA** = terephthalate], were obtained under hydrothermal conditions by reacting cobalt and cadmium salts with **4PBI** and **IPA** or **TPA**. Complexes **1-3** were structurally characterized by X-ray single-crystal diffraction, EA, IR and PXRD. All of these complexes display low dimensional features with one-dimensional (1D) motifs. Complex **1** is a 1D tape-like structure containing bi-nuclear units, which is further extended into a 3D supramolecular architecture through intermolecular hydrogen bonds and stacking interactions. Complex **2** shows a 1D structural motif, which is further assembled into a 3D supramolecular architecture by hydrogen bonds and packing interactions. Complex **3** is also a 1D chain. Through intermolecular hydrogen bonds and $\pi\cdots\pi$ packing interactions, a 3D supramolecular network is also generated for **3**. The luminescent properties of complexes **2** and **3** were investigated in the solid state at room temperature. Additionally, thermogravimetric analysis (TGA) of **1-3** has been discussed in details.

Keywords: hydrothermal reaction; 2-(pyridin-4-yl)-(1*H*)-benzoimidazole; aromatic carboxylic acid; luminescence; crystal structure

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