

Chemical constituents from *Oncocalyx glabratus* and their biological activities

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ARTICLE INFO

Keywords:

Oncocalyx glabratus

Flavan gallates

Anti-HBV activity

PPAR agonistic activity

ABSTRACT

Chemical investigation of the aerial parts of *Oncocalyx glabratus* resulted in the isolation of three new flavan derivatives, 5,3',4'-trihydroxyflavan 7-O-gallate (1), 5,4'-dihydroxyflavan 7-3'-O-digallate (2) and 5,3'-dihydroxyflavan 7-4'-O-digallate (3), named oncoglabrinol A, B and C, respectively, together with four known flavonols, (+)-catechin (4), (+)-catechin-7-O-gallate (5), catechin-7-4'-O-digallate (6A) and catechin-7-3'-O-digallate (6B). The structures of the compounds were established by 1D, 2D NMR and ESI-HRMS spectral analyses. The biological activity of the compounds was tested through a series of *in vitro* assays designed for determining cytotoxicity, antiviral activity against hepatitis B virus, and antidiabetic activity. All compounds were found non-toxic and showed moderate anti-HBV activity. Compounds 3 and 6 showed dual PPAR agonistic activity while others were not effective.

1. Introduction

Oncocalyx glabratus belongs to family Loranthaceae which is the largest family of flowering parasitic plants that includes about 70 genera and 1000 species (Calvin and Wilson, 2006). This family comprises epiphytic and hemiparasitic plants, most popularly known as mistletoe (Lorenzi, 2000). Mistletoes are known as “cure all” and have been found to have remedial effects for many health problems (Adodo, 2004), including diabetes (Obatomi et al., 1994).

O. glabratus is locally available in the Kingdom of Saudi Arabia (Collenette, 1999). A literature survey of the plant revealed antibacterial activity and phytochemical screening showed presence of reducing sugar, terpenoids, steroids, flavonoids and tannins (Waly et al., 2012).

The polyphenolic catechins such as epigallocatechin-3-gallate (EGCG), epicatechin-3-gallate (ECG), epigallocatechin (EGC), and epicatechin (EC), major constituents of green tea (*Camellia sinensis*), are reported to have antioxidative, anti-inflammatory, antitumor, antidiabetic and antibacterial activities *in vitro* and *in vivo* (Jin, 2013). Moreover, the antiviral efficacy of EGCG and ECG has been demonstrated against human immunodeficiency virus (HIV), human T-cell leukemia virus (HTLV), influenza virus, herpes simplex virus (HSV), rotavirus, adenovirus, Epstein–Barr virus (EBV), and hepatitis viruses

(Jin, 2013).

We have recently reported hypoglycemic and antidiabetic activity of this plant (Ahmed et al., 2015). In the present study, we report on the isolation and structure elucidation of three new flavan derivatives, namely oncoglabrinol A (1), oncoglabrinol B (2) and oncoglabrinol C (3) (Fig. 1) from the active ethyl acetate extract of *O. glabratus*. In addition, this material also yielded four known compounds, (+)-catechin (4) (Nonaka et al., 1983), (+)-catechin-7-O-gallate (5) (Tanaka et al., 1983), catechin-7,4'-O-digallate (6A) and catechin-7,3'-O-digallate (6B) (El-Toumy and Mahdy, 2004). The cytotoxicity, antiviral activity against hepatitis B virus and antidiabetic activity of the isolates is also reported.

2. Results and discussion

Compound 1 was obtained as yellowish color sticky material with the molecular formula $C_{22}H_{18}O_9$ determined by ESI-HRMS. The 1H and ^{13}C NMR assignments of 1 (Table 1) were based on the COSY, HSQC, and HMBC spectra. The ^{13}C NMR and DEPT spectra confirmed the presence of twenty-two carbons: twelve quaternary, two methylene and eight methine carbons. The 1H and ^{13}C NMR spectra (Table 1) showed two aromatic signals [δ_H 6.68 (d, J = 7.7 Hz, 1H, H-6'), 6.73 (d,

