



PHYTOCHEMISTRY

Phytochemistry 66 (2005) 2309-2315

www.elsevier.com/locate/phytochem

# Sesquiterpenes from Warburgia ugandensis and their antimycobacterial activity

Abraham Abebe Wube <sup>a</sup>, Franz Bucar <sup>a,\*</sup>, Simon Gibbons <sup>b</sup>, Kaleab Asres <sup>c</sup>

a Department of Pharmacognosy, Institute of Pharmaceutical Sciences, Karl-Franzens University Graz, Universitaetsplatz 4/1, A-8010 Graz, Austria
b Centre for Pharmacognosy and Phytotherapy, The School of Pharmacy, University of London, 29-39 Brunswick Square, London WCIN 1AX, UK
c Department of Pharmacognosy, The School of Pharmacy, Addis Ababa University, P.O. Box 1176, Addis Ababa, Ethiopia

Received 3 May 2005; received in revised form 22 July 2005 Available online 8 September 2005

#### Abstract

The dichloromethane extract of the stem bark of *Warburgia ugandensis* afforded three new coloratane sesquiterpenes, namely:  $6\alpha,9\alpha$ -dihydroxy-4(13),7-coloratadien-11,12-dial (1), 4(13),7-coloratadien-12,11-olide (2), and 7 $\beta$ -hydroxy-4(13),8-coloratadien-11,12-olide (3), together with nine known sesquiterpenes, i.e., cinnamolide-3 $\beta$ -acetate (4), muzigadial (5), muzigadiolide (6), 11 $\alpha$ -hydroxymuzigadiolide (7), cinnamolide (8), 7 $\alpha$ -hydroxy-8-drimen-11,12-olide (9), ugandensolide (10), mukaadial (11), ugandensidial (12), and linoleic acid (13). Their structures were assigned on the basis of 1D and 2D-NMR spectroscopic and GC-MS analysis.

The compounds were examined for their antimycobacterial activity against Mycobacterium aurum, M. fortuitum, M. phlei and M. smegmatis; and the active constituents showed MIC values ranged from 4 to 128  $\mu$ g/ml compared to the antibiotic drugs ethambutol (MIC ranged from 0.5 to 8  $\mu$ g/ml) and isoniazid (MIC ranged from 1 to 4  $\mu$ g/ml). © 2005 Elsevier Ltd. All rights reserved.

Keywords: Warburgia ugandensis; Canellaceae; Antimycobacterial activity; Drimane sesquiterpenes; Coloratane sesquiterpenes; 6α,9α-Dihydroxy-4(13),7-coloratadien-11,12-dial; 4(13),7-Coloratadien-12,11-olide; 7β-Hydroxy-4(13),8-coloratadien-11,12-olide

#### 1. Introduction

Warburgia ugandensis Sprague (Canellaceae), which is commonly known as zogdom in Amharic, is characterized by its bitter and peppery taste. The stem bark has been widely used in East African ethnomedicine for the treatment of stomach-ache, constipation, toothache, cough, fever, muscle pains, weak joints and general body pains (Kokwaro, 1976; Watt and Breyer-Brandwijk, 1962). The Shinasha people in Ethiopia use the stem bark for the treatment of tuberculosis. Species of the genus Warburgia are known to be rich in sesquiterpenes of the drimane and coloratane skeletones (Kioy et al.,

E-mail address: franz.bucar@uni-graz.at (F. Bucar).

1990; Mashimbye et al., 1999), which have been shown to possess insect antifeedant, antimicrobial, antiulcer, molluscicidal (Kubo et al., 1983) and antifungal properties (Kubo and Taniguchi, 1988). Previous phytochemical investigation of *W. ugandensis* showed the presence of muzigadial, ugandensidal, pereniporin B, polygodial, mukaadial, warburganal, cinnamolide and 11α-hydroxymuzigadiolide in the stem bark; and ugandensolide, ugandensidial, warburgin and warburgiadione in the heart wood (Brooks and Draffan, 1969).

The drimanes, a group of sesquiterpenoides isolated from species of the genus Warburgia, are characterized by  $\alpha,\beta$ -unsaturated carbonyl chromophores assembled around a trans-decalin ring system. As part of our search for antimycobacterial agents from Ethiopian medicinal plants, we identified three new and nine known sesquiterpenes along with a known unsaturated

<sup>\*</sup> Corresponding author. Tel.: +43 316 380 5531; fax: +43 316 380

fatty acid from the stem bark of *W. ugandensis* and evaluated their antimycobacterial activity against four rapidly growing species of mycobacteria.

There are several reports on the constituents of this plant, but no report has been found on their antimyco-bacterial activity.

