



# A method for the regioselective synthesis of 1-alkyl-1*H*-indazoles



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## ABSTRACT

A method for the regioselective synthesis of 3-unsubstituted 1-alkyl-1*H*-indazoles, starting with 2-halobenzonitriles and *N*-alkylhydrazines, is described. The two-step reaction pathway proceeds through the intermediacy of 1-alkyl-3-amino-1*H*-indazoles followed by reductive deamination.

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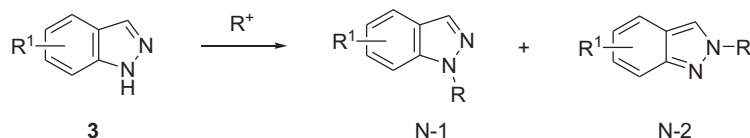
## 1. Introduction

The indazole ring system is recognized to be a highly effective pharmacophore in medicinal chemistry as well as being the core of important nitrogen-containing heterocycles that show a broad range of biological activities, such as nitric oxide synthase<sup>1</sup> and HIV protease<sup>2</sup> inhibitors, anti-inflammatory,<sup>3</sup> antitumor,<sup>4</sup> and anti-cancer<sup>5</sup> agents, and serotonin 5-HT<sub>3</sub> receptor antagonists.<sup>6</sup> A variety of methods for the preparation of indazoles have been reported.<sup>7</sup> The developed approaches include diazotization of 2-alkylaniline derivatives followed by cyclization under basic conditions,<sup>8</sup> base promoted cyclization of (*o*-alkylaryl)azosulfides derived from 2-alkylaniline derivatives,<sup>9</sup> [3+2] cycloadditions of arynes with diazo compounds or hydrazones,<sup>10</sup> condensation reactions of *o*-haloaryl carbonyls or salicylaldehydes with hydrazine,<sup>11</sup> and cyclization reactions of *o*-aminobenzoximes in the presence of bases.<sup>12</sup> Although methods for regioselective synthesis of indazoles have been described,<sup>12–20</sup> only a few of these are

applicable to the regioselective synthesis of 1-alkyl-1*H*-indazoles.<sup>13,14,15b,16,20</sup> Thus, methods for the regioselective synthesis of 1-alkyl-1*H*-indazoles remain in demand.

## 2. Results and discussion

As a part of a recent research effort, we required 3-unsubstituted 1-alkyl-1*H*-indazoles as synthetic intermediates. The most straightforward route to access these substances involves treatment of 3-unsubstituted 1*H*-indazoles with alkylating agents. However, the regiochemistry of these processes is highly dependent on the nature of alkylating agent and, in general, mixtures of *N*-1 and *N*-2 alkylated products are typically produced (Eq. 1). For example, methylation of 5-nitro-1*H*-indazole using iodomethane (NaH, THF, 0 °C, 2 h) results in formation of a mixture of 5-nitro-1-methyl-1*H*- and 5-nitro-2-methyl-2*H*-indazoles in a 55:45 ratio. Moreover, varying the solvent, temperature and base employed in this reaction failed to improve the selectivity.



Eq (1)

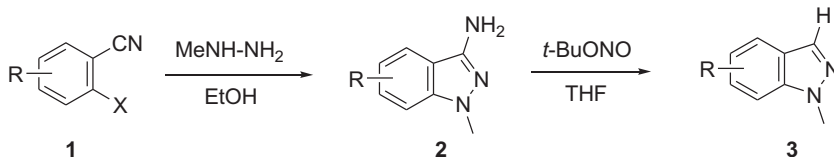
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A common method for the preparation of 3-substituted 1-alkyl-1*H*-indazoles involves cyclization of an arylhydrazone, derived

from reaction of an arylketone possessing a leaving group in the *ortho*-position with an alkylhydrazine.<sup>14,16</sup> We envisioned that 3-unsubstituted 1-alkyl-1*H*-indazoles could be generated utilizing an analogous process in which an arylketone is replaced by an arylaldehyde. However, treatment of methylhydrazine and a mesylate, derived from reaction of 5-bromo-2-hydroxy-benzaldehyde with mesyl chloride, in refluxing xylene in the presence of  $\text{NH}_4\text{OAc}$  for 3 h gave only the corresponding hydrazone product, and an increase in the time for this reaction to 14 h resulted in the formation of multiple products.

Owing to these observations, our attention turned to the use of other substrates. 3-Aminoindazoles, obtained by using a variety of methods,<sup>21–23</sup> are known to have valuable biological activities in several therapeutic areas.<sup>24</sup> Wheeler et al. described a procedure for the regioselective synthesis of 3-amino-1-methyl-1*H*-indazoles utilizing reactions of 2-fluorobenzonitriles with methylhydrazine.<sup>20</sup> We hypothesized that this process, when coupled with the novel reductive deamination reaction of arylamines that produces aromatic hydrocarbons described by Doyle and Piccinello et al.,<sup>24,25</sup> would serve as a regioselective two-step method for the preparation of 3-unsubstituted 1-alkyl-1*H*-indazoles. The viability of this proposal was demonstrated in the studies described below.

As the results displayed in Scheme 1 and Table 1 show, reactions of members of a series of 2-halobenzonitriles, containing either electron-donating or -withdrawing groups, with methylhydrazine in ethanol efficiently produce the corresponding 3-amino-1-methyl-1*H*-indazoles **2**. Reductive deamination reactions of the 3-amino-1-methyl-1*H*-indazole products **2** with *tert*-butyl nitrite in either  $\text{CHCl}_3$ , DMF or THF generated 3-unsubstituted 1-methyl-1*H*-indazoles **3** in high yields. It should be noted that the deamination reaction of 3-amino-1-methyl-5-nitro-1*H*-indazole (**2e**) in THF or  $\text{CHCl}_3$  gave a product that was expected to be 1-methyl-5-nitro-1*H*-indazole (**3e**). In order to confirm its structure, this substance was subjected to catalytic hydrogenation, which yielded 1-methyl-5-amino-1*H*-indazole.



