



# Physakengoses K-Q, seven new sucrose esters from *Physalis alkekengi* var. *franchetii*



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## ABSTRACT

Seven sucrose esters, physakengoses K-Q (1–7) were isolated from the aerial parts of *Physalis alkekengi* var. *franchetii*. Their structures were elucidated on the basis of extensive spectroscopic analyses and chemical methods. These new compounds were tested for their antimicrobial abilities against *Staphylococcus aureus*, *Bacillus subtilis*, *Pseudomonas aeruginosa* and *Escherichia coli*. Among the isolated sucrose esters, compounds 1–5 showed potent antibacterial activity with MIC values ranging from 2.16 to 12.76 µg/mL.

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## 1. Introduction

*Physalis alkekengi* var. *franchetii* (Solanaceae) (Chinese name: “Jindenglong”) [1], is widely distributed and cultivated in Europe and Asia [2]. The calyxes of *P. alkekengi* var. *franchetii* have been used as a traditional Chinese medicine for treatment of sore throat, cough, eczema, hepatitis, urinary problems and tumors [3]. Sucrose esters, structurally featured in sucrose and long fatty acid ester [4,5], have captured the attentions of many researchers due to their potent antibacterial and anti-inflammatory activities in recent years [6–8]. In continuing phytochemical studies of *P. alkekengi* var. *franchetii*, we have previously reported the isolation and structural determination of ten new sucrose esters physakengoses A–J [9], and most of them displayed potent antibacterial activities. This prompted us to further search for bioactive sucrose esters. As a result, 7 new sucrose esters, named physakengoses K–Q, were isolated from the aerial parts of *P. alkekengi* var. *franchetii* (Fig. 1). In this paper, we report the isolation, structure elucidation and antibacterial activity of the isolated sucrose esters from the aerial parts of *P. alkekengi* var. *franchetii*.

## 2. Results and discussion

### 2.1. Structural elucidation

Physakengose K (1, C<sub>38</sub>H<sub>64</sub>O<sub>15</sub>) was obtained as an amorphous solid. The NMR data of 1 (Table 1) revealed that it contained signals for sucrose and long chain fatty acid ester moieties [10,11]. The presence of a sucrose unit was deduced from the analysis of the NMR spectra (Table 1), which showed the anomeric CH signals of the glucopyranose (δ<sub>H</sub> 5.56, d, *J* = 3.5 Hz, H-1; δ<sub>C</sub> 91.3, C-1) and that of the anomeric carbon of the fructofuranose (δ<sub>C</sub> 103.4, C-2'). Alkaline hydrolysis also confirmed the existence of sucrose. Analysis of its NMR spectra (<sup>1</sup>H, <sup>13</sup>C, HSQC and HMBC) and the comparison of NMR data between compound 1 and physakengose G [9] allowed the identification of the acyl groups as myristyl, tigloyl, 3-methylbutanoyl and acetyl. The positions of these groups were established by the HMBC correlations from H-3' (δ<sub>H</sub> 5.31) to C-1 (δ<sub>C</sub> 175.1) of myristyl group, from H-2 (δ<sub>H</sub> 4.86) to C-1 (δ<sub>C</sub> 173.7) of 3-methylbutanoyl group, from H-3 (δ<sub>H</sub> 5.41) to C-1 (δ<sub>C</sub> 168.6) of tigloyl group and from H<sub>2</sub>-1' (δ<sub>H</sub> 4.01, 4.09) to acetoxy carbonyl (δ<sub>C</sub> 172.0) (Fig. 2). Thus, the structure of compound 1 was assigned as 1'-O-acetyl-2-O-(3-methylbutanoyl)-3'-O-myristyl-3-O-tigloylsucrose.

