

Cytotoxic and anti-inflammatory salicin glycosides from leaves of *Salix acmophylla*



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ABSTRACT

The species of genus *Salix*, commonly known as Willow, are well known worldwide as rich source of medicinally important salicin derivatives and phenolic glycosides. The current study focuses on *Salix acmophylla* Boiss with the aim of identifying new bioactive constituents of this plant. Two new salicin glycosides, acmophyllin A (**1**), acmophyllin B (**2**) and five reported phenolic glycosides **3–7**, were identified from *S. acmophylla* Boiss. NMR and mass spectroscopic techniques were employed to elucidate the structure of secondary metabolites of *S. acmophylla*. The new salicin glycosides were evaluated against three different cancer cell lines i.e., PSN-1 (pancreatic cancer cells), MCF-7 (breast cancer cells) and NCI-H460 (lung cancer cells). The acmophyllin A (**1**) exhibited cytotoxicity in a dose dependent manner against all three cancer cells ($IC_{50} \sim 35\text{--}40 \mu\text{M}$). Acmophyllin B (**2**) exhibited mild activity against PSN-1 cells and MCF-7 cancer cells. In addition, compounds **5** and **6** showed potent inhibition of oxidative burst in zymosan activated neutrophils by chemiluminescence technique, while no other compound were found to inhibit the production of reactive oxygen species (ROS).

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1. Introduction

The genus *Salix* (Willow) comprises of over 450 species which mainly occur in temperate and cold regions of Northern Hemisphere. Some species of *Salix* are well known herbs for their analgesic and anti-inflammatory activities. They have been used in rheumatic diseases, headache, fever and infections (Vainio and Morgan, 1997; Poblocka-Olech et al., 2010; Gawlik-Dziki et al., 2014; El-Shazly et al., 2012). *Salix* species are reported to contain many secondary metabolites such as terpenoids, flavonoids, phenolic acids, lignans, and salicin (natural aspirin) (El-Sayed et al., 2015; El-Shazly et al., 2012). Salicin derivatives have been found effective in treatment of pain, fever, and inflammation (Kim et al., 2015).

The plant *S. acmophylla* has wide occurrence in the Central Asia, eastern Turkey, Iran, Pakistan and Afghanistan (Iqbal et al., 2004). The methanolic extract of this plant was reported to have strong

phytotoxicity. The current investigation on ethyl acetate extract of this plant afforded two new and five known phenolic glycosides from the *S. acmophylla*. The cytotoxicity of new constituents of this plant was evaluated against three different cancer cell lines. The isolated constituents of plants were also evaluated for their anti-inflammatory potential.

2. Results and discussion

The new salicin derivatives acmophyllin A (**1**) and acmophyllin B (**2**) (Fig. 1) were obtained from the purification of crude extract of *S. acmophylla* Boiss as white powder. The molecular formula of **1** was determined from the quasi-molecular ion $[M+Na]^+$ at m/z 533.1 in the ESI-MS. The molecular formula $C_{27}H_{26}O_{10}Na$ was deduced from the observed molecular ion peak $[M+Na]^+$ in the HRESI-MS at m/z 533.1409 (Calc. 533.1423 for $C_{27}H_{26}O_{10}Na$) (Supporting information, Fig. S2). This indicated fifteen degrees of unsaturation in compound **1**. Eleven degrees of unsaturation were accounted for eleven double bonds, while, remaining four degrees of unsaturation were due to four rings of compound **1**. The IR spectrum of **1** showed absorption bands at 3413 cm^{-1} (OH),

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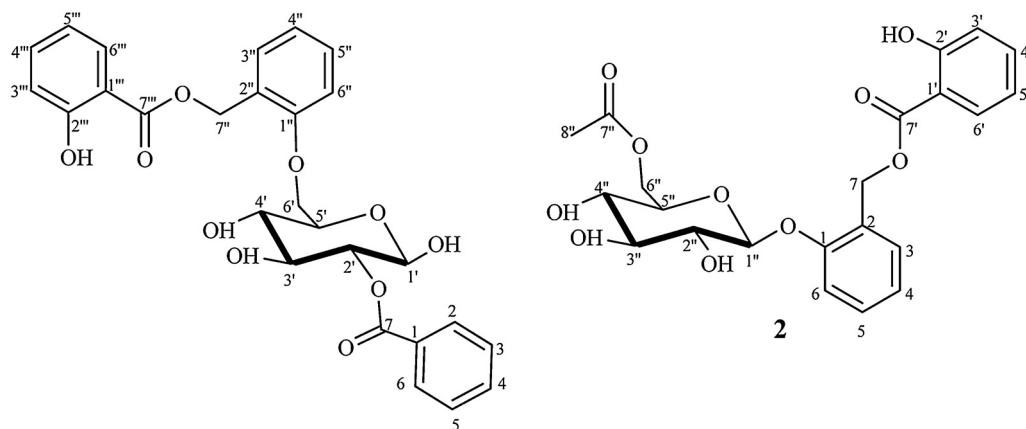


Fig. 1. The Structure of New Compounds 1 and 2.

