

Discovery of pyrazolo[1,5-*a*]pyrimidine-3-carbonitrile derivatives as a new class of histone lysine demethylase 4D (KDM4D) inhibitors



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ABSTRACT

Herein we report the discovery of a series of new small molecule inhibitors of histone lysine demethylase 4D (KDM4D). Molecular docking was first performed to screen for new KDM4D inhibitors from various chemical databases. Two hit compounds were retrieved. Further structural optimization and structure-activity relationship (SAR) analysis were carried out to the more selective one, compound **2**, which led to the discovery of several new KDM4D inhibitors. Among them, compound **10r** is the most potent one with an IC₅₀ value of 0.41 ± 0.03 μM against KDM4D. Overall, compound **10r** could be taken as a good lead compound for further studies.

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Histone lysine demethylase 4D (KDM4D), also known as JMJD2D, is a typical jumonji domain-containing demethylase, which can catalyze the removal of a methyl group from lysine 9 on histone 3 (H3K9) or H1.4K26.^{1–4} Unlike other widely studied histone demethylase, the studies of KDM4D are just getting started.^{5,6} Despite that biochemical functions of KDM4D have been known, its role in pathogenesis is far from clear,^{7,8} and often opposite in different studies. For examples, a recent study by Bur et al. showed that KDM4D overexpression might lead to a higher risk for resistance to radiotherapy in Hodgkin lymphomas.⁹ Kim et al. revealed that KDM4D was required for colon cancer cell proliferation and survival. However, evidence also showed that KDM4D was able to stimulate p53-dependent gene expression, implying a tumor suppressor gene.¹⁰ The research revealing the net effect, if any, of KDM4D in tumorigenesis is complicated, and not the intention of this investigation. We know that small molecule inhibitors of KDM4D, particularly those with good selectivity, could be used as a tool to help researchers to unravel the action and mechanism.¹¹ Nevertheless, very few KDM4D inhibitors have been reported by now.^{12–15} Our purpose here is to discover potent and

selective KDM4D inhibitors. We hope the compounds discovered could enrich the arsenal of KDM4D inhibitors, and facilitate bio-functional and disease treatment studies related to KDM4D in the further.

To retrieve new KDM4D inhibitors, we firstly performed molecular docking-based virtual screening against various chemical libraries. All the calculations were carried out using the platform of Discovery Studio 3.1 (Accelrys Inc., San Diego, CA, USA). The receptor structure for molecular docking was prepared from the crystal structure of jmjC domain of KDM4D (PDB entry: 5FP4). The binding site was defined as a sphere containing the residues that stay within 8 Å from its original ligand, which is large enough to cover the catalytic site. The Charmm force field was assigned. Chemical libraries used in the virtual screening include SKLB (compounds synthesized by our laboratory), Specs (Specs, Inc. Zoetermeer, The Netherlands), and ChemDiv (ChemDiv, Inc., San Diego, CA, USA). All the docking calculations were carried out with the GOLD program.¹⁶ Goldscore¹⁷ was chosen as the scoring function. From the top ranking compounds, ID-Score¹⁸ and RASA¹⁹ were further used to rank the compounds and estimate their synthetic accessibility. Thirty compounds were finally selected for bioactivity validation. Two compounds (**1** and **2**, see Fig. 1) showed inhibition rate more than 50% against KDM4D at the concentration of 10 μM (see Table 1). We also measured the activities of the two hit compounds against other three KDMs, namely KDM2B, KDM3B,

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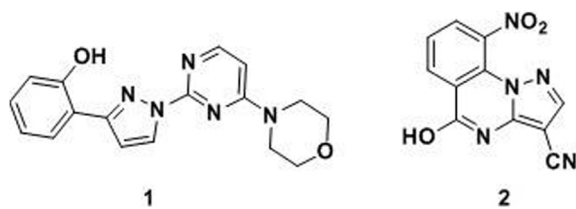


Fig. 1. Chemical structures of the hit compounds.

Table 1
Bioactivities of compounds 1 and 2.

Compound	Inhibition rate @10 μ M ^a			
	KDM4D	KDM2B	KDM3B	KDM5A
1	99.67 \pm 3.96	84.12 \pm 8.21	90.65 \pm 9.16	73.62 \pm 5.83
2	73.33 \pm 19.14	-4.85 \pm 4.72	-2.18 \pm 3.19	23.09 \pm 7.51

^a All assays were conducted in triplicate.

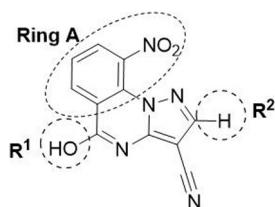


Fig. 2. Schematic showing the focus of structural modifications.

and KDM5A to examine the selectivity roughly; here just the three KDMs are selected because they are available for us at the moment. It was found that only compound 2 showed a better selectivity against KDM4D.

