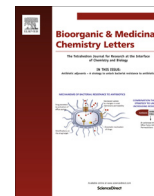




Contents lists available at ScienceDirect

Bioorganic & Medicinal Chemistry Letters

journal homepage: www.elsevier.com/locate/bmcl

Synthesis and biological evaluation of pyridinium-functionalized carbazole derivatives as promising antibacterial agents



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ARTICLE INFO

Article history:

Received 5 June 2017

Revised 15 August 2017

Accepted 17 August 2017

Available online 19 August 2017

Keywords:

Pyridinium

Carbazole

Synthesis

Optimization

Antibacterial

ABSTRACT

Various pyridinium-functionalized carbazole derivatives were constructed by coupling the key fragments of carbazole skeleton and pyridinium nucleus in a single molecular architecture. Antibacterial bioassays revealed that some of the title compounds displayed impressive bioactivities against plant pathogens such as *Xanthomonas oryzae* pv. *oryzae*, *Ralstonia solanacearum*, and *Xanthomonas axonopodis* pv. *citri* with minimal EC₅₀ values of up to 0.4, 0.3, and 0.3 mg/L, respectively. These bioactivities were achieved by systematically tuning and optimizing bridging linker, alkyl length of the tail, and substituents on the carbazole scaffold. Compared with the bioactivity of the lead compound (**AP-10**), antibacterial efficacy dramatically increased by approximately 13-, 104- and 21-fold. This finding suggested that these compounds can serve as new lead compounds in research on antibacterial chemotherapy.

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Disease-causing bacteria are gaining considerable attention over the last decade due to the significant threats they impose on agricultural products and their remarkable ability in acquiring additional resistance mechanisms.^{1–3} *Xanthomonas oryzae* pv. *oryzae* (*Xoo*), *Ralstonia solanacearum* (*R. solanacearum*) and *Xanthomonas axonopodis* pv. *citri* (*Xac*) are three widely distributed Gram-negative opportunistic pathogens that can infect an array of individual species including rice, tomato, potato, tobacco and citrus.^{4–6} Infection by these bacteria can present necrotic lesions on leaves, stems and/or fruits, which consequently result in a serious loss in agricultural output.^{7,8} In addition, the emergence and worldwide spread of multidrug resistant pathogens has exacerbated the management of these persistent plant bacterial diseases. Although a few of the commercial drugs have been used to combat these diseases such as bismethiazol (**BT**) and thiodiazole copper (**TC**), they failed to effectively treat the infected plants under field conditions considering their poor efficiency, high phytotoxicity and/or bacterial resistance.^{9,10} Therefore, exploring and developing new anti-bacterial drugs is imperative with novel chemical motifs preferably owning unique modes of action rather than analogues of the existing ones.

Carbazole skeleton, owning impressive electronic and charge-transport properties, is a crucial type of nitrogen-containing aromatic heterocyclic scaffold, present in many naturally occurring products and biologically active substances.^{11,12} In addition, this privileged building block can be easily tuned and modified with various functional groups, which endow carbazole-based derivatives and analogues with an admirable array of pharmacological activities such as anticancer, anti-inflammatory, antiviral, antifungal and antioxidant activities.^{13–16} In particular, the antimicrobial activity has been widely investigated for their extensively potential applications in the pharmaceutical industry.^{17,18} For example, Bremner and co-workers reported several carbazole-linked cyclic and acyclic peptoids as growth inhibitors of *Staphylococcus aureus* with a minimum inhibitory concentration (MIC) of 15 µg/mL.¹⁹ Gu and co-workers had synthesized and evaluated the antimicrobial activity of a series of new carbazole derivatives of ursolic acid and found that some compounds exhibited significant antibacterial activities against both Gram-positive and Gram-negative bacteria with MIC values ranging from 3.9 µg/mL to 15.6 µg/mL.²⁰ Thus, the fusion of a carbazole motif with the target molecule might probably lead to improved biological activity owing to the synergistic effect of these valuable moieties.