Scheme 1. Regioselective synthesis of 1-methyl-1*H*-indazoles via 1-methyl-3-amino-1*H*-indazoles.

In order to determine if this methodology is suitable for the preparation of other 1-alkyl-1*H*-indazoles, reactions of ethyl hydrazine and benzyl hydrazine with the respective benzonitriles **1b** and **1e** were examined. As the results displayed in Scheme 2 show, application of the two-step procedure gave the corresponding 1-ethyl- (**3j**) and 1-benzyl- (**3i**) 1*H*-indazoles in high yields. Other hydrazines, such as phenylhydrazine ( $\text{Ph-NH-NH}_2$ ) and isopropylhydrazine ( $\text{Me}_2\text{CH-NH-NH}_2$ ) were examined and both reactions gave a mixture of multiple products. Based on the results, a plausible mechanism is proposed and shown in Scheme 3.

### 3. Conclusion

In summary, the study described above has resulted in the development of an alternative, two-step method for the regioselective synthesis of 3-unsubstituted 1-alkyl-1*H*-indazoles starting with *N*-alkylhydrazines and 2-halobenzonitriles.

## 4. Experimental section

### 4.1. General

All commercially available chemicals were used without further purification. TLC analyses were run on a TLC glass plate (Silica gel 60  $\text{F}_{254}$ ) and were visualized using UV and a solution of phosphomolybdic acid in ethanol (5 wt %) or *p*-anisaldehyde stain. Flash chromatography was performed using silica gel (70–230 mesh).  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra were recorded on a 300 MHz spectrometer. Chemical shifts are reported relative to  $\text{CHCl}_3$  [ $\delta_{\text{H}}$  7.24,  $\delta_{\text{C}}$  (central line) 77.0]. Mass spectra were recorded under fast atom bombardment (FAB) or electron impact ionization (EI) conditions. High-resolution mass spectra were recorded by electron impact ionization with a magnetic sector analyzer.

### 4.2. Synthesis

**4.2.1. General procedure for synthesis of 3-amino-1-methyl-1*H*-indazole **2**.** A mixture of benzonitrile **1** (10.0 mmol) and methylhydrazine (2.8 mL, 50.0 mmol) in EtOH (10.0 mL) was heated to reflux overnight. The mixture was cooled to rt and then concentrated.  $\text{H}_2\text{O}$  (10.0 mL) and EtOAc (20.0 mL) were added to the residue. The organic layer was washed with  $\text{H}_2\text{O}$  (10.0 mL), brine (10.0 mL), dried over  $\text{Na}_2\text{SO}_4$ , filtered, and concentrated in vacuo. The residue was subjected to silica-gel chromatography by using EtOAc/hexanes (1:1) as eluent to give the product **2**.

**4.2.1.1. 1-Methyl-1*H*-indazol-3-ylamine (**2a**).** As described in the general procedure, reaction of 2-fluorobenzonitrile **1a** (1.21 g, 10.0 mmol) and methylhydrazine (2.8 mL, 50.0 mmol) in EtOH (10.0 mL) afforded the title compound (1.35 g, 92%). Solid (EtOAc/hexanes=3:1), mp 94–95 °C; TLC (EtOAc/hexanes (1:1))  $R_f$ =0.2;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  3.79 (s, 3H), 4.13 (br s, 2H), 6.96 (dd,  $J$ =8.6, 7.2 Hz, 1H), 7.15 (d,  $J$ =8.6 Hz, 1H), 7.30 (dd,  $J$ =8.0, 7.2 Hz, 1H), 7.47 (d,  $J$ =8.0 Hz, 1H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CDCl}_3$ )  $\delta$  34.6 ( $\text{CH}_3$ ), 108.5 (CH),

114.3 (C), 118.2 (CH), 119.4 (CH), 126.7 (CH), 141.3 (C), 146.9 (C); MS  $m/z$  (rel intensity) 148 ( $\text{M}^+ + \text{H}$ , 100), 133 (5). These data are in agreement with those reported in the literature.<sup>26</sup>

**4.2.1.2. 4-Fluoro-1-methyl-1*H*-indazol-3-amine (**2b**).** As described in the general procedure, reaction of 2,6-difluorobenzonitrile **1b** (1.39 g, 10.0 mmol) and methylhydrazine (2.8 mL, 50.0 mmol) in EtOH (10.0 mL) afforded the title compound (1.42 g, 86%). Solid (EtOAc/hexanes=1:1), mp 125–126 °C; TLC (EtOAc/hexanes (1:1))  $R_f$ =0.2; IR (neat) 3438, 3308, 3206, 1634  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  3.77 (s, 3H), 4.12 (br s, 2H), 6.54 (dd,  $J$ =9.0, 6.0 Hz, 1H), 6.88 (d,  $J$ =6.0 Hz, 1H), 7.18 (m, 1H);  $^{13}\text{C}$  NMR (75 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  35.0 ( $\text{CH}_3$ ), 102.7 (CH), 104.6 (CH), 128.2 (CH), 144.0 (C), 145.5 (C), 155.1 (C), 158.4 (C); MS  $m/z$  (rel intensity) 165 ( $\text{M}^+$ , 100), 122 (28); HRMS  $[\text{M}]^+$  for  $\text{C}_8\text{H}_8\text{N}_3$ : 165.0702, found 165.0708.

**4.2.1.3. 5-Fluoro-1-methyl-1*H*-indazol-3-amine (**2c**).** As described in the general procedure, reaction of 2,5-difluorobenzonitrile

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