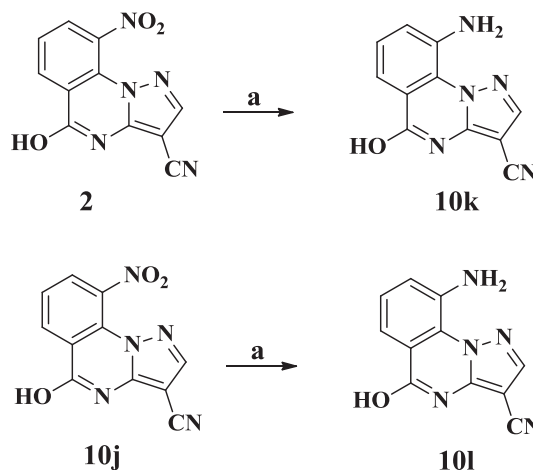
We then carried out a structural optimization to the selective compound, 2. The structural modifications were focused on three regions (see Fig. 2): ring A (nitrobenzene), R¹ (the hydroxyl group), and R² (the hydrogen atom). Various substituents or groups were used to replace the ring A, R¹, and R², and a total of twenty compounds (10a–t) were synthesized.

The synthetic routes for compound 10a–j are depicted in Scheme 1. Acid chloride intermediates 4a–j were first prepared through chlorination of commercially available reagents (3a–j) by thionyl chloride or oxalyl chloride. Nucleophilic substitution reaction between 4d or 4e and 3-amino-1H-pyrazole-4-carbonitrile (6) gave target compounds 10d and 10e. Reaction of 4a–c or 4f–j and 6 produced amide derivatives 5a–c and 5f–j, which then went through a ring close reaction to give target compounds 10a–c and 10f–j. Compounds 10k and 10l were prepared with 2 and 10j, respectively, by Pd/C-catalyzed reduction reactions (see Scheme 2). Target compounds 10m–o were synthesized through very similar reaction routes as those of compounds 10d, 10e (see Scheme 3).

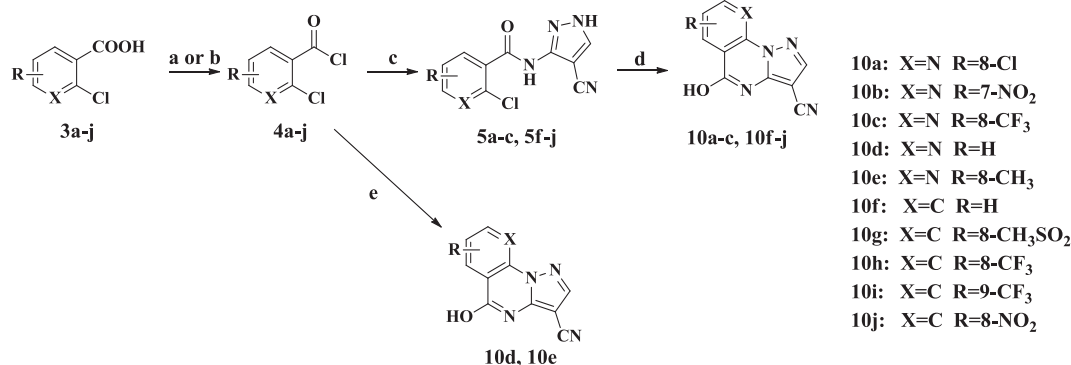
Compounds 10p and 10q were prepared through copper-catalyzed tandem reactions of 2-chloronicotinaldehyde (3p) or 1-(2-chloropyridin-3-yl)ethanone (3q) and 3-amino-1H-pyrazole-4-carbonitrile (6) (see Scheme 4).²⁰

Reaction routes for compounds 10r–t are shown in Scheme 5. Commercially available reagent malononitrile (7) reacted with triethyl orthoacetate derivatives through a Knoevenagel reaction to give intermediate 8a–c, which then reacted with hydrazine hydrate to produce intermediates 9a–c.²¹ Treatment of 4a with 9a–c provided 2-chloro-N-(4-cyano-1H-pyrazol-3-yl)nicotinamide derivatives 5r–t. A ring close reaction of 5r–t was carried out to afford the desired compounds 10r–t.

All the synthesized compounds were first tested for their inhibitory rates against KDM4D at the fixed concentration of 10 μ M. Bioactivities for all the compounds are summarized in Table 2. From Table 2, we can see that there are three compounds (10p, 10r, 10s) that showed more potent activities than compound 2. For these compounds whose inhibitory rates are higher than



Scheme 2. Synthetic routes for compounds 10k and 10l. Reagents and conditions: (a) Pd/C, methanol, H₂O, 60 °C.



Scheme 1. Synthetic routes for compounds 10a–j. Reagents and conditions: (a) thionyl chloride, DMF, 80 °C; (b) oxalyl chloride, DMF, DCM, 0 °C-rt; (c) 3-amino-1H-pyrazole-4-carbonitrile (6), THF, 0 °C-rt; (d) K₂CO₃, DMF, Argon, 90 °C; (e) 3-amino-1H-pyrazole-4-carbonitrile (6), K₂CO₃, DMF, Argon, 0 °C-rt-140 °C.

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