As an important functional fragment, pyridinium nucleus exists extensively in various kinds of pharmaceutical agents.^{21,22} Normally, compounds featuring this scaffold always acquire various physicochemical properties and subsequently resulted in reformed

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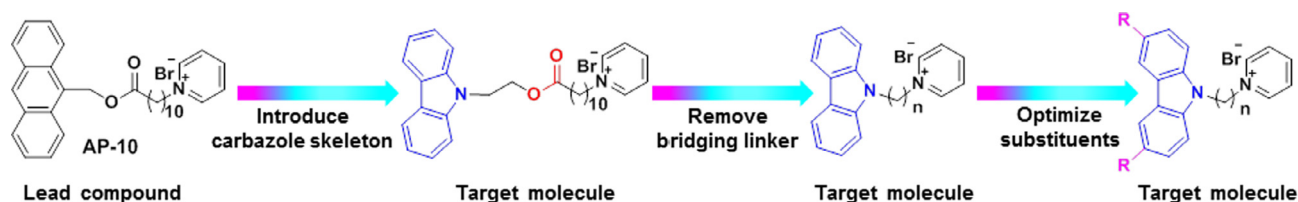
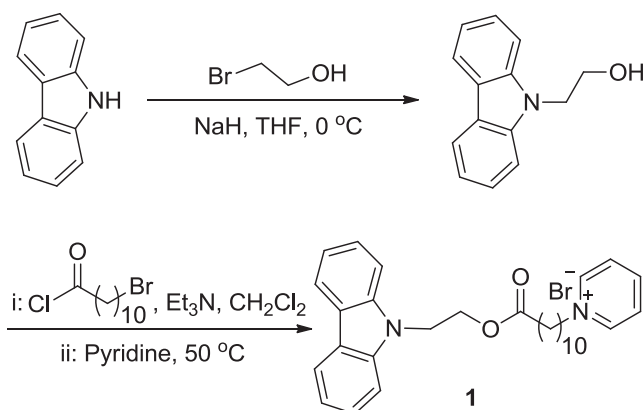


Fig. 1. Design strategy for the target compounds.



Scheme 1. Synthesis of compound 1.

or enhanced biological activities, because the positive charge can strengthen their specificity for the target species.²³ Moreover, pyridinium-tailored amphiphiles are considered one of the powerful molecular templates to create biologically active compounds, especially in the field of antimicrobial drug discovery.^{24,25} For example, Eren et al. investigated the antibacterial activity of some pyridinium functionalized polynorbornenes and found that compounds bearing octyl tails on the pyridine rings showed potent inhibitory effects against *Escherichia coli* and *Bacillus subtilis*.²⁶ Apparently, these repeatedly numerous studies on this functional scaffold opened a new avenue for the discovery and development of novel high-efficient bioactive molecules.

In our previous work, 1-[11-(9-anthracenyl methoxy)-11-oxoundecyl] pyridinium bromide (**AP-10**) exhibited good antibacterial activities towards plant pathogens.²⁷ Encouraged by the aforementioned facts and in a continuing search for high-efficient antibacterial agents, we report herein the design and synthesis of a series of pyridinium-tailored carbazole derivatives by coupling key fragments of carbazole skeleton and pyridinium nucleus in a single molecular architecture (Fig. 1). The carbazole group, which

has desirable electron characteristics, is probably responsible for better binding with the enzymes or the receptors of bacteria; pyridinium, which has a positive charge, is employed to possibly interact with anionic cell components and increase the water solubility as well as membrane permeability; bridging linker and alkyl chain length of the tail are used to tune the balance of hydrophobicity/hydrophilicity of target compounds. Better bioactive structures would be fabricated by utilizing the synergistic effect of these privileged moieties. The antibacterial activities of all the title compounds were tested against plant pathogens such as *Xoo*, *R. solanacearum* and *Xac*.

