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Diarylheptanoids and phenylphenalenones from Musa itinerans fruits



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ARTICLE INFO

Article history: Received 28 May 2013 Received in revised form 2 December 2013 Available online 21 April 2014

Keywords: Musaceae Musa itinerans Diarylheptanoids Phenylphenalenones Cytotoxicity

ABSTRACT

Two diarylheptanoids, musaitinerins A and B, one heterodimeric phenylphenalenone musaitinerone and four known phenylphenalenones, identified as 4-hydroxy-2-methoxy-9-phenyl-1*H*-phenalen-1-one, musanolone E, hydroxyanigorufone and irenolone were isolated from the fruits of *Musa itinerans* Cheesm. Their structures were elucidated using spectroscopic analyses. The antimicrobial activity of these compounds was evaluated against *Escherichia coli*, *Staphylococcus aureus* and *Candida albicans*; the cytotoxic activity of these compounds was also evaluated against human erythromyeloblastoid leukemia (K562) and human alveolar carcinoma epithelial (A549) cell lines, respectively. Musaitinerone and musanolone E exhibited weak effects against the A549 cell line, as compared with adriamycin. However, these two compounds did not exhibit any growth inhibition against K562 cells, *S. aureus*, *E. coli* or *C. albicans*. The other compounds were inactive against all of the tested cell lines and microorganisms, even at concentrations as high as 50 μM.

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Introduction

The genus *Musa* (Musaceae) encompasses approximately 30 species, and these are mainly distributed in Southeast Asia with 11 species in China (Wu and Kress, 2000). Certain species, such as *Musa basjoo* Sieb. & Zucc., *Musa balbisiana* Colla, *Musa× paradisiaca* Linn. and *Musa itinerans* Cheesm, are used as traditional Chinese herbs (Editorial Board of Zhonghua Bencao, 1999). The major chemical constituents of *Musa* include phenylphenalenones and their respective dimers; these compounds have a variety of biological activities, such as antifungal (Kamo et al., 2001; Luis et al., 1995; Otálvaro et al., 2007), α-glucosidase inhibitory (Zhang and Kang, 2010) and cancer chemopreventive properties (Jang et al., 2002). *M. itinerans*, which is synonymous to *Musa wilsonii* Tutch., is mainly distributed in southern China and has been used to treat malaria (Liu and Li, 2000; Liu et al., 2002). However, its chemical constituents have not yet been reported. With an

ongoing interest in natural anticancer agents (Zhao et al., 2010; Tang et al., 2011; Su et al., 2013; Yang et al., 2013) and cytotoxic phenylphenalenones (Dong et al., 2011), a phytochemical analysis of the fruits of *M. itinerans*, was carried out. Thus led to isolation of two new diarylheptanoids (1 and 2), one new heterodimeric phenylphenalenone (3) and four known phenylphenalenones (4–7). This report describes the structural characterization of the new compounds and the bioassay results obtained for all isolates.

Results and discussion

A methanol (MeOH) extract of M. itinerans fruits was suspended in water (H_2O) and partitioned with ethyl acetate (EtOAc). Repeated column chromatography of the EtOAc-soluble portion, which was performed on silica gel, C_{18} silica gel and Sephadex LH-20, respectively, was followed by preparative thin-layer chromatography (TLC) and semipreparative high-performance liquid chromatography (HPLC) to yield compounds $\bf 1$ to $\bf 7$ (Fig. 1).

Compound **1** was obtained as a white amorphous solid, and its molecular formula, $C_{41}H_{44}O_{12}$, was assigned based on high-resolution electrospray ionization mass spectroscopy (HRESIMS) data, which indicated 20 degrees of unsaturation. Its IR spectrum showed absorption bands for hydroxyl (3428 cm⁻¹), conjugated

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Fig. 1. Structures of compounds 1-7 and 1a.

carboxyl (1706 and 1632 cm $^{-1}$) and aromatic (1605, 1585, 1516 and 1453 cm $^{-1}$) functional groups. The 1 H and 13 C NMR spectra for this compound displayed signals indicative of four substituted benzene rings, two *E*-double bonds [$\delta_{\rm H}$ 7.64 (d, J = 15.9 Hz), 6.64 (d, J = 15.9 Hz), 6.38 (dt, J = 15.9 Hz) and 6.1 Hz) and 6.38 (d, J = 15.9 Hz)], one ester carbonyl group ($\delta_{\rm C}$ 167.2) and four O-methyl groups (Table 1).

