



## Mini review

# *Dianthus erinaceus* var. *erinaceus*: Extraction, isolation, characterization and antimicrobial activity investigation of novel saponins



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## ABSTRACT

A phytochemical analysis of *Dianthus erinaceus* Boiss. var. *erinaceus* (Caryophyllaceae) has led to the isolation of two novel triterpenoid saponins, containing an oleanane type skeleton, named dianosides K and L (1, 2), along with six known triterpenoid saponins (3–8). On the basis of chemical and spectrometric data, the structures of the new compounds were elucidated as 3-O-[β-D-glucopyranosyl (1 → 3)]-β-D-glucopyranosyl-olean-12-ene-23α,28-β-dioic acid 28-O-β-D-glucopyranosyl ester (1) and 3-O-[β-D-glucopyranosyl (1 → 3)]-β-D-glucopyranosyl(1 → 6)-β-D-glucopyranosyl-olean-12-ene-23α,28-β-dioic acid 28-O-α-L-mannopyranosyl (1 → 6)-β-D-glucopyranosyl ester (2). All isolated natural compounds were structurally characterized by 1D- (<sup>1</sup>H, <sup>13</sup>C, DEPT); 2D- (COSY, HMQC, HMBC) NMR and HR-ESI/MS methods. The antimicrobial activity of compounds 1 and 2 were tested against four Gram-negative, three Gram-positive bacteria and the yeast *Candida albicans* by the MIC method.

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## Contents

1. Introduction	219
2. Results and discussion	220
3. Experimental	221
3.1. General experimental procedure	221
3.2. Plant material	221
3.3. Extraction and purification	221
3.4. Dianoside K (1)	223
3.5. Dianoside L (2)	223
3.6. Determination of sugar stereostructures	223
3.7. Antimicrobial assay	223
Conflict of interest	223
Acknowledgements	223
References	223

## 1. Introduction

The plant genus *Dianthus* L. (fam. Caryophyllaceae) includes more than 300 species mainly originating in Eurasia. Generally, these species grow as wild weeds but some are cultivated as ornamental plants. The *Dianthus* genera has approximately

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67 species in Turkey, 42 of which are endemic. *Dianthus erinaceus* Boiss. var. *erinaceus* is a perennial endemic plant which grows only on Spil Mountain, Manisa, western Turkey (Davis, 1967). Several *Dianthus* species have been medicinally used to tackle infections and diseases in China, Iran and Mongolia for millennia. Previous chemical investigations of *Dianthus* species revealed that plants belonging to this genera are a rich source of triterpenoid saponins (Chen et al., 2010; Koike et al., 1994; Nakano et al., 2011; Oshima et al., 1984a,b,c). In addition, there have been isolation studies of flavonoids (Obmann et al., 2012), pyrane type glycosides (Plouvier et al., 1986), macrocyclic anthocyanins (Nakayama et al., 2000) and cyclopeptides (Tong et al., 2012). Previous phytochemical and activity studies have demonstrated that isolated natural compounds from *Dianthus* species and the crude plant itself show antibacterial (Gou et al., 2011), antifungal (Galeotti et al., 2008), antioxidant (Durucasu et al., 2009), analgesic, antihepatotoxic (Hikino et al., 1984), cytotoxic (Yu et al., 2007) and proliferative (Tong et al., 2012) activities.

To the best of our knowledge, there have been no reports on the chemical composition or biological activity of *Dianthus erinaceus* var. *erinaceus*. Thus, this investigation focusses on *Dianthus erinaceus*, with the aim of analyzing its saponins. Purification procedures resulted in the isolation of two novel (dianosides K–L, **1–2**) and six known triterpenoid glycosides (**3–8**). The structures of the isolated compounds were identified using 1D (<sup>1</sup>H, <sup>13</sup>C, DEPT); 2D (COSY, HSQC, HMBC) NMR spectra and HR-ESI/MS. The antimicrobial effects of the new compounds (**1–2**) were examined against different microorganisms using MIC.

