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Synthesis and characterization of symmetrical sulfur-fused polycyclic aromatic hydrocarbons with controlled shapes



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1. Introduction

Enormous interest exists in the development of π -electron materials because they exhibit electronic, optoelectronic, and magnetic properties that are suitable for their potential application in molecular electronics.^{1,2} Compared to π -conjugated polymers, polycyclic aromatic hydrocarbons (PAHs) possessing unique physicochemical characteristics that have served as active components of organic electronic devices such as field-effect transistors, lightemitting diodes, solar cells and supramolecular materials field.³ Compared with full carbon analogues, partially introduced the heteroatom, such as nitrogen,⁴ sulfur,⁵ as well as oxygen,⁶ into the aromatic skeleton of PAHs, will unexpectedly lead to dramatic changes in their physicochemical and/or electronic properties by changing such as polarity, electron density, dipole and band-gap of the molecule.⁷ Sulfur has a similar electronegativity (2.58) to that of carbon (2.55) in the graphitic layers,⁸ which is effective in modifying the electronic arrangement. Consequently, special emphasis is given to sulfur heteroatom PAHs (Scheme 1), and their unique electronic, optical, and redox properties and self-assembling properties are interesting.⁹ Herein, we are particularly interested in S-PAHs and reported the synthesis, characterization of

ABSTRACT

Here we report and establish a facile synthetic method for these unprecedented sulfur-fused polycyclic aromatic hydrocarbons (S-PAHs) with symmetrical structures (C_2 -rectangle and D_{6h} -hexagonal shape). Characterization by laser desorption/ionization time-of-flight mass spectrometry unambiguously validate the successful formation of these novel S-PAHs. The C_2 -rectangle and D_{6h} -hexagonal-shaped S-PAHs (**1** and **2**) are obtained by Scholl-type cyclodehydrogenation of the corresponding precursor using FeCl₃ as an oxidant, and their spectral characteristics, thermal and electrical properties are investigated. © 2017 Elsevier Ltd. All rights reserved.

symmetrical S-PAHs (1 and 2) with controlled shapes (Scheme 2). The C_2 -rectangle and D_{6h} -hexagonal-shaped S-PAHs (1 and 2) in which four or six double bonds in the periphery are thieno-fused, respectively. The structures of the S-PAHs (1 and 2) were characterized by MALDI-TOF MS, and their spectral characteristics, thermal and electrical properties were investigated.

2. Results and discussion

2.1. Synthesis of S-PAHs (1 and 2)

The C_2 -rectangle-shaped S-PAH (1) possesses four hydrophilic TEG chains on the around. To achieve the C_2 -symmetric functionalization of the core, a strategy with introduction of hydrophilic segments to compound **11** followed by oxidative cyclization was adopted (Scheme 3). 1,3-dibromo-5-(bromomethyl)-benzene was allowed to react with *p*-toluenesulfonylacetonitrile (**6**) in DMSO-Et₂O in the presence of 37% HCl/NaOH, yielding **7** in 65% yield.¹⁰ Subsequently, the reaction of 1,3-bis(3,5-dibromophenyl)-2propanone (**7**) with 1,2-bis(4-dodecylphenyl)-1,2-ethanedione (**8**)¹¹ in dioxane using Bu₄NOH afforded cyclopentadienone derivative **9** in 41% yield. Compound **9** was subjected to Diels-Alder reaction with bis(4-dodecylphenyl)acetylene (**10**),¹¹ giving **11** in high yield (71%). Then, the Suzuki-coupling reaction between **11** and thiophene- α -boronic esters **5** afforded the precursor **12** in 36% yield. Finally, the Oxidative cyclization of **12** by FeCl₃/CH₃NO₂ in



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Scheme 1. Molecular structures of S-PAHs with specific shapes.



Scheme 2. Molecular structures of 1 and 2.



Scheme 3. Synthetic route of 1.

 CH_2Cl_2 resulted in the formation of the fully cyclodehydrogenated **1** in 68% yield.

The D_{6h} -hexagonal-shaped S-PAH (**2**) possesses six hydrophilic TEG chains in the periphery. To achieve the D_{6h} -symmetric functionalization of the core, a strategy with introduction of hydrophilic segments to **22** followed by oxidative cyclization was adopted (Scheme 4). 2,2'-(1,2-ethynediyldi-3,1-phenylene)bis[5-(trime-thylsilyl)-thiophene] (**20**) was synthesized using thiophene and 3-



Scheme 4. Synthetic route of 2.

bromoiodobenzene as starting materials. The reaction of thiophene with trimethylchlorosilane and further reaction with 2-isopropoxy-4,4,5,5-tetramethyl-1,3,2-dioxa-borolane afforded 4,4,5,5tetramethyl-2-[5-(trimethylsilyl)-2-thienyl]-1,3,2-dioxaborolane (14).¹² Next, the Suzuki-coupling reaction between 14 and bis(3bromophenyl) acetylene $(19)^{13}$ in THF using Pd(PPh₃)₄/K₂CO₃ afforded 20 in 80% yield. Then the cobalt-catalyzed cyclotrimerization of 20, giving 2,2'-[2',3',5',6'-tetrakis[3-[5-(trimethylsilyl)-2-thienyl]phenyl][1,1':4',1"-terphen-yl]-3,3"-diyl]-bis[5-(trimethylsilyl)-thiophene] (21) in 80% yield, and further substitution with iodine monochloride afforded 22 in 62% yield. Next, the Suzuki-coupling reaction between 22 and [4-[2-[2-(2-Methoxyethoxy)-ethoxy]ethoxy]phenyl]boronic acid (17)¹⁴ was carried out in toluene using Pd(PPh_3)4/K2CO3, affording the precursor 23 in 58% yield. Finally, the Oxidative cyclization of 23 by FeCl₃/CH₃NO₂ in CH₂Cl₂ resulted in the formation of **2** in 60% yield. However, the Suzuki-coupling reaction between 22 and 17 generated a small amount of five sites substitution impurity, which has brought difficulty to the purification (Fig. S1). Therefore, we decided to give up this synthetic route and explore the better one.

Eventually, we established a facile and rapid synthetic method for S-PAH **2** (Scheme 5). Bis(3-bromophenyl) acetylene was allowed to react with bispinacolatebiboron in DMF in the presence of PdCl₂(dppf)/AcOK, yielding **25** in 70%. And then the Suzukicoupling reaction between **25** and **28** in toluene using Pd(PPh₃)₄/ K₂CO₃ gave **29** in 65% yield. The cobalt-catalyzed cyclotrimerization of **29**, giving the precursor **23** in 78% yield, which was transformed into the target compound **2** by cyclodehydrogenation in 60% yield.

The recording of NMR spectroscopy did not succeed for S-PAHs (**1** and **2**) due to the aggregation propensity of the molecules in solution,¹⁵ so they were characterized by MALDI-TOF mass spectrometry. MALDI-TOF mass spectra showed the molecular ion-peak at 2164.3990 for **1** (Fig. 1a) and 2433.5918 for **2** (Fig. 1b) respectively, which coincided with their theoretical value of molecular

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