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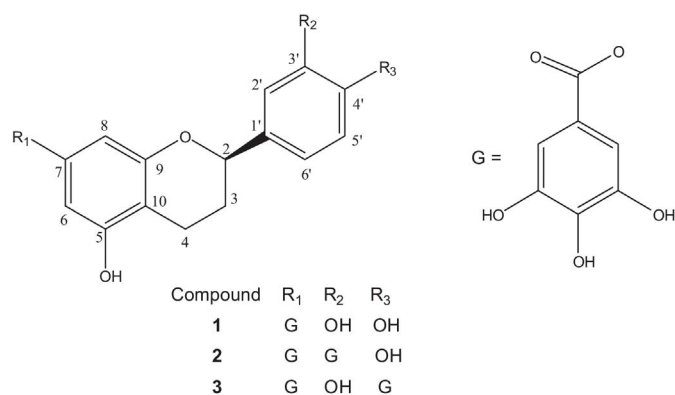


Fig. 1. Structure of new compounds 1–3.

Table 1
¹H NMR (700 MHz) and ¹³C NMR (175 MHz) data of compound 1–3 in DMSO-*d*₆.

No.	1		2		3	
	δ_{C}	δ_{H} (<i>J</i> in Hz)	δ_{C}	δ_{H} (<i>J</i> in Hz)	δ_{C}	δ_{H} (<i>J</i> in Hz)
2	77.4	4.87 d (9.8)	77.2	4.97 d (8.4)	77.2	5.04 d (7.0)
3	29.1	2.09 m, 1.88 m	29.1	2.13 m, 1.97 m	29.0	2.17 br s, 1.94 br s
4	19.7	2.60 m	19.6	2.63 m	19.7	2.66 br s
5	156.4	–	156.4	–	156.2	–
6	101.3	6.11 s	101.4	6.19 s	101.3	6.17 s
7	150.0	–	150.0	–	150.1	–
8	100.9	6.18 s	101.0	6.15 s	101.1	6.21 s
9	156.5	–	156.4	–	156.5	–
10	107.4	–	107.3	–	107.4	–
1′	132.8	–	132.6	–	140.2	–
2′	114.2	6.80 s	121.8	7.15 s	114.8	7.02 s
3′	145.5	–	138.9	–	149.4	–
4′	145.4	–	149.2	–	138.7	–
5′	115.7	6.73 d (8.4)	117.0	6.97 d (5.6)	123.7	7.07 merge in 7.09–7.06
6′	117.7	6.68 d (7.7)	124.9	7.16 br s	117.2	6.89 br s
7-O-galloyl						
1″	118.8	–	118.8	–	118.9	–
2″, 6″	109.5	7.05 s	109.4	7.05 s	109.4	7.09 s
3″, 5″	146.2	–	146.2	–	146.2	–
4″	139.4	–	139.4	–	139.3	–
7″	164.7	–	164.6	–	164.7	–
3′-O-galloyl						
1″″			118.9	–		
2″″, 6″″			109.6	7.09 s		
3″″, 5″″			146.1	–		
4″″			139.5	–		
7″″			164.9	–		
4′-O-galloyl						
1″″″					118.8	–
2″″″, 6″″″					109.6	7.06 s
3″″″, 5″″″					146.1	–
4″″″					139.6	–
7″″″					164.9	–
OH-5				9.76 s		
OH-4′				9.78 s		
OHs				9.41 s, 9.12 s		9.79 s, 9.40 s, 9.13 s

$J = 8.4$ Hz, 1H, H-5'); δ_C 117.7 (C-6'), 115.7 (C-5')], showing COSY correlation suggested that the two methine carbons were adjacent to each other. The signals resonating at [δ_H 6.80 (s, 1 H, H-2'); δ_C 114.2 (C-2')] and the three quaternary carbons at δ_C 132.8 (C-1'), 145.5 (C-3') and 145.4 (C-4') suggesting the B ring of flavan. The down field shift of C3' (δ_C 145.5) and C4' (145.4) suggested the presence of hydroxyl group (Ahmed et al., 2014). Other signals at δ_H 1.88 (m, 1H, Ha-3),

