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Rhodium(III)-catalyzed heteroatom-directed C—H allylation with allylic phosphonates and allylic carbonates at room temperature



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ABSTRACT

The rhodium(III)-catalyzed mild and site-selective C—H allylation of 2-arylbenzo[d]thiazoles and 2-arylbenzo[d]oxazoles with allylic phosphonates and allylic carbonates is described. This transformation provides an efficient construction of C2-allylated, crotylated and prenylated 2-arylbenzo[d] thiazoles and 2-arylbenzo[d]oxazoles. In addition, this protocol can be applied to the formation of 2-arylbenzo[d]thiazole scaffolds containing an allylic alcohol group by using of 4-vinyl-1,3-dioxolan-2-one and vinyl oxirane as coupling partners.

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1. Introduction

2-Arylbenzoldlthiazoles are common structural motifs found in heterocyclic compounds with biological and medicinal applications including anticancer,¹ antibacterial,² potassium channels activation,³ neurotransmission blockage,⁴ and neuroprotection.⁵ They also serve as versatile building blocks for organic light-emitting diodes (OLEDs),6 chemosensors,7 and photosensitizers.8 Therefore, the development of efficient protocols for the functionalization of these heterocyclic architectures are of great interest in organic synthesis.9 Recently, the transition-metal-catalyzed C-H functionalization of 2-arylbenzothiazoles and 2-arylbenzoxazoles with diverse coupling partners has been investigated. In this area, arylation, ¹⁰ acetoxylation, ¹¹ acylation, ¹² hydroxylation, ¹³ and halogenations ¹⁴ under palladium catalysis were explored. In addition, the ruthenium-catalyzed olefination 15 and amination 16 reactions of 2-arylbenzothiazoles were also examined. Moreover, the rhodiumcatalyzed C-H alkylation reactions of 2-arylbenzothiazoles with α diazo compounds were reported.¹⁷ However, to the best of our knowledge, the direct and catalytic C-H allylation of 2arylbenzothiazoles and 2-arylbenzoxazoles with allylic compounds has been unexplored.

Catalytic C-H allylation reaction has recently emerged as a versatile tool to deliver structurally intricate organic molecules. ¹⁸ For example, Oi and Inoue first described the Ru(II)-catalyzed C-H allylation of 2-phenylpyridines with allylic acetates providing a regioisomeric mixture of olefin products. 18a Later, Glorius reported beautiful works on the Rh(III)- or Co(III)-catalyzed regioselective terminal allylations of benzamides and indoles with allylic carbonates. 18c,d Direct C-H allylations of electron-deficient polyfluoroarenes with allylic phosphonates and allylic carbonates under Cu(0),^{18f} Cu(I)^{18g} and Pd(II)^{18h} catalysis were respectively reported. In addition, allenes were used in the Ir(I)-, 18i Rh(I)-, 18i and Rh(III) 18k catalyzed allylation reactions to give allylated benzamide adducts. More recently, Li¹⁸¹ and Wang^{18m} independently applied vinyl oxiranes and 4-vinyl-1,3-dioxolan-2-ones into aryl C-H allylation reaction under rhodium catalysis affording aromatic products with allylic alcohol moieties.

Inspired by our recent studies on the rhodium-catalyzed C—H functionalization of (hetero)aromatic compounds ¹⁹ and in consideration of the biological application of functionalized 2-arylbenzothiazoles, we herein present the Rh(III)-catalyzed allylation, crotylation and prenylation of 2-arylbenzothiazoles and 2-arylbenzoxazoles with allylic phosphonates or allylic carbonates to afford *ortho*-allylated 2-arylbenzothiazoles and 2-arylbenzoxazoles via C—H bond activation.

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2. Results and discussion

In a previous literature, we found that the combination of [Cp*RhCl₂]₂, AgSbF₆ and Cu(OAc)₂ in DCE solvent was most effective catalytic system to couple with indoline C-H bonds and allylic substrates.²⁰ Thus, we used above reaction conditions in our initial study to couple with 1a and allyl acetate (2a) (Table 1), but our desired allylated product 3a was formed in 13% yield (Table 1, entry 1). Additionally, allylic carbonate **2b** did not also provide a reasonable yield (Table 1, entry 2). Further investigation revealed that allylic phosphonate **2c** as a coupling partner is unique in its ability to facilitate high levels of conversion (Table 1, entry 3). However, cationic ruthenium and cobalt catalysts were found to be ineffective for this transformation (Table 1, entries 4 and 5). Also, exclusion of either [Cp*RhCl₂]₂ and AgSbF₆ resulted in no observation of the desired product 3a (data not shown). Screening of solvents under otherwise identical conditions revealed that THF is found to be an optimal solvent to furnish 3a in 76% yield, but other solvents such as toluene, MeCN, and t-AmOH were less effective (Table 1, entries 6–9). Further screening of additives revealed that Cu(OAc)₂ was found to be the most effective in this coupling reaction (Table 1, entries 10 and 11). Furthermore, decreasing amount of Cu(OAc)₂ to 50 or 30 mol % provided comparable yields (Table 1, entries 12 and 13). Finally, the coupling reaction was performed under a nitrogen atmosphere leading to a comparable yield (83%) of **3a** (Table 1, entry 14). This result indicate that a role of Cu(OAc)₂ as an oxidant can be ruled out in the catalytic cycle.

