

Limonoids from the stem bark of *Khaya grandifoliola*

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Abstract

Three limonoids, deacetylkhayanolide E (**1**), 6S-hydroxykhayalactone (**2**), and grandifolide A (**3**), along with three known ones, were isolated from stem bark of the Nigerian medicinal plant *Khaya grandifoliola*. Their structures were characterized on the basis of the application of spectroscopic methods. *In vitro* antimicrobial and cytotoxic activities of the isolates were also tested, but were all found to be inactive.

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1. Introduction

Some plant species in the Meliaceae and Rubiaceae families, e.g. *Entandrophragma utile*, *Azadirachta indica*, *Morinda lucida*, and *Khaya* species, produce structurally diverse limonoids with important antiplasmodial and cytotoxic activities (Obih et al., 1985; Bray et al., 1990; Weenen et al., 1990). There are eight species of *Khaya* growing in tropical regions (Chen et al., 1997), and *K. grandifoliola* has been used for treatment of fever and malaria in Nigeria. In particular, a crude extract of stem bark showed *in vitro* mollusc-killing properties (Makanga and Odyek, 1989), trypanocidal activity against *Trypanosoma brucei* (Owolabi et al., 1990), and antimalarial activity against *Plasmodium bergeri bergeri* (Makinde et al., 1988; Awe and Makinde, 1991; Awe et al., 1997). Recent investigations of both the stem bark and the seeds of this plant have led to isolation of eight limonoids and one flavanol (Agbedahunsi and Elujoba, 1998; Tchuendem et al.,

1998; Bickii et al., 2000). In the present paper, we report the isolation and structural elucidation of six additional limonoids, deacetylkhayanolide E (**1**), 6S-hydroxykhayalactone (**2**), grandifolide A (**3**), khayanolide A (**4**, Abdelgaleil et al., 2001), anthothecanolide (**5**, Tchimine et al., 2005), and 3-*O*-acetylanthothecanolide (**6**, Tchimine et al., 2005), of which compounds **1–3** are new. The *in vitro* antimicrobial and cytotoxic activities of compounds **1–6** were also tested, but all tested compounds were inactive.

2. Results and discussion

Deacetylkhayanolide E (**1**) had a molecular formula of C₂₇H₃₂O₁₀ as deduced from HREIMS at *m/z* 516.1979 [M]⁺ (calcd. 516.1995). Its IR spectrum showed the presence of hydroxyl (3498 cm⁻¹) and carbonyl (1732 and 1713 cm⁻¹) groups. The spectroscopic data of compound **1** (EIMS, IR, and NMR) were very similar to those of khayanolide E (Nakatani et al., 2002). Comparison of the ¹H and ¹³C NMR spectroscopic data (Tables 1 and 2) of **1** with those of khayanolide E showed that the only structural difference was the absence of the 1-*O*-acetyl group in **1**, and this was supported by its molecular composition.

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Table 1
¹H NMR spectroscopic data for compounds **1–6** (400 MHz)

