Tetrahedron 70 (2014) 9668-9675

Contents lists available at ScienceDirect

Tetrahedron

journal homepage: www.elsevier.com/locate/tet

Reductive coupling of isatins with ketones and aldehydes by low-valent titanium

Naoki Kise*, Kouta Sasaki, Toshihiko Sakurai

Department of Chemistry and Biotechnology, Graduate School of Engineering, Tottori University, 4-101, Koyama-cho Minami, Tottori 680-8552, Japan

A R T I C L E I N F O

Article history: Received 4 September 2014 Received in revised form 25 October 2014 Accepted 28 October 2014 Available online 8 November 2014

Keywords: Reductive coupling Low-valent titanium Isatins 3-Hydoxy-3-(1-hydoxyalkyl)oxindoles 3-Alkylideneoxindoles

ABSTRACT

The reductive coupling of isatins with ketones and aldehydes by Zn–TiCl₄ in THF gave two- and fourelectron reduced products, 3-hydoxy-3-(1-hydoxyalkyl)oxindoles and 3-alkylideneoxindoles, selectively by controlling the reaction conditions. Although the 3-(1-hydoxyalkyl)oxindoles were also produced as the four-electron reduced products in some cases, these products were readily dehydrated to 3-alkylideneoxindoles. The 3-alkylideneoxindoles derived from aldehydes were formed as mixtures of geometric isomers. The both geometric isomers were isomerized to the equilibrium mixtures by reflux in cat. PPTS/benzene.

© 2014 Elsevier Ltd. All rights reserved.

1. Introduction

Recently the synthesis and reaction of 3-alkylideneoxindoles attract much attention, since this type of compound exists in a number of biologically important compounds and is a useful precursor for the synthesis of other types of oxindoles.^{1–3} Although many methods have been reported to date for the synthesis of 3alkylideneoxindoles,^{1a,2} the use of the reductive cross-coupling of isatins with carbonyl compounds is unprecedented for this purpose. Reductive coupling with low-valent titanium has been well known as a powerful tool for the reductive cross-coupling between two different carbonyl compounds because of its versatility, convenience, and economical efficiency.^{4,5} In the course of our recent study on the reductive cross-coupling with low-valent titanium,⁶ we report in this paper the reductive coupling of isatins with ketones and aldehydes by low-valent titanium generated from Zn–TiCl₄ (Scheme 1). We found that two- and four-electron reduced products, 3-hydoxy-3-(1-hydoxyalkyl)oxindoles and 3alkylideneoxindoles, could be prepared selectively by controlling the reaction conditions. In some cases, the four-electron reduced products were obtained as mixtures of two products, 3alkylideneoxindoles and 3-(1-hydoxyalkyl)oxindoles. However, the latter products were easily transformed to the former by reflux in cat. PPTS/benzene. Therefore, the reductive coupling of isatins with carbonyl compounds provides new synthetic method for already known 3-alkylideneoxindoles and hitherto unknown 3hydoxy-3-(1-hydoxyalkyl)oxindoles, which are promising precursors of other oxindole heterocycles.



Scheme 1. Reductive coupling of isatins with ketones and aldehyde by Zn-TiCl₄.

2. Results and discussion

2.1. Reductive coupling of isatins with ketones by Zn–TiCl₄

The reaction conditions of the cross-coupling were scrutinized using N-methylisatin (**1a**) and acetone (**2a**) as the substrates and





Tetrahedror

^{*} Corresponding author. E-mail address: kise@bio.tottori-u.ac.jp (N. Kise).

the results are summarized in Table 1. The molar ratio of Zn/TiCl₄ was fixed to 2/1. Initially, the reaction was carried out with the molar ratio of **1a**/TiCl₄ as 1/1 (condition a) in THF at 0 °C for 1 h with varying the molar ratio of **1a/2a** from 1/1 to 1/5 (runs 1–4). In these cases. 3-hydroxy-3-(2-hydroxypropan-2-yl)-1-methylindolin-2one (3a) was obtained in 64–74% vields as the two-electron reduced product together with small amounts of the four-electron reduced products. 1-methyl-3-(propan-2-ylidene)indolin-2-one (4a) and 3-(2-hydroxypropan-2-yl)-1-methylindolin-2-one (5a). With the ratio of **1a/2a** as 1/5 (run 4), **3a** was obtained as the sole product, although the yield was somewhat lowered (64%) and 1a was recovered (15%). Therefore, the molar ratio of 1a/TiCl₄ was increased to 1/2 (condition b) to complete the reaction (run 5). This attempt increased the yield of **3a** (71%) and, however, brought about the production of small amounts of **4a** (10%) and **5a** (5%). Consequently, the best yield of **3a** (74%) was obtained with the 1/2ratio of 1a/2a (run 2). Next, the reaction was performed with the ratio of 1a/2a/Zn/TiCl₄ as 1/2/4/2 (condition b) in THF at 0 °C, 20 °C and 30 °C for 2 h (runs 6–8). From the result in run 6, it found that the further reduction to 4a and 5a proceeded considerably even at 0 °C under condition b. Four-electron reduced product 4a was formed at 30 °C in 70% yield together with 5a (17%), and 3a was not obtained (run 8). It was found that dehydration of 5a gave 4a quantitatively (vide infra). Therefore, crude product mixture obtained from run 8 was refluxed in cat. PPTS/benzene for 1 h (condition c). Expectedly, 4a was formed as the sole product in 87% yield (run 9). When isatin (1b) was employed in place of 1a, 3b (80%) and **4b** (89%) were produced under the same conditions as runs 2 and 9. respectively (runs 10 and 12).