#### 2. Results and discussion

Compound 1 was obtained as colourless needles in nhexane/CH<sub>2</sub>Cl<sub>2</sub> (see Fig. 1). A molecular formula  $C_{15}H_{20}O_4$  was determined by HRMS (m/z; measured  $287.1264 \text{ [M + Na]}^+$ ; calc. 287.1259). In addition, a prominent peak at m/z 235  $[M - CHO]^+$  was present in the EI-MS spectrum, corresponding to a molecular formula C<sub>15</sub>H<sub>20</sub>O<sub>4</sub>, which is in agreement with 15 carbon signals observed in the <sup>13</sup>C NMR spectrum. A broad band absorption at 3406 cm<sup>-1</sup> in the IR spectrum suggested the presence of hydroxyl groups. In addition, the IR spectrum showed carbonyl absorptions at 1725, 1683 cm<sup>-1</sup> and an olefinic absorption at 1640 cm<sup>-1</sup>. An absorption maximum at 234 nm in the UV spectrum was also indicative of an  $\alpha,\beta$ -unsaturated lactone. The <sup>1</sup>H NMR spectrum (Table 1) of **1** showed characteristic signals of a coloratadiene sesquiterpene ring system (Ying et al., 1995) with signals at  $\delta$  5.03 and 5.13 attributable to two exocyclic methylene protons, as well as a one proton doublet at  $\delta$  7.10 for H-7. A singlet at  $\delta$ 0.96 and a doublet at 1.11 for two methyl groups were also observed in the <sup>1</sup>H NMR spectrum. The signals for H-13a and H-13b protons were unambiguously assigned based on NOE correlations observed between CH<sub>3</sub>-14 and H-13a and between H-6β and H-13b in the NOESY spectrum. The signals at  $\delta$  9.50 and 9.65 were attributed to two aldehyde groups which were further confirmed by the carbonyl signals at  $\delta$  192.6 and

Fig. 1. The structure of compounds 1-4 isolated from *W. ugandensis* stem barks.

Table 1 <sup>1</sup>H NMR spectral data for compounds 1–3<sup>a</sup> (500 Hz, CDCl<sub>3</sub>)

Proton	1	2	3
1α	1.02 (dt) (13.5, 4.0)	1.54 (dt) (13.5,	1.47 (dt) (13.5,
		4.0)	4.0)
1β	2.05 (m)	1.65 (td) (13.5, 3.0)	2.58 (m)
$2\alpha$	1.12 (m)	1.25 (m)	1.33 (m)
2β	1.73 (m)	1.70 (td) (13.0, 3.0)	$1.80 \ (m)$
3	2.00 (m)	2.05 (m)	2.11 (m)
5	2.65 (d) (10.0)	2.18 (m)	2.38 (bd) (13.0)
6α		2.35(m)	1.92 (m)
6β	4.70 (dd) (10.0, 2.5)	2.32 (m)	1.99 (m)
7	7.10 (d) (2.5)	6.91(q)(3.5)	4.54 (d) (4.0)
9		2.99(m)	
11α	9.65 (s)	4.48 (t) (9.0)	
11β		4.02 (t) (9.0)	
$12\alpha$	9.50 (s)		4.68 (d) (17.0)
12β			4.96 (d) (17.0)
13α	5.13 (s)	4.90(s)	4.86 (s)
13β	5.03(s)	4.73(s)	4.62(s)
14	1.11 (d) (6.5)	1.11 (d) (6.5)	1.10 (d) (7.0)
15	0.96(s)	0.65(s)	0.89(s)
6-OH	1.59 (bs)		
7-OH			1.94 (s)
9-OH	4.07 (bs)		·

Coupling constant values (in parentheses) are in Hz.

200.5 in the <sup>13</sup>C spectrum. The remaining two broad singlets at  $\delta$  1.59 and 4.07 in the <sup>1</sup>H NMR spectrum, which did not exhibit any correlations in the HMOC spectrum, were assigned to two hydroxyl groups. The former was assigned to a hydroxyl group bonded to the tertiary carbon, C-6, whereas the downfield broad singlet at  $\delta$  4.07 was assigned to a hydroxyl group attached to the quaternary carbon, C-9. This was further supported by the downfield carbon resonances,  $\delta$  66.1 and 77.6, observed in the <sup>13</sup>C NMR spectrum of 1 for C-6 and C-9, respectively. The <sup>13</sup>C and DEPT analyses gave signals corresponding to two methyl, three methylene, six methine and four quaternary carbons further confirming the presence of a coloratane type sesquiterpene. Carbon resonances at  $\delta$  106.7, 139.3, 149.1 and 153.7 were assigned to four olefinic carbons, whereas carbon signals at  $\delta$  66.1 and 77.6 were assigned to methine and quaternary carbons bearing hydroxyl groups, respectively. The HMQC, HMBC and NOESY experiments allowed unambiguous assignment of the chemical shift values of the methylene protons at C-1 and C-2. Assignment of the relative stereochemistry of the two hydroxyl groups in 1 was accomplished by analyses of the coupling constants and NOESY spectrum. The proton H-6 showed NOE correlation (Fig. 2) to CH<sub>3</sub>-15, thus H-6 occupied a position axial to the axial CH<sub>3</sub>-15 group at C-10 leaving OH-6  $\alpha$ -oriented. This is in agreement with the observed coupling constants for H-6 (J = 10.0 Hz) with H-5 and (J = 2.5 Hz) with H-7. Similarly, a cross NOE peak was observed between the aldehyde proton H-11 and CH<sub>3</sub>-15, indicating that the

<sup>&</sup>lt;sup>a</sup> Chemical shifts are in ppm relative to TMS.

### Download English Version:

## https://daneshyari.com/en/article/5168122

Download Persian Version:

https://daneshyari.com/article/5168122

<u>Daneshyari.com</u>