Several connections, such as the linkages from C-1 to C-7 and from C-7" to C-9", were elucidated based on ¹H-¹H correlation spectroscopy (COSY) (Fig. 2). Furthermore, three fragments, namely a substituted diarylheptanoid, a sinapyl alcohol and an Ep-coumaroyl group, were deduced through ¹H-¹H COSY, heteronuclear multiple bond correlation (HMBC) and rotational frame nuclear Overhauser effect spectroscopy (ROESY) experiments (Fig. 2), respectively. The *E-p-*coumaroyl group was connected to the 9"'-OH of the sinapyl alcohol moiety by examining HMBC correlations between H₂-9" and C-9". Furthermore, the same group was also linked to the diarylheptanoid moiety through a C-6-O-C-4" bonding pattern by analyzing the HMBC correlation between H-6 and C-4". This linkage was also indicated by the relatively lowfield chemical shifts of H-6 (δ_H 4.26) and C-6 (δ_C 82.4) compared with the analogous chemical shifts (δ_H 3.51 and δ_C 73.6) of the known compound, (3S,5S,6S,7R)-5,6-dihydroxy-1,7-bis(4-hydroxyphenyl)-4"-de-O-methylcentrolobine (Ali et al., 2001).

The relative configuration of 1 was deduced from analysis of the coupling constants in its ¹H NMR spectrum (Table 1) and ROESY correlations (Fig. 3). If the pyran ring is in a chair conformation, H-7 and H-6 should have a trans-diaxial relationship, which is indicated by the $J_{6,7}$ = 9.5 Hz coupling constant. H-7 was arbitrarily assigned to have a β -orientation, which would make H-6 α -oriented. The $J_{5,6}$ = 2.4 Hz value indicated that H-5 and H-6 were in a cis relationship, with an H-5 equatorial- and α -orientation. Two protons [δ_H 1.90 (br d, J = 12.1 Hz) and 1.56 (br t, J = 12.1 Hz)] were linked to C-4; the former should be equatorial, and the latter should be axial. In the ROESY spectrum of 1, the correlations between H-4 α and H-6 indicated that these two protons were co-facial. The ROESY correlations between H-4α and H₂-2 indicated that H-4 α and H-3 were in a trans relationship, which was also indicated by the $J_{3,4\alpha}$ = 12.1 Hz. Therefore, H-3 was assigned as β oriented. Other ROESY correlations, such as that between H-3 and H-7, as well as that observed between 5-OH and H-7, further confirmed the deduction above. Thus, the relative configuration of 1 was deduced and was given the common name musaitinerin A.

To determine its absolute configuration, its 4',4'',4'''-tri-0-methyl derivative $\mathbf{1a}$ (Fig. 1) and Mosher esters of $\mathbf{1a}$ were synthesised. However, the $\Delta\delta_{\rm H}$ ($\delta_{\rm S}-\delta_{\rm R}$) values contradicted Mosher's rule (Supplementary Data), which might be due to steric compression of H-3 and H-7 to the MTPA group (Ohtani et al., 1991). In addition, the calculated and experimental electron circular dichroism (ECD) curves were not self-consistent (data not shown). The absolute configuration of $\mathbf{1}$ remains unclear.

Musaitinerin B (**2**) was obtained as a white amorphous solid, and its molecular formula, $C_{41}H_{44}O_{12}$, was assigned based on the HRESIMS data, which indicated 20 degrees of unsaturation. The 1H and ^{13}C NMR spectra (Table 2) of **2** were similar to those of **1**. Analysis of the 2D NMR spectra (Supplementary Data) of **2** indicated that it was also composed of three fragments: a substituted diarylheptanoid, a sinapyl alcohol and a Z-p-coumaroyl group. According to the 2D NMR correlations (Supplementary Data), the linkage patterns and the relative configuration were analogous to those deduced for **1**. Thus, the structure of musaitinerin B (**2**) was determined, as shown in Fig. 1.

Musaitinerone (**3**) was obtained as an orange amorphous solid, and its molecular formula, $C_{38}H_{22}O_7$, was assigned based on the HRESIMS data (m/z 613.1267 [M+Na]*), which indicated 28 degrees of unsaturation. The 1H and ^{13}C NMR spectra (Table 2) displayed signals that indicated two para-disubstituted benzene rings [δ_H 7.36 (2H, d, J = 8.5 Hz), 7.15 (2H, d, J = 8.5 Hz), 6.92 (2H, d, J = 8.5 Hz) and 6.87 (2H, d, J = 8.5 Hz)], one O-methyl group [δ_H 2.87 (3H, s); δ_C 51.8] and three carbonyl groups (δ_C 183.1, 175.6 and 169.9).

Phenylphenalenones, such as compounds **4** to **7**, are the major constituents of *Musa* plants. These compounds usually possess a C₁₉ molecular skeleton, with approximately 14 degrees of unsaturation. Comparison of the collected data to the aforementioned characteristics indicated that compound **3** may be tentatively considered a phenylphenalenone dimer. Based on the ¹H–¹H COSY, HMBC and ROESY correlations (Fig. 2) for **3**, fragments A (left moiety, shown in blue) and B (right moiety, shown in black) were elucidated. Specifically, the key HMBC correlations such as 4′-OH to C-5′, H-6′ to C-9, H-8 to C-1′ and C-9a, H-7 to C-9b, H-6 to C-4 and H-5 to C-3a supported the presence of fragment A. Although the HMBC spectrum of **3** showed no long-range correlations for C-1, C-2 and C-3, this fragment should consist of a phenylphenalenone skeleton based on a comparison of its NMR spectroscopic data with those of known compounds **4** to **7**. Fragment B was

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