## 2. Results and discussion

Compound **1** was obtained as an amorphous, white powder. Its molecular formula was determined to be C<sub>54</sub>H<sub>86</sub>O<sub>25</sub> from its pseudo-molecular ion peak at *m/z* 1157.5352 (calcd. *m/z* 1157.5350) [M+Na]<sup>+</sup> in the HR-ESI/MS. Its IR spectrum revealed absorption bands at 3326 (OH), 1664 (C=O), 1657 (C=C) and 1020 cm<sup>−1</sup> (C–O–C). The <sup>1</sup>H NMR spectra of the aglycone section indicated the presence of six singlets at δ<sub>H</sub> 0.65, 0.82, 0.84, 0.85, 0.91, 1.07 of six tertiary methyl protons and a broad singlet at δ<sub>H</sub> 5.15 of an olefinic proton. The correlation in the heteronuclear single quantum coherence (HSQC) spectrum demonstrated the presence of six methyl carbons at δ<sub>C</sub> 13.2, 16.0, 17.1, 23.9, 26.0, 33.2, a pair of olefinic carbons at δ<sub>C</sub> 122.5 (CH) and δ<sub>C</sub> 143.8 (C) and two carboxylic groups at δ<sub>C</sub> 175.0 and 181.0. In addition, the heteronuclear multiple bond correlation (HMBC) of H-24 (δ<sub>H</sub> 0.91) and H-3 (δ<sub>H</sub> 3.76) of δ<sub>C</sub> 181.0, showed that the carboxyl carbon at δ<sub>C</sub> 181.0 could be assigned to C-23. Therefore, the other carboxyl carbon at δ<sub>C</sub> 175.0 was attributed to C-28. Thus, the aglycone was determined to be 3-β-hydroxyolean-12-en-23,28-dioic acid, which is also known as gypsogenic acid (Koike et al., 1994). The downfield <sup>13</sup>C NMR chemical shift at δ<sub>C</sub> 85.0 (C-3) and the upfield <sup>13</sup>C NMR chemical shift at δ<sub>C</sub> 175.0 (C-28) proved that **1** is a bis-desmosidic saponin with glycosidic linkages at C-3 through an *O*-heterosidic bond and at C-28 through an ester bond. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **1** contained four clear signals for anomeric protons and carbons at δ<sub>H</sub> 4.12 (d, *J*=7.2 Hz), 4.15 (d, *J*=7.8 Hz), 4.29 (d, *J*=7.2 Hz), 5.34 (d, *J*=7.2 Hz) and δ<sub>C</sub> 103.7, 104.0, 104.7, 93.7, respectively. All the proton signals for the sugar moieties were associated with one bond coupled with carbon signals using the HSQC spectrum. In the HMBC spectrum, the H-1 proton of glucose I at δ<sub>H</sub> 4.12 and the H-1 proton of glucose IV at δ<sub>H</sub> 5.34, showed long-range correlations with C-3 of the aglycone at δ<sub>C</sub> 85.0 and C-28 of the aglycone at δ<sub>C</sub> 175.0, respectively. On the other hand, long-range correlations between the H-1 proton of glucose II at δ<sub>H</sub> 4.15 and the C-6 carbon of glucose I at δ<sub>C</sub> 68.0; and the H-1 proton of glucose III at δ<sub>H</sub> 4.29 and the C-3 carbon of glucose I at δ<sub>C</sub> 88.6, showed the points at

which the sugar molecules are linked to each other. All sugar residues were identified as β-D-glucose by gas chromatography of the hydrolyzed product and by the coupling constant of their anomeric protons. Thus, compound **1** was confirmed to be 3-*O*-[β-D-glucopyranosyl (1→3)]-[β-D-glucopyranosyl (1→6)]-β-D-glucopyranosyl-olean-12-ene-23α,28-β-dioic acid 28-*O*-β-D-glucopyranosyl ester, namely dianoside K.

Compound **2** was shown to have the molecular formula C<sub>60</sub>H<sub>96</sub>O<sub>30</sub> on the basis of the HR-ESI/MS data at *m/z* [M+Na]<sup>+</sup> = 1319.5885 (calcd. 1319.5879). Comparison of the <sup>1</sup>H and <sup>13</sup>C NMR data (see Tables 1–2) of **2** with those of **1** showed considerable structural similarity except for the presence of one extra mannose moiety. The sugar part of **2** was found to consist of four glucose units and one mannose unit. The <sup>1</sup>H NMR chemical shift at δ<sub>H</sub> 4.16 (d, *J*=7.8 Hz), 4.19 (d, *J*=7.8 Hz), 4.33 (d, *J*=7.8 Hz), 4.65 (d, *J*=3.0 Hz), and 5.35 (d, *J*=7.8 Hz) are attributed to anomeric proton signals for the sugar part of the compound. The anomeric carbons

**Table 1**  
<sup>1</sup>H NMR data for compounds **1** and **2**.<sup>a–d</sup>

position	<b>1</b>	<b>2</b>
3	3.76, m	3.75, m
5	1.50, m	1.52, m
12	5.15, s	5.18, s
24	0.91, s	0.92, s
25	0.82, s	0.84, s
26	0.65, s	0.65, s
27	1.07, s	1.10, s
29	0.85, s	0.88, s
30	0.84, s	0.86, s
	Glc I at C-3	Glc I at C-3
1	4.12, d (7.2)	4.16, d (7.8)
2	2.92, m	2.97, m
3	3.40, m	3.44, m
4	2.90, m	2.94, m
5	3.16, m	3.15, m
6	3.60, 3.90, m	3.61, 3.90, m
	Glc II	Glc II
1	4.15, d (7.8)	4.19, d (7.8)
2	2.80, m	2.84, m
3	3.06, m	3.12, m
4	3.02, m	3.10, m
5	3.00, m	3.10, m
6	3.40, 3.62, m	3.42, 3.65, m
	Glc III	Glc III
1	4.29, d (7.2)	4.33, d (7.8)
2	3.02, m	3.05, m
3	3.10, m	3.21, m
4	3.30, m	3.12, m
5	2.94, m	2.95, m
6	3.42, 3.68, m	3.40, 3.70, m
	Glc IV at C-28	Glc IV at C-28
1	5.34, d (7.2)	5.35, d (7.8)
2	nd	3.23, m
3	3.40, m	3.44, m
4	3.04, m	3.06, m
5	3.18, m	3.20, m
6	3.30, 3.64, m	3.08, 3.60, m
		Man
1		4.65, d (3.0)
2		3.74, m
3		3.68, m
4		3.58, m
5		3.35, m
6		3.45, 3.52, m

<sup>a</sup> <sup>1</sup>H NMR data (δ) were measured in DMSO-*d*<sub>6</sub> at 600 MHz.

<sup>b</sup> Coupling constants (*J*) in Hz are given in parentheses.

<sup>c</sup> The assignments are based on COSY, HSQC and HMBC experiments.

<sup>d</sup> nd: not detected.

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