Proton	1 ^a	2 ^a	3 ^a	3 ^b	4 ^a	5 ^c	6 ^a
2	4.83 <i>d</i> (10.4)						
3		4.01 <i>s</i>	5.74 <i>s</i>	4.74 <i>s</i>	4.25 <i>s</i>	3.60 <i>s</i>	5.73 <i>s</i>
5	4.09 <i>br s</i>	4.22 <i>d</i> (1.6)	3.32 <i>d</i> (7.9)	2.38 <i>d</i> (4.0)	2.61 <i>d</i> (6.5)	2.26 <i>dd</i> (9.6, 7.7)	2.26 <i>dd</i> (9.0, 7.2)
6	4.90 <i>br s</i>	4.92 <i>br s</i>	4.93 <i>d</i> (7.9)	4.38 <i>br s</i>	4.86 <i>d</i> (6.5)	2.70 <i>dd</i> (15.6, 9.6)	2.92 <i>dd</i> (14.2, 9.0)
						2.41 <i>dd</i> (15.6, 7.7)	2.74 <i>dd</i> (14.2, 7.2)
9	3.07 <i>d</i> (8.7)	2.44 ^h	2.78 <i>m</i>	1.61 ^l	2.49 <i>dd</i> (13.2, 5.6)	2.48 <i>br d</i> (12.0)	3.22 <i>br d</i> (11.4)
11	2.16 ^d	2.15 <i>m</i>	2.32 <i>m</i>	1.88 <i>m</i>	1.52 ⁿ	1.65 <i>m</i>	2.32 <i>m</i>
	1.92 ^e	2.04 <i>m</i>	2.02 <i>m</i>	1.59 ^l	1.33 <i>m</i>	1.39 ^o	2.00 ^p
12	2.22 ^d	1.35 <i>m</i>	1.87 ^l	1.82 <i>br d</i> (14.9)	1.60 <i>m</i>	1.69 <i>m</i>	1.97 ^p
	1.08 <i>br d</i> (11.2)	1.10 <i>dd</i> (14.3, 8.7)	1.37 <i>m</i>	1.33 <i>m</i>	1.47 ⁿ	1.39 ^o	1.54 <i>br d</i> (14.7)
14		1.89 ⁱ	2.42 ^k				
15	3.83 <i>d</i> (18.6)	3.60 <i>dd</i> (15.0, 13.0)	3.48 <i>d</i> (19.1)	2.86 ^m (2H)	3.52 <i>d</i> (18.2)	3.35 <i>d</i> (19.2)	3.80 <i>d</i> (18.7)
	3.36 <i>d</i> (18.6)	3.00 <i>dd</i> (15.0, 6.0)	3.02 <i>dd</i> (19.1, 8.0)		3.15 <i>d</i> (18.2)	2.73 <i>d</i> (19.2)	3.15 <i>d</i> (18.7)
17	5.85 <i>s</i>	5.37 <i>s</i>	6.10 <i>s</i>	5.32 <i>s</i>	5.87 <i>s</i>	5.56 <i>s</i>	6.11 <i>s</i>
18	1.39 <i>s</i>	0.87 <i>s</i>	1.02 <i>s</i>	0.98 <i>s</i>	1.18 <i>s</i>	0.98 <i>s</i>	0.75 <i>s</i>
19	1.94 <i>s</i>	1.92 <i>s</i>	2.11 <i>s</i>	1.39 <i>s</i>	1.94 <i>s</i>	4.42 <i>d</i> (11.6)	4.83 <i>d</i> (11.2)
						4.15 <i>d</i> (11.6)	4.59 <i>d</i> (11.2)
21	7.65 ^f	7.65 <i>br s</i>	7.53 <i>br s</i>	7.49 <i>br s</i>	7.35 <i>br s</i>	7.62 <i>br s</i>	7.66 <i>br s</i>
22	6.55 <i>br s</i>	6.51 <i>d</i> (1.2)	6.49 <i>d</i> (1.0)	6.40 <i>d</i> (1.2)	6.45 <i>d</i> (1.2)	6.48 <i>d</i> (1.2)	6.54 <i>dd</i> (1.7, 0.6)
23	7.64 ^f	7.63 <i>t</i> (1.6)	7.61 <i>t</i> (1.5)	7.43 <i>t</i> (1.7)	7.61 <i>t</i> (1.7)	7.52 <i>t</i> (1.8)	7.60 <i>t</i> (1.8)
28	1.72 <i>s</i>	1.89 <i>s</i>	1.23 <i>s</i>	0.95 <i>s</i>	1.66 <i>s</i>	0.94 <i>s</i>	1.18 <i>s</i>
29	2.67 <i>d</i> (12.2)	2.60 <i>d</i> (13.7)	1.83 <i>s</i>	1.30 <i>s</i>	2.79 <i>d</i> (12.1)	1.29 <i>s</i>	1.64 <i>s</i>
	2.17 <i>d</i> (12.2)	2.46 <i>d</i> (13.7)			2.40 <i>d</i> (12.1)		
30	3.49 <i>d</i> (10.4)	4.03 <i>d</i> (12.6)	3.60 <i>d</i> (14.1)	2.53 <i>d</i> (14.3)	3.41 <i>s</i>	2.93 <i>d</i> (13.9)	3.59 <i>d</i> (13.8)
		2.78 <i>d</i> (12.6)	2.41 ^k	1.57 ^l		1.50 <i>dd</i> (13.9, 1.5)	2.40 <i>dd</i> (13.8, 1.4)
–COOMe	3.73 <i>s</i>	3.92 <i>s</i>	3.83 <i>s</i>	3.81 <i>s</i>	3.67 <i>s</i>		
CH ₃ COO–			1.74 <i>s</i>	2.15 <i>s</i>			1.74 <i>s</i>
1-OH	6.99 <i>br s</i>	8.05 ^A		3.90 <i>br s</i>			7.23 <i>br s</i>
OH	7.57 ^g (6-OH)	8.04 ^A (3-OH)		4.76 <i>br s</i> (2-OH)			10.08 <i>br s</i> (2-OH)
	7.36 <i>br s</i> (8-OH)	7.23 <i>br s</i> (6-OH)		2.88 ^m (6-OH)			7.08 <i>s</i> (14-OH)