The NMR data (Table 1) of physakengose L (2, C<sub>35</sub>H<sub>58</sub>O<sub>15</sub>) and physakengose M (3, C<sub>35</sub>H<sub>58</sub>O<sub>15</sub>) also showed characteristic signals for sucrose and fatty acid ester. A comparison of the spectroscopic data of 2 with those of physakengose B [9] revealed an overall similarity except for the presence of signals attributable to an acetyl

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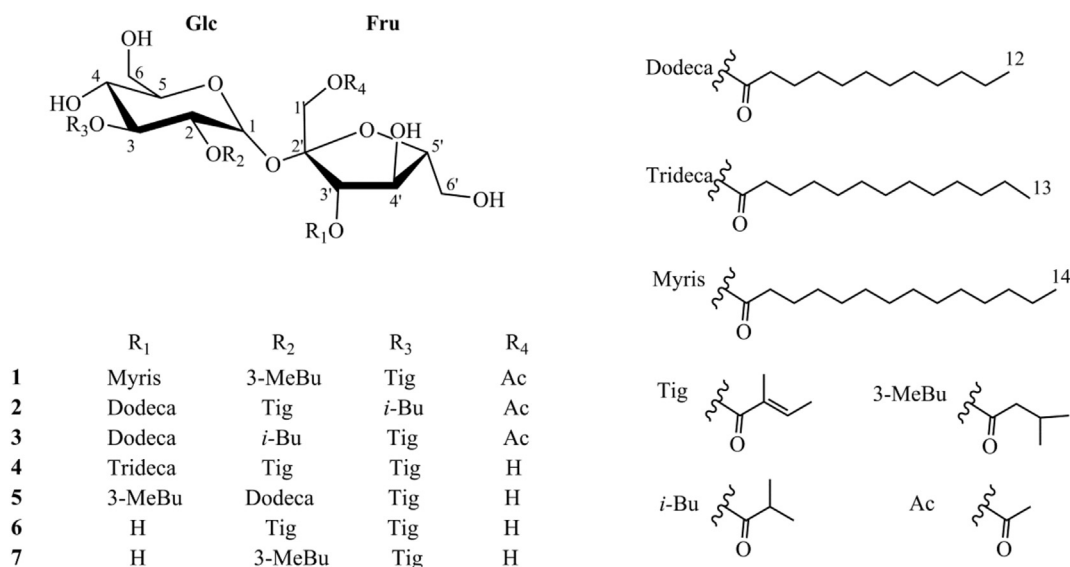


Fig. 1. Structures of compounds 1–7.

unit [ $\delta_{\text{H}}$  2.04 (3H, s);  $\delta_{\text{C}}$  171.8, 20.6], and the downfield shift of H<sub>2</sub>–1' ( $\delta_{\text{H}}$  3.92, 4.04) relative to those in physakengose B ( $\delta_{\text{H}}$  3.32, 3.46). The location of the acetyl unit was determined by analysis of HMBC spectrum. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR data (Table 1) of compound 3 closely resemble those of 2. The differences between them were determined by the HMBC spectrum; the correlations from H-2 ( $\delta_{\text{H}}$  4.84) to the carbonyl carbon of isobutyryl unit ( $\delta_{\text{C}}$  177.7) and from H-3 ( $\delta_{\text{H}}$  5.42) to the carbonyl carbon of tigloyl unit ( $\delta_{\text{C}}$  168.7) suggested that the isobutyryl and tigloyl units were attached to C-2 and C-3, respectively. Accordingly, the structures of compounds 2 and 3 were elucidated as 1'-O-acetyl-3'-O-dodecanoyl-3-O-isobutyryl-2-O-tigloylsucrose and 1'-O-acetyl-3'-O-dodecanoyl-2-O-isobutyryl-3-O-tigloylsucrose, respectively.

Physakengose N (4, C<sub>35</sub>H<sub>58</sub>O<sub>14</sub>), isolated as an amorphous solid, contained one more CH<sub>2</sub> than physakengose F [9]. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR data (Table 1) of 4 were almost superimposable with those of physakengose F. A comprehensive study on the 1D and 2D NMR spectra of compound 4 indicated that it had one more methylene group in its fatty acid chain. Thus, the structure of 4 was identified as 3'-O-tridecanoyl-2, 3-di-O-tigloylsucrose.