To investigate the antibacterial effect after replacing the anthracene ring of **AP-10** into carbazole skeleton, compound **1** was firstly designed and synthesized. As indicated in Scheme 1, intermediate 2-(9H-carbazol-9-yl)ethanol was obtained by adding 2-bromoethanol into a mixture of carbazole and NaH in dry THF, followed by treating it by two-step consecutive reactions with 11-bromoundecyl chloride and pyridine to provide the title molecule **1**. The structure was confirmed by ¹H NMR, ¹³C NMR and MS experiments (detailed information see Supplementary data). The antibacterial bioassays against *Xoo*, *R. solanacearum* and *Xac* were performed as previously described,^{27,28} and the commercial antibacterial agents (**BT** and **TC**) were co-assayed as positive controls under similar conditions. The inhibitory effect of **1** was indeed improved by introducing carbazole moiety in the target molecule (Table 1). In particular, anti-*R. solanacearum* activity was significantly increased with EC₅₀ values from 31.3 mg/L to 1.6 mg/L, approximately 19-fold enhancement in the antibacterial efficacy, which validated the author's assumption.

Further, we wonder if the removed ester group within the alkyl chain would reform the bioactive efficacy, because compound **1** exhibited good antimicrobial potentials. Thereby, compound **2** was primarily constructed by sequential substitution reactions of carbazole with 1,12-dibromododecane to produce bromide-tailored carbazole, which was then treated with pyridine at 50 °C (Scheme 2). The structure was also characterized by ¹H NMR, ¹³C NMR and MS experiments (detailed information see Supplementary data). The bioassay result is shown in Table 1. A dramatically

Table 1
Antibacterial activities of target compounds **1–5** against plant pathogen *Xoo*, *R. solanacearum* and *Xac* in vitro.

No.	<i>Xoo</i>			<i>R. solanacearum</i>			<i>Xac</i>		
	Regression equation ^a	r	EC ₅₀ (mg/L)	Regression equation	r	EC ₅₀ (mg/L)	Regression equation	r	EC ₅₀ (mg/L)
AP-10	y = 3.29x + 2.60	0.93	5.3 ± 0.7	y = 1.01x + 3.50	0.97	31.3 ± 8.2	y = 2.74x + 2.80	0.96	6.3 ± 0.5
1	y = 3.54x + 3.36	0.98	2.9 ± 0.3	y = 12.73x + 2.39	0.98	1.6 ± 0.2	y = 12.87x - 2.71	0.94	4.0 ± 0.2
2	y = 16.27x + 10.85	0.98	0.4 ± 0.1	y = 16.91x + 15.26	0.96	0.3 ± 0.1	y = 3.59x + 6.60	0.95	0.3 ± 0.1
3	y = 9.67x + 5.07	0.98	1.0 ± 0.1	y = 9.55x + 8.89	1.00	0.4 ± 0.1	y = 2.03x + 4.57	1.00	1.6 ± 0.4
4	y = 14.16x - 2.52	0.96	3.4 ± 0.1	y = 4.73x + 4.89	0.98	1.1 ± 0.1	y = 5.18x + 4.38	0.95	1.3 ± 0.1
5	y = 3.91x + 0.26	0.95	16.4 ± 2.4	y = 3.85x + 2.07	0.99	5.8 ± 0.5	y = 2.12x + 4.24	1.00	2.3 ± 0.2
BT	y = 1.50x + 2.05	0.98	92.6 ± 2.2	/	/	/	/	/	/
TC	/	/	/	y = 1.03x + 2.94	0.99	99.1 ± 5.1	y = 2.15x + 0.94	0.98	77.0 ± 2.0

^a Five different concentrations (such as 80, 40, 20, 10, 5 mg/L, depending on the bioactivity of different compounds, the concentrations were chosen in two times decline trend to make sure the EC₅₀ values are inside the concentration ranges tested) of the test compounds and positive control were selected to test the corresponding inhibition rates. By using the SPSS 17.0 software and the obtained inhibition rates at different concentrations, a related regression equation was provided to calculate the related EC₅₀ values.^{10,27,28}

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