2.09 (m, 1H, Hb-3), δ_C 29.1 (C-3); δ_H 2.60 (m, 2H, H-4), δ_C 19.7 (C-4) and 4.87 (d, $J = 9.8$ Hz, 1H, H-2); δ_C 77.4 (C-2) suggested the occurrence of sets of two methylene groups and one oxygenated methine moiety in ring C (Li et al., 2006). Additionally Ha-3 was showing COSY correlations with the geminal Hb-3 and both Ha/b-3 showing COSY correlations with vicinal H-2 and H₂-4. Furthermore the ¹H NMR spectrum displayed two singlets at δ 6.11 (1H, s, H-6) and 6.18 (1H, s, H-8), which were consistent with a 5,7-dioxygenated A ring of flavan (Al-Taweel et al., 2012). The down field chemical shift of C-5 (δ_C 156.4) suggested the presence of OH group at C-5 position in ring A (Lee et al., 1992). Two methine aromatic carbons [δ_H 7.05 (s, 2 H); δ_C 109.5, 2 C] and five quaternary carbons (δ_C 139.4, 146.2 \times 2, 118.8 and 164.7) indicating the presence of galloyl moiety (Al-Taweel et al., 2012). Up field shift of C-7 (δ_C 150.0) carbon as well as downfield shift of C-6 (δ_C 101.3), C-8 (δ_C 100.9) and C-10 (δ_C 107.4) and comparison of the data with 7-O-galloyl catechin (Tanaka et al., 1983), 5-O-galloyl catechin (Al-Taweel et al., 2012) and 7-O-galloyltricitiflavan (Li et al., 2006) in conformity of the attachment of galloyl group at C-7 position. On comparison of the NMR spectra, compound 1 was found closely similar to that of 7-O-galloyl tricitiflavan (Li et al., 2006) except that in 1, OH-5' was absent, thus presenting compound 1 as a new compound, named as oncoglabinol A.

Compound **2** was obtained as colorless sticky material and its molecular formula, $C_{29}H_{22}O_{13}$ was determined by ESI-HRMS. The 1H and ^{13}C NMR spectra of compound **2** directed a close structural similarity with that of **1** except one additional signal in 1H NMR at δ_H 7.09 (s) and five carbon signals at δ_C 118.9, 109.6, 146.1 and 139.5, 164.9 in ^{13}C NMR and difference in chemical shifts of the B-ring of **2** as compared to that of **1** indicating the presence of one more galloyl moiety. The position of galloyl moiety at C-3' was established by the up field shift of C-3' (δ_C 138.9). Also due to the ester linkage at C-3' position, the C-2' appeared downfield at δ_C 121.8 which was in consistent with the previously reported data (El-Toumy and Mahdy, 2004). Compound **2** is also representing a new compound and named as oncoglabrinol B.

Compound **3** was obtained as colorless sticky material with the molecular formula, $C_{29}H_{22}O_{13}$ as determined by ESI-HRMS. The 1H and ^{13}C NMR spectra of **3** have a very close structural similarity with that of **2** except the position of galloyl moiety at C-4' in **3** instead of at C-3' as in **2**. Due to this C-1', C-2', C-5' and C-6' appeared at δ_C 140.2, 114.8, 123.7 and 117.2, respectively. This was also demonstrated by the down field shift of C3' (δ_C 149.4) which was in agreement with earlier described data having close structure similarities (El-Toumy and Mahdy, 2004). In this series compound **3** is also new and named as oncoglabin C. The configuration of C-2 for compounds **1-3** was defined as β , by comparing optical rotation values with that of 7-O-galloyltricetifavan and 7,4'-di-O-galloyltricetifavan (Li et al., 2006).

During the course of isolation of the above compounds, ethyl acetate extract of *O. glabratus* also yielded four known compounds. These compounds were identified by comparison of their physical and spectroscopic data with those reported in the literature. Upon subjecting compounds **1–6** to anti-HBV activity on cultured HepG2.2.15 cells by MTT assay at a concentration range of 6.25–100 µg/mL, all of them were non-cytotoxic and inhibited expression of HBsAg level at 100 µg/mL but not at 50 µg/mL. Nevertheless, all the compounds had mild antiviral activities (31–37% inhibition), compared to the standard drug lamivudine (Fig. 3). Although these compounds showed the anti-HBV activity at lower efficacy, further structural modifications may enhance their antiviral potential.

The ethyl acetate fraction of *O. glabratus* was found to show agonistic activity toward both PPAR α and PPAR γ . The activation of PPAR γ (5.35 fold) was stronger than the activation of PPAR α (2.6 fold) at 100 μ g/mL. Among the six compounds isolated from this fraction, two compounds **3** and **6** showed activation of both PPAR α and PPAR γ with a fold induction of 2 or higher at 50 μ g/mL while others were inactive. As shown in [Table 2](#) the effect was more pronounced on

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