The 1D and 2D NMR spectroscopic data of physakengose O (5, C<sub>34</sub>H<sub>58</sub>O<sub>14</sub>) showed highly similarity to those of physakengose E [9], except for the positions of dodecanoyl and 3-methylbutanoyl units. The HMBC correlations from H-2 ( $\delta_{\text{H}}$  4.84) to the carbonyl carbon of dodecanoyl ( $\delta_{\text{C}}$  174.4) and from H-3' ( $\delta_{\text{H}}$  5.42) to the carbonyl carbon of 3-methylbutanoyl ( $\delta_{\text{C}}$  174.4) allowed us to formulate 5 as 2-O-dodecanoyl-3'-O-(3-methylbutanoyl)-3-O-tigloylsucrose.

Physakengoses P (6, C<sub>22</sub>H<sub>34</sub>O<sub>13</sub>) and Q (7, C<sub>22</sub>H<sub>36</sub>O<sub>13</sub>) were obtained as amorphous solids. Their  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra (Table 2) were similar to those of physakengoses F and E, except for the absence of signals for dodecanoyl unit attached to C-3', which were further confirmed by their molecular formulas. The further analysis of their 2D NMR spectra allowed us to formulate 6 as 2, 3-di-O-tigloylsucrose and 7 as 2-O-(3-methylbutanoyl)-3-O-tigloylsucrose, respectively.

## 2.2. Antibacterial activity

The antibacterial activity of compounds 1–7 against *S. aureus*, *B. subtilis*, *P. aeruginosa* and *E. coli* was tested using disk diffusion assay with penicillin and streptomycin as positive controls for Gram-

positive and Gram-negative bacteria, respectively. As shown in Table 3, compounds 1–5 had potent positive bacteriostatic effect both against Gram-positive and Gram-negative bacteria with MIC values ranging from 2.16 to 12.76  $\mu\text{g/mL}$  but compounds 6 and 7 had no antibacterial activity. These results revealed that the long fatty acid chain attached to C-2 or C-3' played an important role in the antibacterial activity.

## 3. Conclusion

Sucrose esters, characterized by containing long chain fatty acids attached to the disaccharide, are relatively rare compounds which have been isolated from the Solanaceae, Asteraceae, Cannaceae, and Polygalaceae families [11]. Regarding antibacterial activity, compounds 1–5 showed strong activity, but 6 and 7 were inactive. These results indicate that long chain fatty acid esters attached to sucrose are essential for the inhibitor of the strains tested.

## 4. Experimental

### 4.1. General experimental procedures

The optical rotation values were recorded on a Jasco P-1020 polarimeter and IR data were detected on a Bruker Tensor 27 spectrometer. 1D and 2D NMR experiments were carried out in methanol-*d*<sub>4</sub> on a Bruker Avance III NMR instrument at 500 MHz ( $^1\text{H}$ ) and 125 MHz ( $^{13}\text{C}$ ), and TMS was set as the internal standard. Agilent UPLC-Q-TOF (6520B) was used to acquire HRESIMS data. HPLC analysis was performed on an Agilent 1260 Series instrument equipped with a DAD detector and a Shim-pack VP-ODS column (4.6  $\times$  250 mm, i.d.). Silica gel (200–300 and 100–200 mesh, Qingdao Marine Chemical Co., Ltd.), MCI gel (75–150  $\mu\text{m}$ , Mitsubishi Chemical Corporation, Tokyo, Japan), and ODS (40–63  $\mu\text{m}$ , Fuji) were used for column chromatography (CC). All chemical reagents used were analytical grade (Jiangsu Hanbon Science and Technology Co., Ltd., Nanjing, China).

### 4.2. Plant material

The aerial parts of *P. alkekengi* var. *franchetii* were collected in

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