Next, the reductive coupling of **1a**,**b** with aliphatic cyclic ketones **2b**–**e** was carried out under the same conditions as runs 2, 7 and 8 (conditions a, b and c) in Table 1 (Table 2). In most cases, 3-hydoxy-

Table 1



^a a: $1/Zn/TiCl_4=1/2/1$, 1 h. b: $1/Zn/TiCl_4=1/4/2$, 2 h. c: crude product mixture obtained by condition b was refluxed in cat. PPTS/benzene for 1 h.

^b Isolated yields.

3-(1-hydoxyalkyl)oxindoles **3** were produced selectively in good yields under the condition a. However, the reaction of **1a** with cycloheptanone (2d) brought about 3e in a low yield (38%) probably due to the low reactivity of 2d at 0 °C (run 6). Especially in the reaction of 1b with 2d (run 15), no cross-coupled product could be detected. In these cases, the major products were 3hvdroxvindolin-2-ones 6a.b. The products 3f and 3i were obtained from 4-t-butylcvclohexanone (2e) as the single diastereomers (runs 8 and 17) and their stereo structures were confirmed to be cis by X-ray crystallography (Fig. 1). These results show that the less-hindered equatorial attack of 1a,b to 2e proceeded predominantly. In almost all cases, 3-alkylideneoxindoles 4 were formed selectively in good to moderate yields under the condition c. Even in the reactions with 2d (runs 7 and 16), 4e and 4i were formed in moderate yields (41% and 64%). In these cases, the major by-products were indolin-2-ones 7a.b. These results show that the cross-coupling of 1a,b with 2d proceeded at 30 °C (condition b). Incidentally, the reduction of **1a**,**b** in the absence of **2** under the conditions a and b gave the corresponding two- and fourelectron reduced products, **6a**,**b** and **7a**,**b**, as shown in Scheme 2.⁷ In addition, the reduction of **1a** with benzophenone or acetophenone also gave **6a** and **7a**, and the corresponding cross-coupling products could not be obtained.

Table 2

Reductive coupling of 1a,b with 2b-e by Zn-TiCl₄



3c-f: X = Me **3g-j**: X = H **5c-f**: X = Me **5g-j**: X = H

4c-f: X = Me

4g-j: X = H

| Run | 1 | 2 | Condition ^a | % Yield ^b | | | |
|-----|----|----|------------------------|----------------------|-----------------|----|----|
| | | | | | 3 | 4 | 5 |
| 1 | 1a | 2b | a | с | 74 | 6 | 5 |
| 2 | 1a | 2b | b | с | — | 56 | 34 |
| 3 | 1a | 2b | с | с | — | 85 | _ |
| 4 | 1a | 2c | a | d | 82 | — | _ |
| 5 | 1a | 2c | с | d | _ | 77 | |
| 6 | 1a | 2d | a | e | 38 | 7 | _ |
| 7 | 1a | 2d | с | e | — | 41 | _ |
| 8 | 1a | 2e | a | f | 75 | — | _ |
| 9 | 1a | 2e | с | f | — | 72 | _ |
| 10 | 1b | 2b | a | g | 79 | 8 | 4 |
| 11 | 1b | 2b | b | g | — | 71 | 22 |
| 12 | 1b | 2b | с | g | — | 92 | _ |
| 13 | 1b | 2c | a | h | 72 | — | _ |
| 14 | 1b | 2c | с | h | — | 66 | _ |
| 15 | 1b | 2d | a | i | ND ^c | _ | _ |
| 16 | 1b | 2d | с | i | — | 64 | _ |
| 17 | 1b | 2e | a | j | 64 | _ | _ |
| 18 | 1b | 2e | с | j | — | 61 | — |

^a a: $1/2n/TiCl_4=1/2/1$, 0 °C, 1 h. b: $1/2n/TiCl_4=1/4/2$, 30 °C, 2 h. c: crude product mixture obtained by condition b was refluxed in cat. PPTS/benzene for 1 h. ^b Isolated yields.

^c Not detected.

Download English Version:

https://daneshyari.com/en/article/5216090

Download Persian Version:

https://daneshyari.com/article/5216090

Daneshyari.com