To evaluate the scope and limitation of this process, the optimal reaction conditions were applied to a range of 2-aryl substituted heteroarenes **1b–1o** (Table 2). The reactions of *meta-*substituted 2-arylbenzothiazoles **1b–1d** were found to be favored for this transformation to afford the desired products **3b–3d** in moderate to good yields. Particularly noteworthy were the mono-selectivity and site-selectivity found at the less hindered position, as well as the tolerance of the reaction conditions to the bromo moiety, providing a versatile synthetic handle for further cross-coupling reactions. However, highly electron-rich 2-arylbenzothiazole **1e** at the *meta*-position was found to be relatively less reactive under the

present reaction conditions. This reaction was also compatible with *ortho*-substituted 2-arylbenzothiazole **1f** to furnish **3f** in 74% yield. In addition, symmetric 2-phenylbenzothiazole (**1g**) was coupled with **2c** under the optimal reaction conditions, resulting in a mixture of bis-allylated product **3g** and mono-allylated product **3ga** with 1:1 ratio in 51% combined yield. Logically, it was thought that the ratio of **3g** and **3ga** can be controlled by the amount of allylic phosphonate. Indeed, upon use of 3 equiv of **2c**, the bis-allylated compound **3g** was obtained as a major compound in 63% combined yield, albeit resulting in a low level of bis-selectivity (2:1). Moreover, 2-(p-methoxyphenyl)benzothiazole (**1h**) underwent smooth the bis-allylation reaction to afford the corresponding product **3h** in 61% yield.

However, 2-(4-fluorophenyl)benzothiazole (1i) gave the bisallylated compound 3i in 37% yield in conjunction with monoallylated compound 3ia in 36% yield. In sharp contrast, 2-phenylbenzoxazole (1j) displayed a significant bis-selectivity under the identical reaction conditions to furnish the corresponding bis-allylated product 3j in 61% yield, and a trace amount of monoallylated product was observed by ¹H NMR or GC-MS analysis. In addition, *meta*-substituted 2-arylbenzoxazole 1k was found to be a good substrate in this transformation. Furthermore, we were pleased to observe the allylation reaction at a vinyl C-H bond, which provided the corresponding product 3l in 83% yield. Finally, this reaction was found to be comparable with 2-arylthiazoles 1m and 1n, but in the case of 2-arylbenzimidazole 1o, a relatively low amount of product 3o was formed.

To further explore the scope and limitation of this transformation, substituted allylic phosphonates and allylic carbonates 2d-2g were screened to couple with 2-arylbenzothiazole 1a and 2-arylbenzoxazoles 1j and 1k, as shown in Table 3. In sharp contrast to results of allylation reaction with allyl methyl carbonate (2b), both α -methyl-substituted allylic phosphonate 2d and α -methyl-substituted allylic carbonate 2e provided a crotylation product 4a in high yield. In addition, allyl octyl carbonate 2f was smoothly coupled with 1a to give a diastereomeric mixture of crotylation product 4b in 42% yield. Notably, these crotylation reactions proceeded readily with complete γ -selectivity in case of branched allylic

Table 1Selected optimization of reaction conditions^a

Entry	Allyl source	Catalyst (mol %)	Additive (mol %)	Solvent	Yield (%) ^b
1	2a	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	DCE	13
2	2b	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	DCE	24
3	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	DCE	52
4	2c	$[Ru(p-cymene)Cl_2]_2$ (2.5)	Cu(OAc) ₂ (100)	DCE	35
5	2c	$[CoCp^*(CO)I_2]$ (5)	Cu(OAc) ₂ (100)	DCE	N.R.
6	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	Toluene	Trace
7	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	MeCN	31
8	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	t-AmOH	52
9	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (100)	THF	76
10	2c	[RhCp*Cl ₂] ₂ (2.5)	CsOAc (100)	THF	Trace
11	2c	[RhCp*Cl ₂] ₂ (2.5)	PivOH (100)	THF	38
12	2c	$[RhCp^*Cl_2]_2$ (2.5)	Cu(OAc) ₂ (50)	THF	92
13	2c	[RhCp*Cl ₂] ₂ (2.5)	Cu(OAc) ₂ (30)	THF	78
14 ^c	2c	[RhCp*Cl ₂] ₂ (2.5)	$Cu(OAc)_2$ (50)	THF	83

^a Reaction conditions: 1a (0.3 mmol), 2a-2c (0.6 mmol), catalyst (quantity noted), AgSbF₆ (10 mol %), additive (quantity noted), solvent (1 mL) under air at room temperature for 20 h in reaction tubes.

^b Isolated yield by flash column chromatography.

^c Under N₂ atmosphere.

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