2923 cm^{-1} (C—H due to asymmetric stretching), 1398 cm^{-1} , 1616 cm^{-1} (C=C of benzene ring). The UV spectrum exhibited absorption bands at λ_{max} (ϵ) at 307, 278, 237, 212 nm.

Analysis of ^1H NMR spectrum of **1** revealed the signals at δ_{H} 7.98 (2H, d, $J=7.6$ Hz, H-2, 6), 7.31 (2H, m, H-3, 5), and 7.43 (1H, m, H-4) which implied the presence of benzoyl group, while, the signals at δ_{H} 7.30 (1H, m, H-6''), 7.03 (1H, m, H-3'') and 7.31 (2H, H-4'', 5'') indicated salicylic alcoholic moiety. The remaining signals at δ_{H} 6.88 (1H, d, $J=8.4$ Hz, H-3'''), 7.41 (1H, m, H-4'''), 6.73 (1H, m, H-5''') and 7.55 (1H, d, $J=8.0$ Hz, H-6''') were assigned to *ortho*-hydroxybenzoyl group. The ^{13}C NMR spectral analysis of **1** revealed the presence of 27 carbons, including two methylene, eighteen methine and seven quaternary carbons (Table 1).

The structure of new compound **1** was fully supported by 2D-NMR spectral data such as HSQC, HMBC and COSY spectrum. In the HMBC spectrum (Fig. 2), the H-7'' methylene protons H_a δ_{H} 5.20, d, ($J=12.4$); H_b δ_{H} 5.15, d, ($J=12.4$) exhibited correlation with

quaternary carbons C-7''' (δ_{C} 170.8), C-1'' (δ_{C} 156.7), C-2'' (δ_{C} 125.9), which supported the linkage of salicylic alcoholic moiety with *ortho*-hydroxybenzoyl group via ester connection. Similarly, the correlation of aromatic proton H-6'' (δ_{H} 7.30) with C-6' (δ_{C} 62.4) methylene carbon of sugar unit, C-2'' (δ_{C} 125.9) and C-1'' (δ_{C} 156.7) supported the linkage of C-6' position of glucose moiety with salicylic alcoholic group. H-2' (δ_{H} 5.28) of sugar unit showed correlation with C-7 carbonyl carbon (δ_{C} 167.2) of benzoyl group.

In the COSY 45° spectrum (Fig. 2), correlations of H-2 and H-6 (δ_{H} 7.98) were observed with H-3 and H-5 (δ_{H} 7.31), respectively. While H-4 (δ_{H} 7.43) was found to be correlated with both H-3 and H-5 (δ_{H} 7.31). The overlapped signals of H-4'' (δ_{H} 7.31), H-5'' (δ_{H} 7.31), and H-6'' (δ_{H} 7.30) were correlated with H-3'' (δ_{H} 7.03) in the salicylic alcoholic group. The ^1H - ^1H COSY correlation of H-3''-H-4'', H-4''-H-5'' and H-5''-H-6'' supported the *ortho*-hydroxybenzoyl group in **1**. The ^1H NMR, ^{13}C NMR chemical shift, and COSY, HMBC connections for **1** are presented in Table 1.

Table 1
 ^1H - and ^{13}C NMR Spectral Data of Acrophyllin A (**1**) in CD_3OD .

C. No.	^1H - δ (J/Hz)	^{13}C (δ)	Multiplicity	HMBC Correlations	COSY Correlations
1	–	131.5	C		
2	7.98, d, ($J_{2,3}=7.6$)	130.9	CH	1, 4	3
3	7.31, m	129.4	CH		2
4	7.43, m	134.2	CH		3, 5
5	7.31, m	129.4	CH		4, 6
6	7.98, d, ($J_{5,6}=7.6$)	130.9	CH		5
7	–	167.2	C=O		
1'	5.29, m	100.7	CH		2'
2'	5.28, m	75.2	CH	1', 3', 7	1', 3'
3'	3.78, m	76.0	CH		2', 4'
4'	3.57, m	71.6	CH		3', 5'
5'	3.57, m	78.5	CH		4', 6'
6'	$\text{H}_a=3.96$, dd, ($J=11.2, 2.4$); $\text{H}_b=3.78$, overlap	62.4	CH_2		5'
1''	–	156.7	C		
2''	–	125.9	C		
3''	7.03, m,	123.7	CH		4''
4''	7.31, overlap	130.2	CH		3'', 5''
5''	7.31, overlap	131.0	CH		4'', 6''
6''	7.30, m	116.6	CH	2'', 3'', 6'	5''
7''	$\text{H}_a=5.20$, d, ($J=12.4$); $\text{H}_b=5.15$, d, ($J=12.4$)	63.0	CH_2	2'', 4'', 7'''	Geminal coupling
1'''	–	112.1	C		
2'''	–	162.6	C		
3'''	6.88, d, ($J_{3''', 4''}=8.4$)	118.2	CH	1'''	4'''
4'''	7.41, m	136.6	CH		3''', 5'''
5'''	6.73, m	120.2	CH	1'', 3'''	4''', 6'''
6'''	7.55, d, ($J_{5''', 6''}=8.0$)	130.5	CH		5'''
7'''	–	170.8	C=O		

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