^A Interchangeable signals.

^a Recorded in C₅D₅N.

^b Recorded in CDCl₃.

^c Recorded in CD₃OD.

^{d–p} Overlapping signals.

The structure of compound **1** was further confirmed by 2D NMR spectra, including HSQC, HMBC, and NOESY spectra (Supplementary data). Compound **1** was therefore determined to be deacetylkhayanolide E.

6*S*-Hydroxykhayalactone (**2**) showed a molecular formula of C₂₇H₃₄O₁₀ as determined by HREIMS at *m/z* 518.2146 [M]⁺ (calcd. 518.2152), which is 16 mass units more than that of khayalactone (Tchuendem et al., 1998). The spectroscopic data (EIMS, IR, and NMR) of **2** were very similar to those of khayalactone, indicating that the structures of both compounds were related, and the only difference is likely the presence of one more hydroxyl group in **1**. The NMR spectroscopic data further indicated that compound **2** possessed one more oxygenated methine (δ_H 4.92, *br s*; δ_C 73.4) and one fewer methylene group than khayalactone, supporting the above deduction. In the HMBC, the correlations of proton signal (δ_H 4.92, *br s*) to C-4 (δ_C 47.3), C-5 (δ_C 44.1), C-7 (δ_C 176.7), and C-10 (δ_C 56.6) established that the additional hydroxyl group in **2** was located at C-6. Comparison of the chemical shifts of both H-6 and C-6 of **2** with those of compound **1** sug-

gested that the 6-OH had the *S*-configuration. The structure of **2** was thus elucidated, and confirmed by HMQC, HMBC, and NOESY spectra.

Grandifolide A (**3**) was determined to have a molecular formula C₂₉H₃₈O₁₁ by HREIMS at *m/z* 562.2392 [M]⁺ (calcd. 562.2414). The NMR spectroscopic data (Tables 1 and 2, in C₅D₅N) of compound **3** were very similar to those of 3-*O*-acetylanthothecanolide (**6**, Tchimene et al., 2005), indicating that they were structurally related analogs. In the HMBC spectrum, the correlations of OMe (δ_H 3.83, *s*)/C-7 (δ_C 175.8) and H₃-19 (δ_H 2.11, *s*)/C-10 (δ_C 46.2) showed that the partial structure of **3** (C-5 to C-7, C-5 to C-10, and C-10 to C-19) is the same as that of compound **2**. The proton signal at δ_H 2.42 was assignable to H-14 by its HMBC correlations with C-8 (δ_C 79.2), C-13 (δ_C 35.7), and C-15 (δ_C 28.5), indicating that the C-14 was a methine. The relative stereochemistry of compound **3** was confirmed by NOESY spectroscopic analysis (Supplementary data). The complete assignments of the ¹H and ¹³C NMR spectroscopic data of **3** were therefore fully achieved by HMQC spectroscopy.

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